

***FLOATING CATALYST CVD SYNTHESIS OF SWNTs –  
LESSONS EARNED AND FUTURE DIRECTIONS***

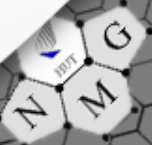
**Prof. Dr. Esko I. Kauppinen**

**Department of Applied Physics, Aalto University School of Science**

**[esko.kauppinen@aalto.fi](mailto:esko.kauppinen@aalto.fi)**

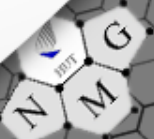
***Tokyo University, Tokyo, JAPAN***

***March 7, 2014***



# Content

- Status SWNT thin film transparent conductors (TCFs) and Thin Film Field Effect Transistors (TFT-FETs) via Direct Dry Deposition from FC-CVD reactor
- TEM of active catalyst particles – correlation of particle and tube diameters ?
- Comparison of (n,m) Distributions of SWNTs from Arc Discharge, Laser Vaporization and both Supported and Floating catalyst CVD & steps forward



# Aalto University Novel dry, direct CNT film deposition method: DPP – Direct Dry Printing

## Industrial manufacturing – Canatu Oy

Synthesis

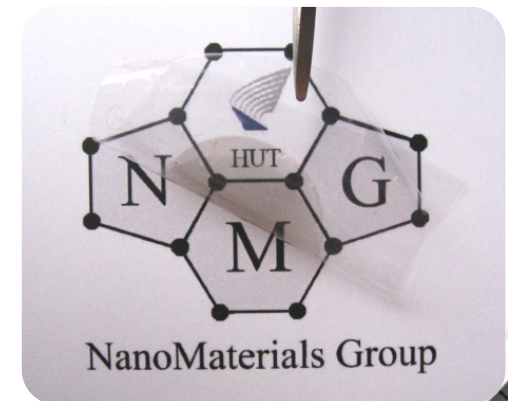
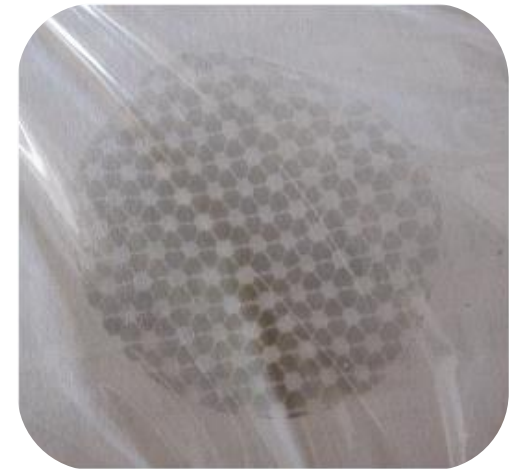
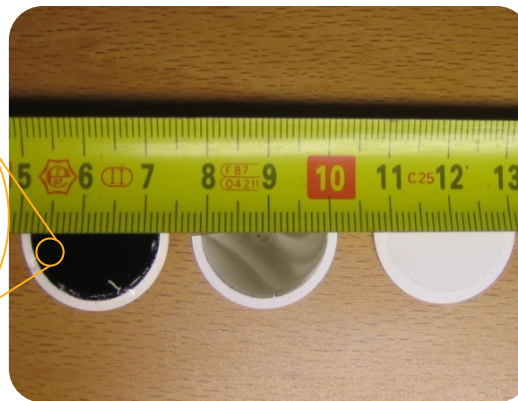
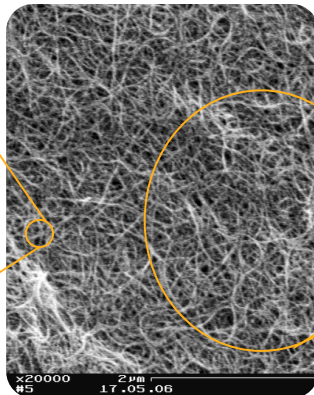
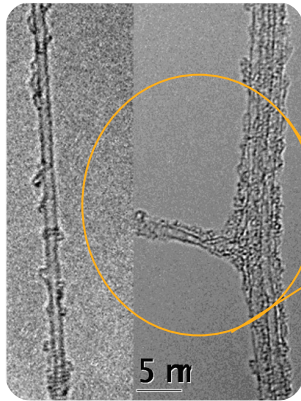
Deposition

Thin Films

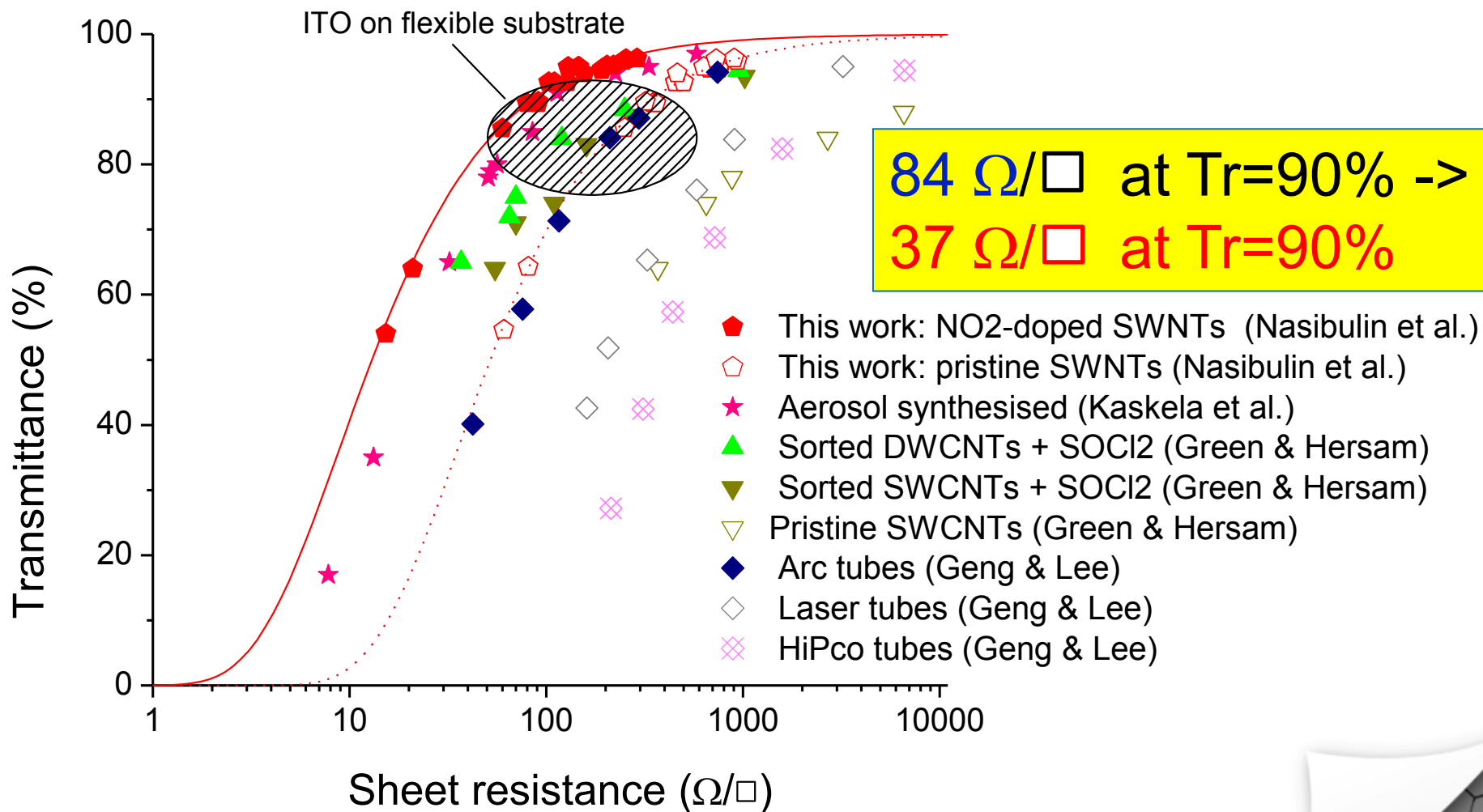
SWCNTs in the reactor gas

Control of SWCNT properties

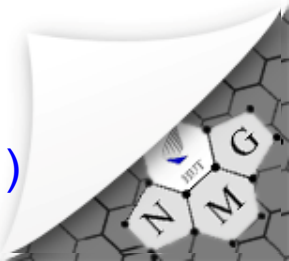
Patterned/non-patterned

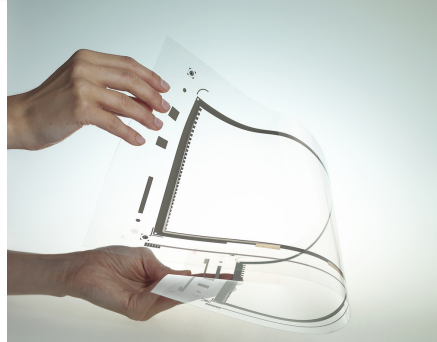


# Record performance level of SWCNT-based transparent electrodes (1.7 nm tubes)



Nasibulin, Kaskela, Mustonen, Anisimov, Kauppinen, *et al.* (2011)  
*ACS Nano*, 5(4), p.3214





## Transparent conductive films for flat, flexible or formable touch devices

CNB™ sheet resistivity	CNB™ transmission	CNB™ reflectivity	CNB™ haze
100 Ω/□	94%	0,20%	0,15%
150 Ω/□	96%	0,20%	0,15%
300 Ω/□	98%	0,20%	0,15%

Canatu is a leading manufacturer of transparent conductive films for an entirely new class of touch applications. Canatu's CNB™ films are based on a new type of carbon nanomaterial (**Carbon NanoBud®**), and a new Roll-to-Roll manufacturing process (**Direct Dry Printing®**), which combines the synthesis of the Carbon NanoBuds directly from carbon gases and direct deposition to the substrate in one single process. Canatu offers its customers in consumer electronics and automotive industry next level of design freedom with its innovative technologies.



Canatu to talk at High Value Manufacturing Graphene Conference 2013

Canatu releases new transparent conductive films with ultra-low reflections and record flexibility

Canatu has been invited to talk about flexible touch sensors in the upcoming global trade events

Canatu to showcase and launch Generation 5 Carbon NanoBud® films in FPD International 2013

Canatu will give a talk on advanced printed touch sensors at the ICFPE conference in Korea

**Carbon Nanobuds®**  
increase display contrast



Meet Canatu at FPD International 2013



Canatu's CNB™ transparent conductive films are designed for demanding touch sensors, paving the way for high-contrast touch displays also in flexible, foldable and 3D format.

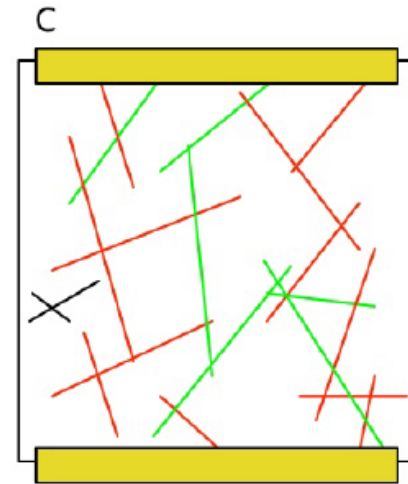
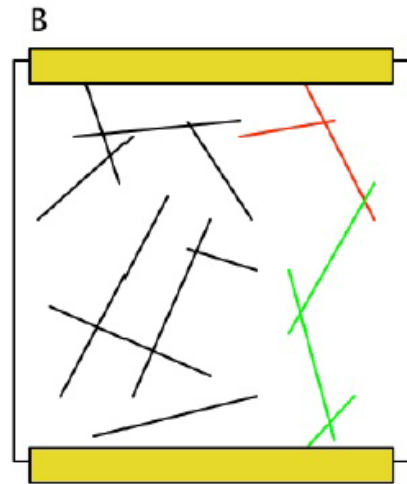
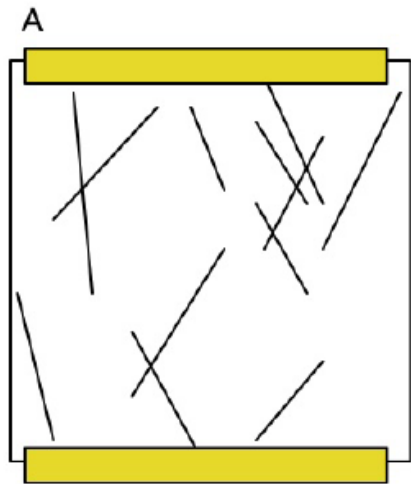
Canatu's high-quality CNB™ films serve touch module and display companies in the rapidly-expanding markets of mobile phones, tablets, cameras, wearable consumer devices, white goods, home appliances and automotive. Canatu also supplies CNB™ touch sensors for touch module prototyping and small to medium volume production.

Canatu can help electronics hardware companies and brand names who are seeking ways to differentiate products, reduce production costs and increase product performance. Existing materials such as silicon, metals and metal oxides are expensive, not flexible, stretchable or transparent, and require complex and bulky support structures. Moreover, these traditional technologies are reaching their performance limits. With Canatu's CNB™ films it is possible to replace Indium Tin Oxide (ITO) in capacitive touch sensors using existing driving electronics.

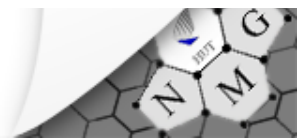
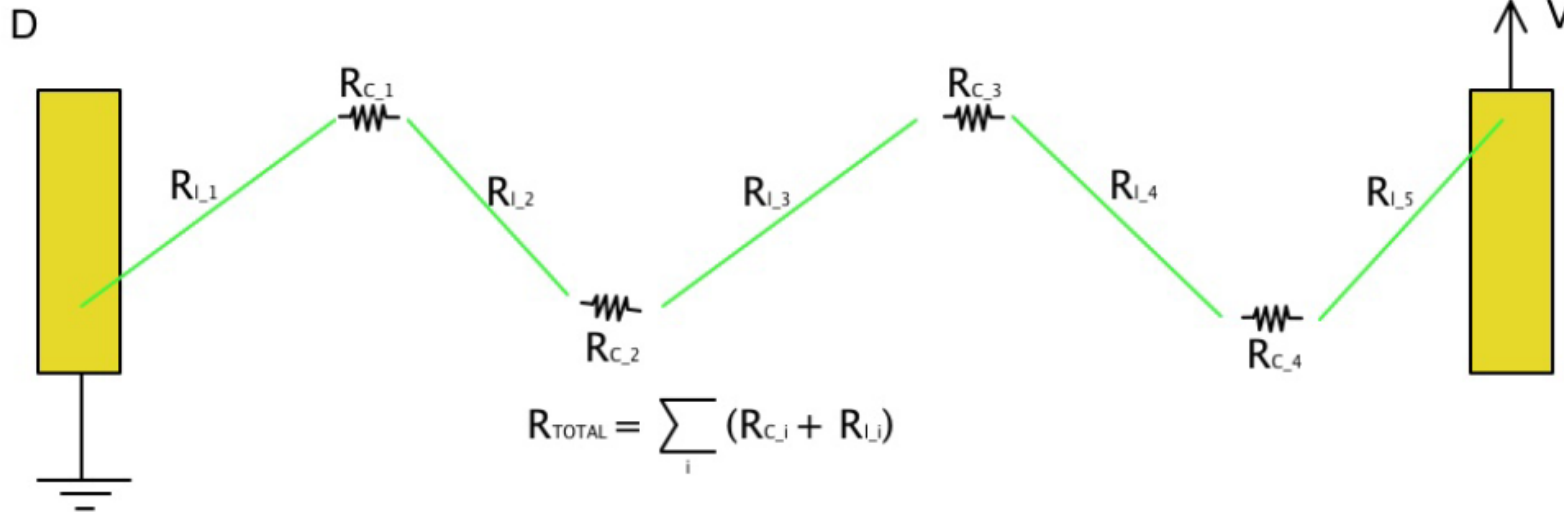


Canatu Ltd: 96% T @ 150 Ohm/□ for PET+CNT film => 97% T @ 150 Ohm/□ for SWNT film only. Stability: T240 h @ 60°C/ 90%RH with < 20% variation of sheet resistance

# Simplified resistance model for SWNT network

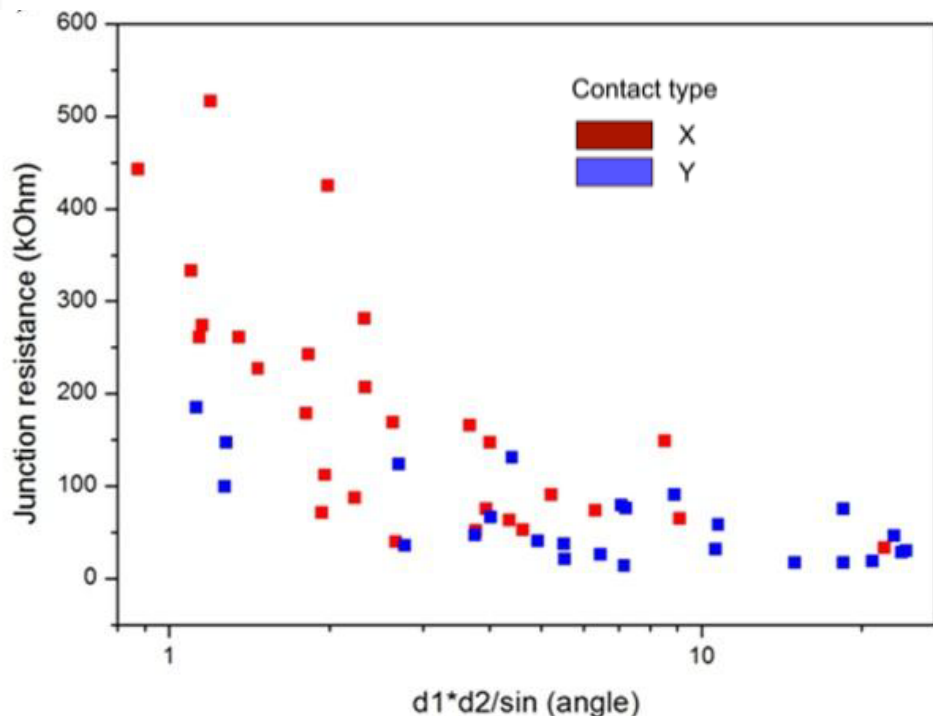


- Not Connected
- Semiconductive
- Metallic



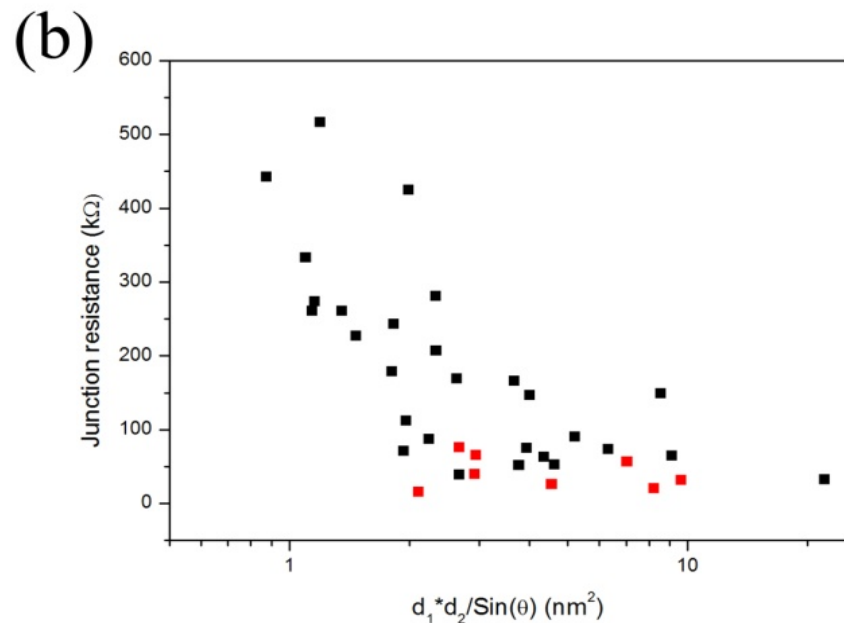
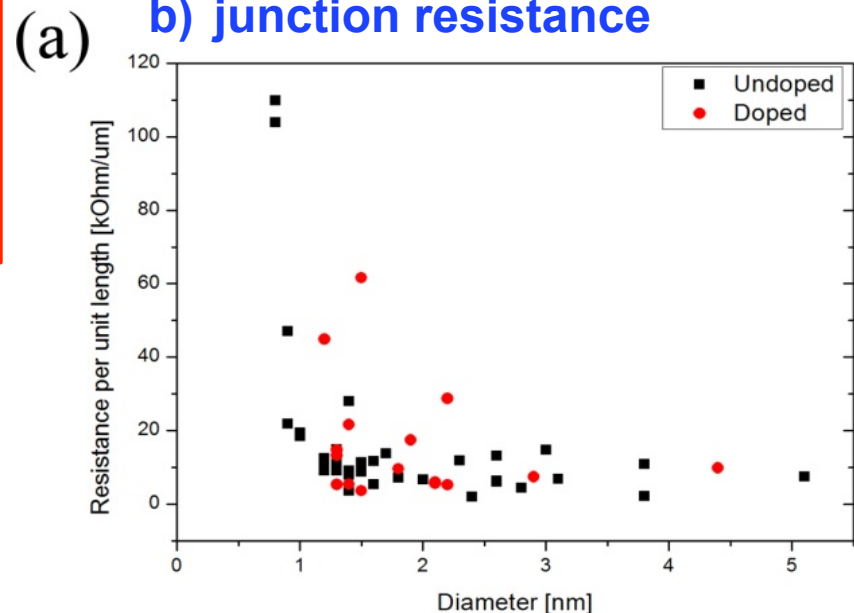
# Tube-to-tube contact resistance as well along tube resistance measurement of pristine SWNTs with conductive atomic force microscope (C-AFM)

The junction resistance versus interfacial area.  $d_1$  = tube 1 diameter  
 $D_2$  = tube 2 diameter.

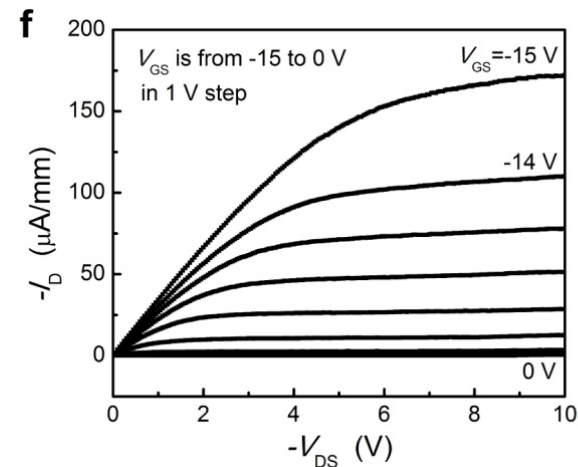
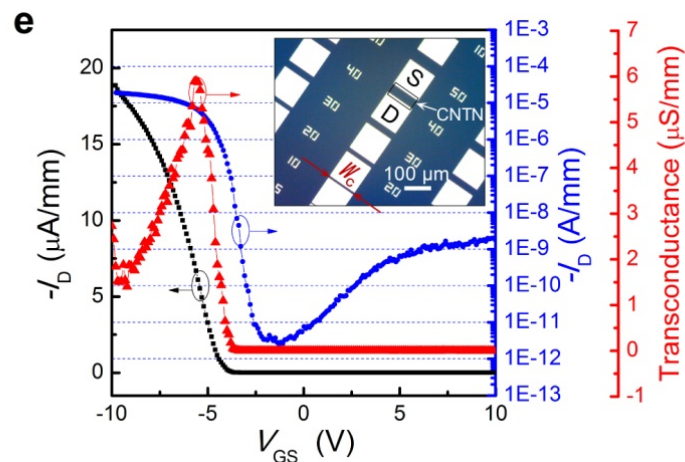
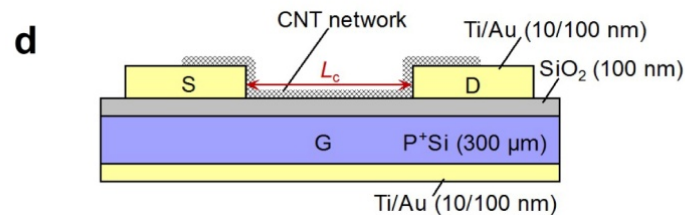
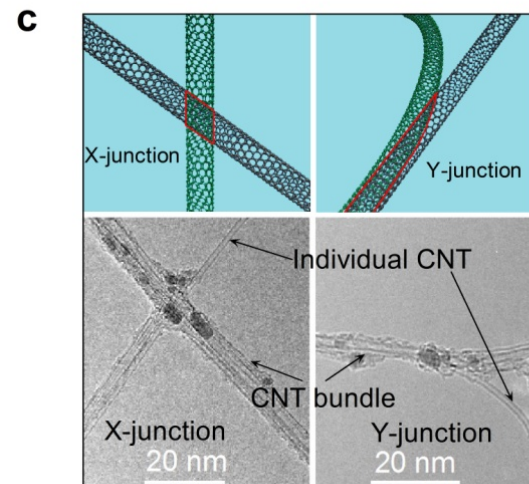
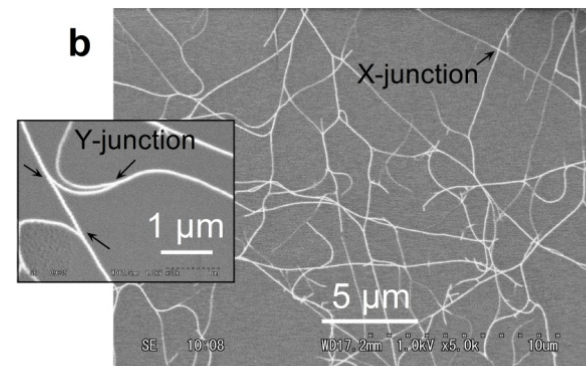
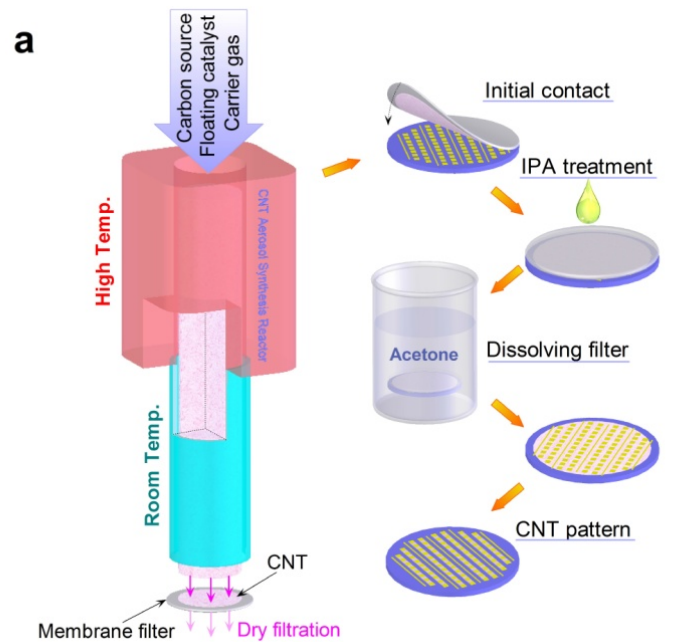


Znidarsic, Kaskela et al. (2013) *J. Phys. Chem. C* **117**, 13324-13330.

## The effect of acid doping on a) resistance per unit length and b) junction resistance



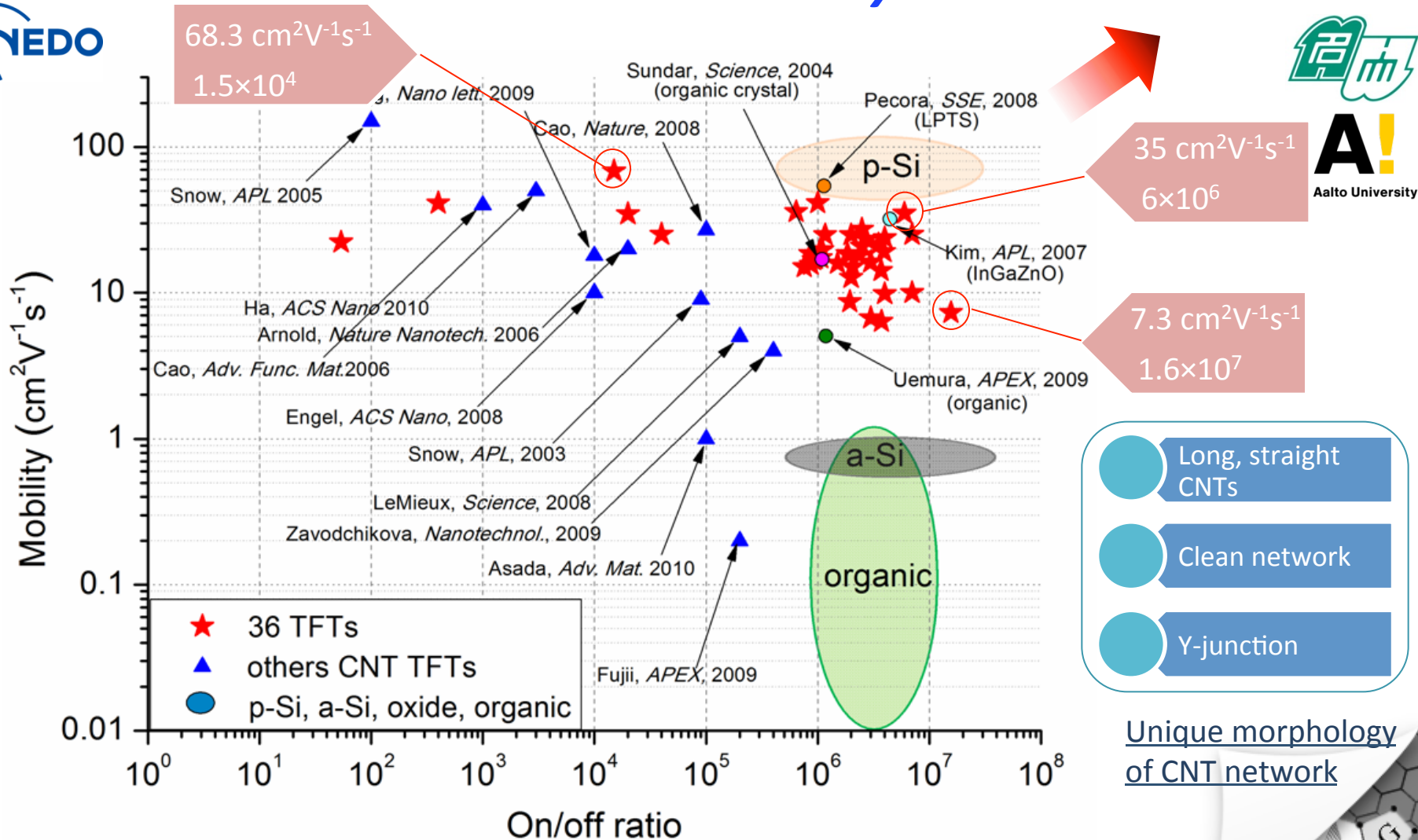
**Dissolve Filter (DF):**  
**High mobility & large on/off ratio CNTN-FETs**  
 by better controlling SWCNT bundle length, orientation and SWCNT chiral distribution  
 (Sun, Zavodchikova, Tian, Nasibulin, Kauppinen, Kishimoto, Mizutani & Ohno (2011), Nature Nanotechnology)



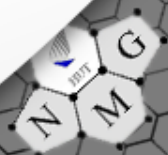


# High performance carbon nanotube TFTs

- High mobility (thin film capacitance model) and high on/off achieved concurrently ←

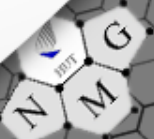


Sun, Timmermans, Tian, Nasibulin, Kauppinen, Kishimoto, Mizutani and Ohno, *Nature Nanotechnology* (2011) **6**, 156–161.



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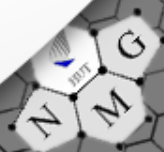
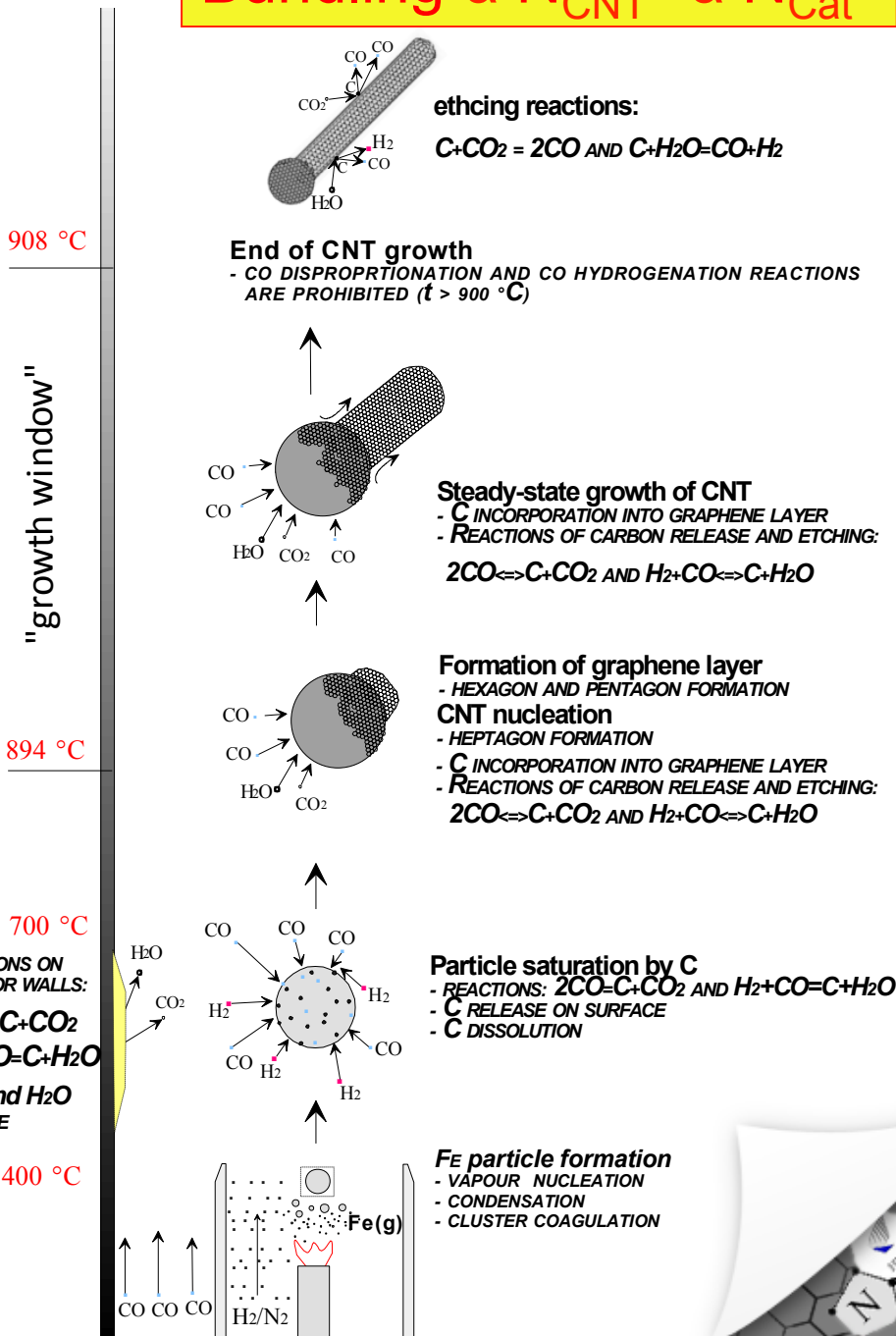


Schematics of FC-CVD reactor with pre-made Fe catalyst clusters formed inside the SWNT reactor via PVD – catalyst growth and SWNT nucleation coupled (Nasibulin et al., 2005)

**Control :**

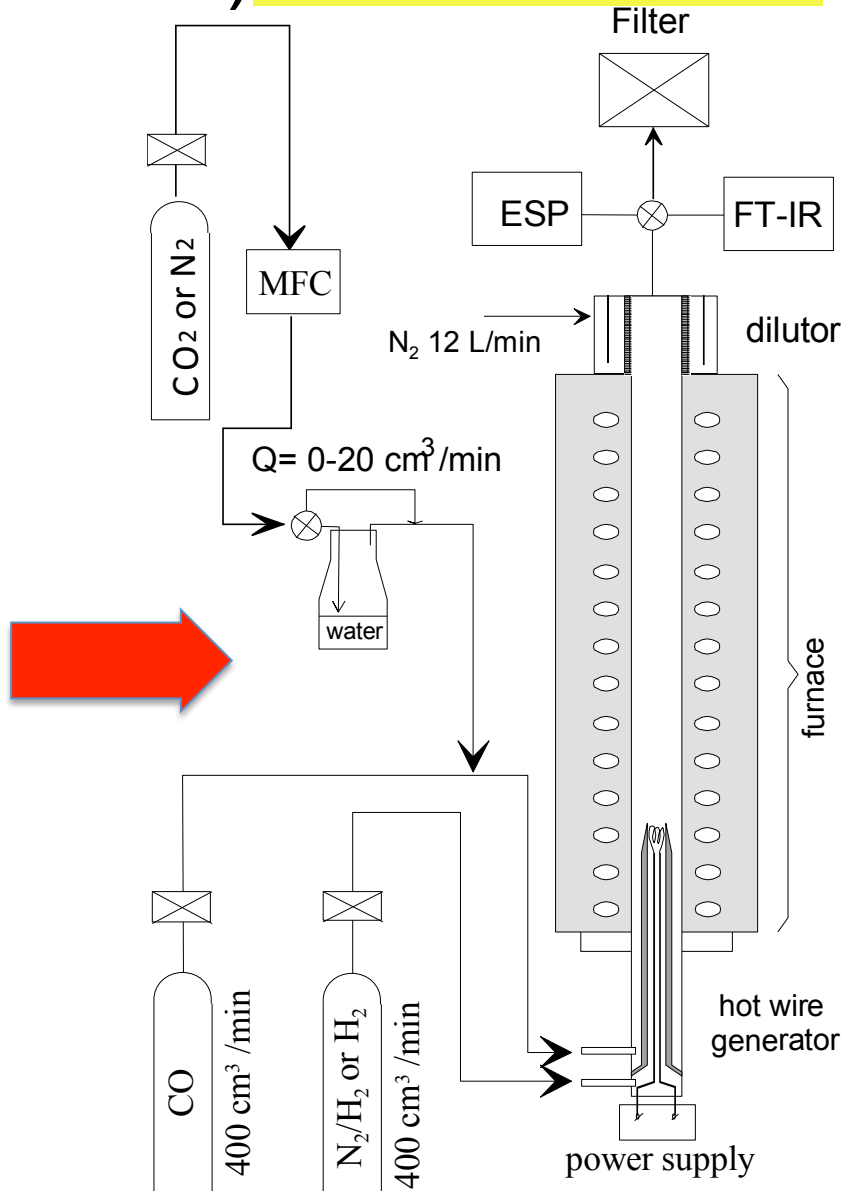
- Diameter and length via temperature and oxidant (e.g. CO<sub>2</sub>, NH<sub>3</sub>, H<sub>2</sub>O) concentration
- Bundling via catalyst concentration

Bundling  $\propto N_{CNT}^2 \propto N_{Cat}^2$

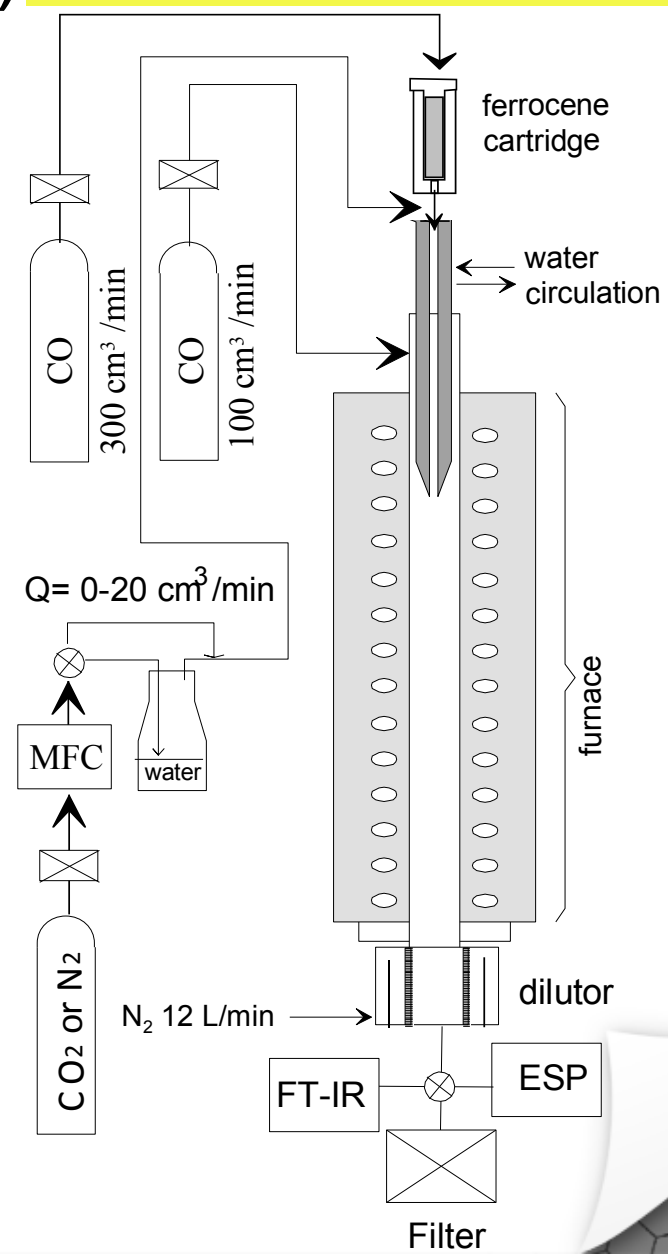


# Floating Catalyst Methods for CNT Synthesis

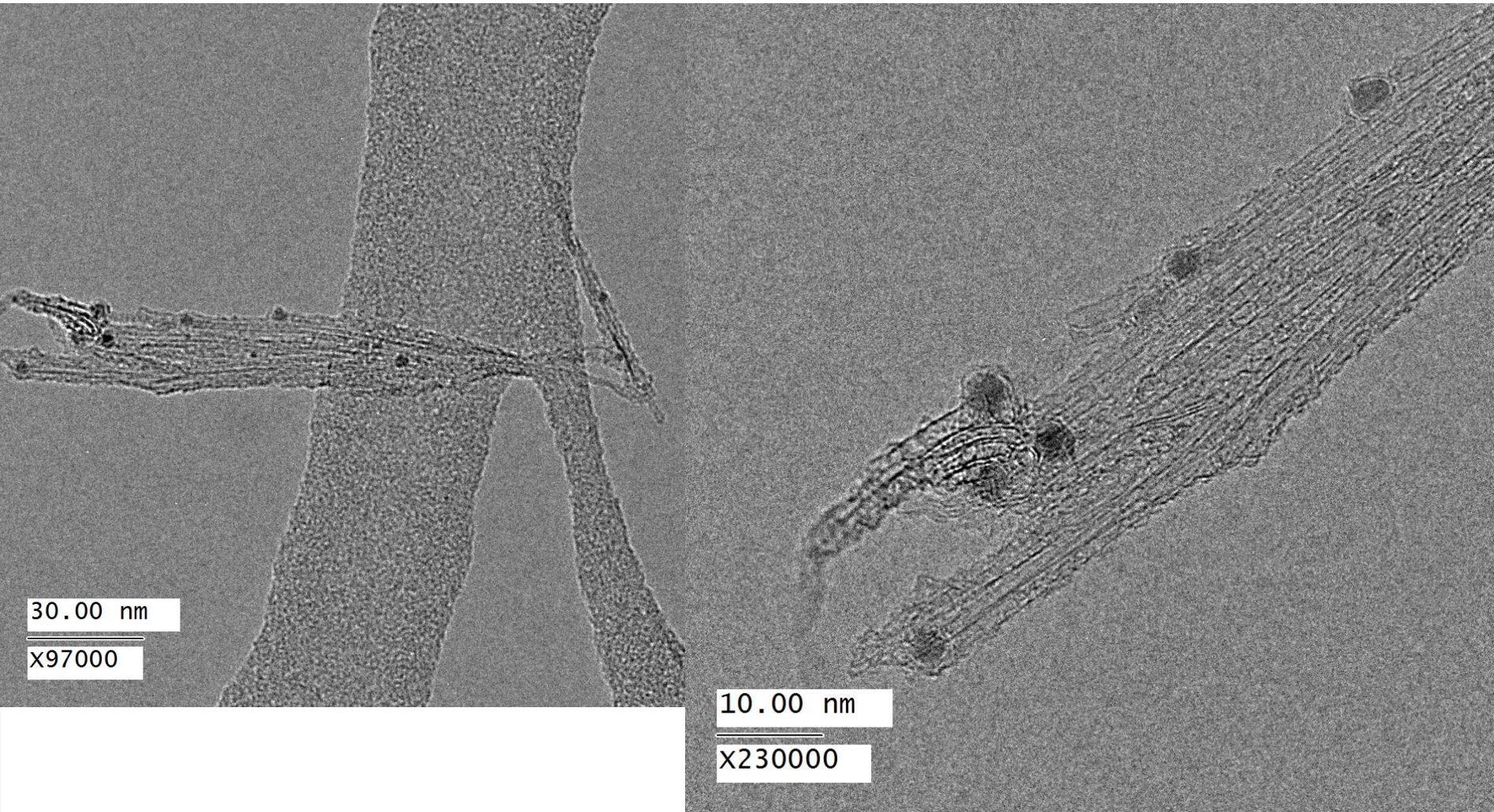
a) **HWG-based CVD of CO**



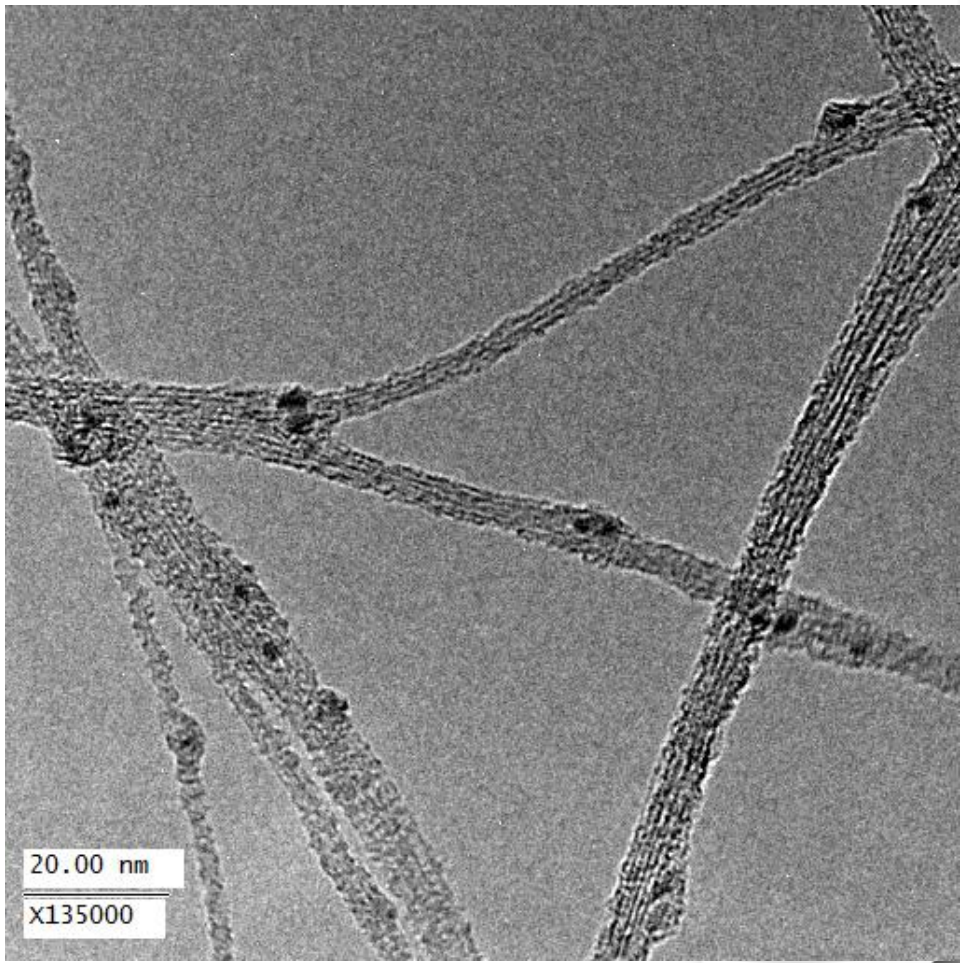
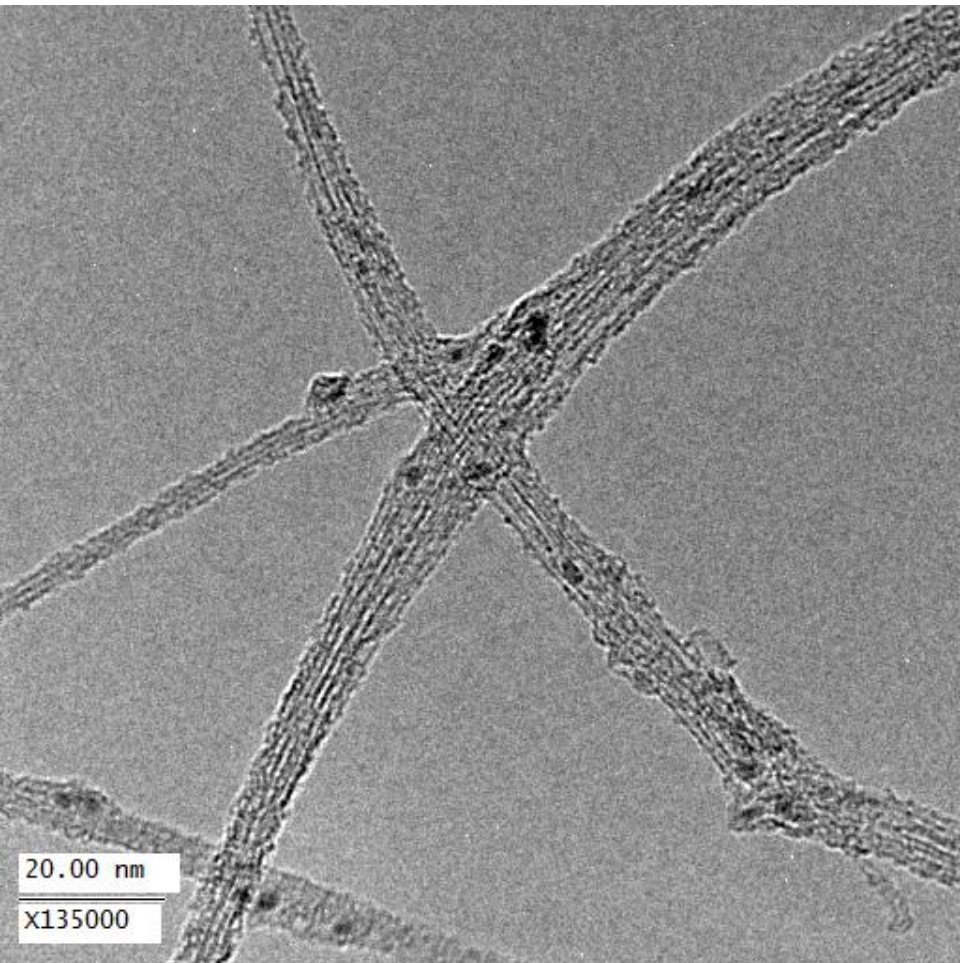
b) **Ferrocene-based CVD of CO**



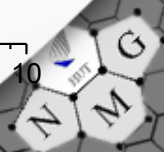
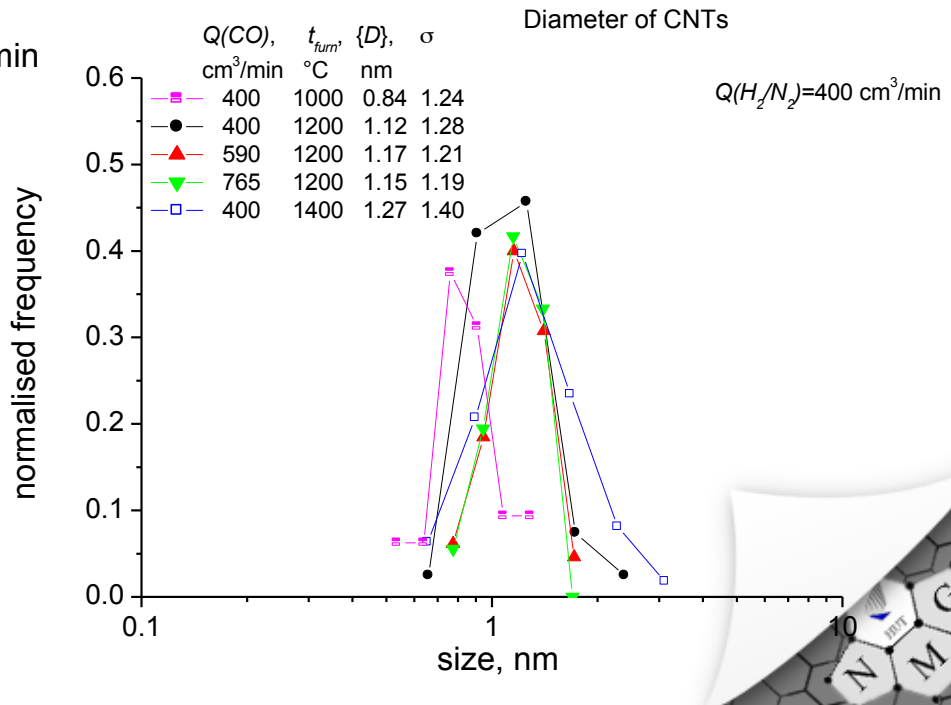
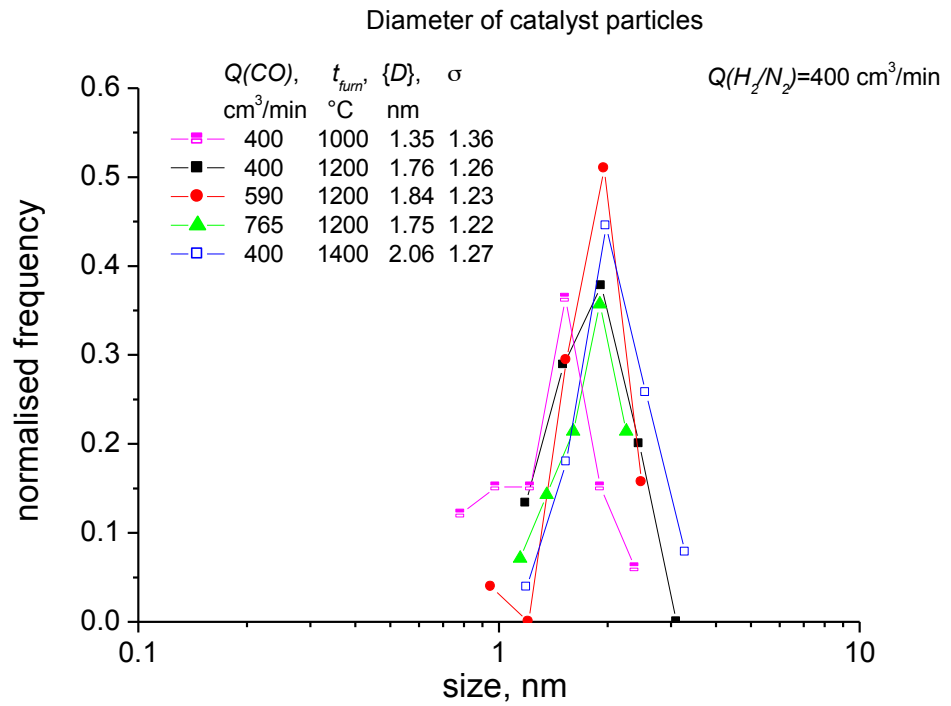
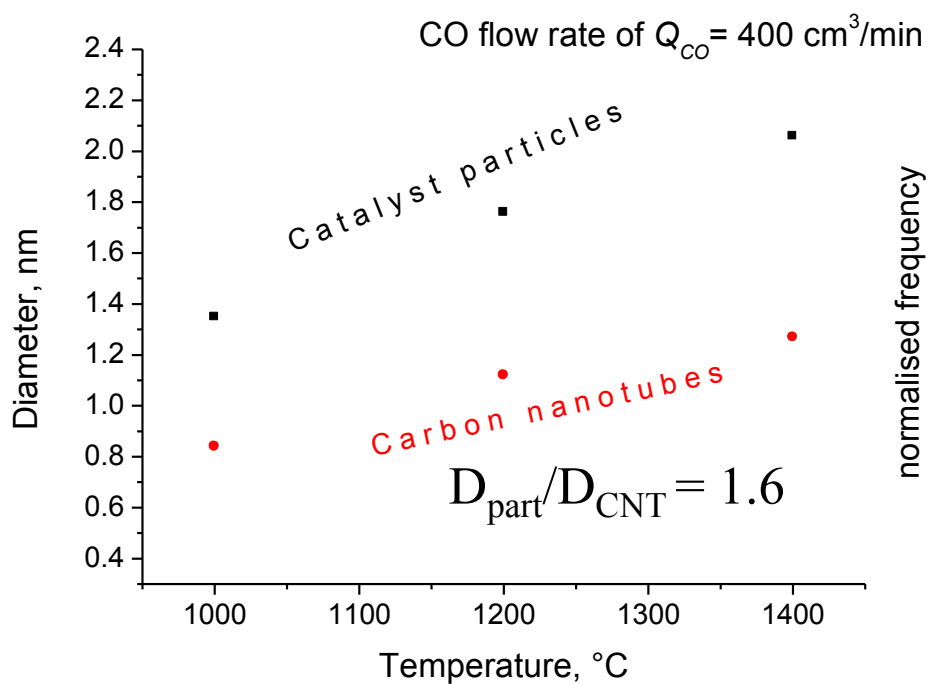
TEM of short SWNTs with Fe catalyst particles  
made from CO with HWG FC-CVD reactor 2002  
200 kV, no heating prior imaging



TEM of SWNTs with Fe catalyst particles  
made from CO with HWG FC-CVD reactor 2003  
200 kV, no heating prior imaging



# Correlation between diameter of Fe particles and SWCNTs – control of catalyst size



# Ratio between diameters of catalyst particles and CNTs

## Pre-made Fe particles introduced in conditions of CNT formation

Experimental conditions		Ratio
$t_{furn}$ , °C	$Q_{CO}$ , cm <sup>3</sup> /min	$D_{part}/D_{CNT}$
1000	400	<b>1.60</b>
1200	400	<b>1.57</b>
1200	590	<b>1.57</b>
1200	765	<b>1.52</b>
1400	400	<b>1.62</b>

$H_2$ concentration	$D_{CNT}$ , nm	$D_{Fe}$ , nm	$D_{part}/D_{CNT}$
0.07	1.4	2.1	<b>1.50</b>
7	1.3	2.1	<b>1.62</b>
100	1.3	2.0	<b>1.54</b>

Ethanol,  $t_{furn} = 1200$  °C:

$$D_{CNT} = 1.7 \text{ nm}, D_{Fe} = 2.4.$$

$$D_{part}/D_{CNT} = \mathbf{1.41}$$

## In-situ CVD Synthesis of Particles :

Ferrocene vapor decomposition  
in CO at 1150 °C

$$D_{CNT} = 1.3 \text{ nm}, D_{part} = 3.1 \text{ nm}$$

$$D_{part}/D_{CNT} = \mathbf{2.4}$$

**MORE DETAILS LATER**

HiPco CNTs

(Nikolaev *et al.* ChemPhysLett 1999)

$$D_{CNT} = 0.7\text{-}1.4 \text{ nm}, D_{part} = 5\text{-}10 \text{ nm}$$

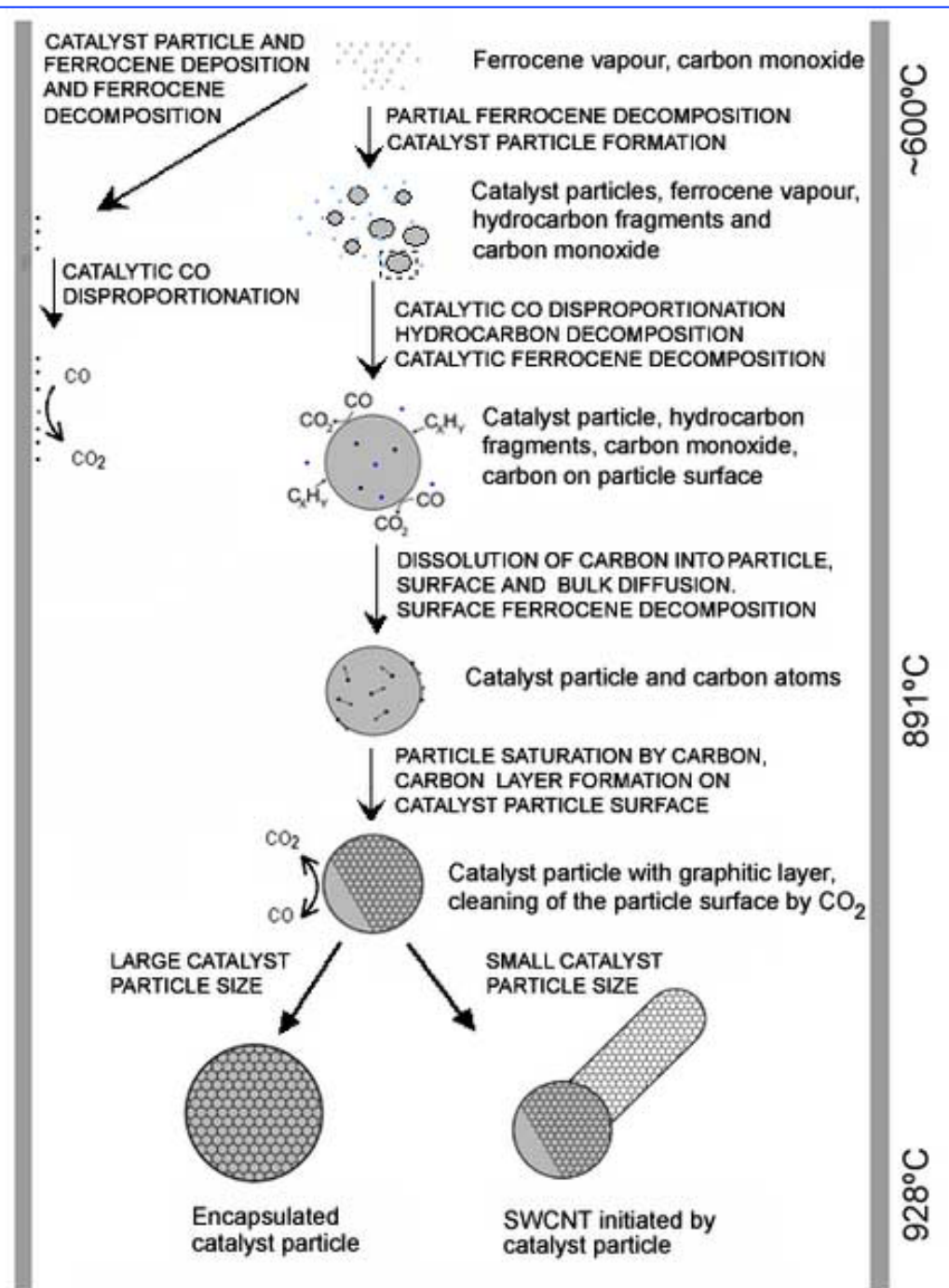
$$D_{part}/D_{CNT} > \mathbf{3}$$



Catalyst formed in-situ during *ferrocene* decomposition in  $CO/CO_2$  – catalyst nucleation & growth and SWNT nucleation coupled.

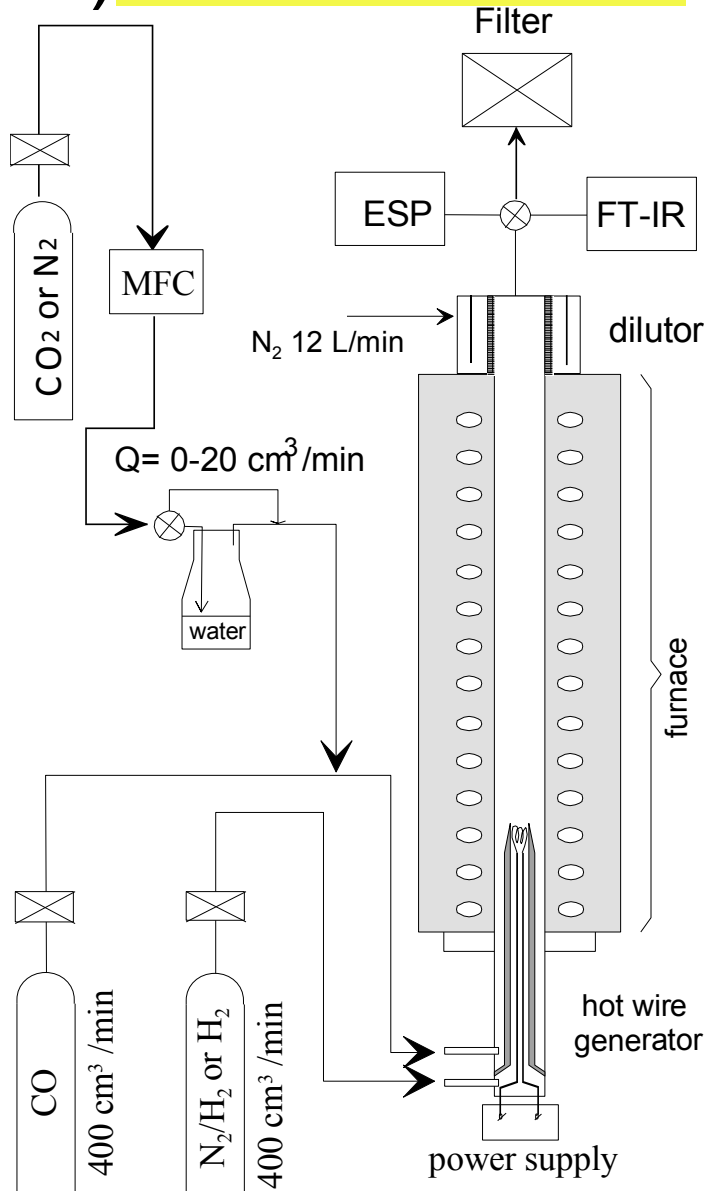
Moisala et al., CES 2006

Ferrocene decomposition slow – overlapp between CNT nucleation & growth and cerrocene decomposition – CVD growth of Fe clusters

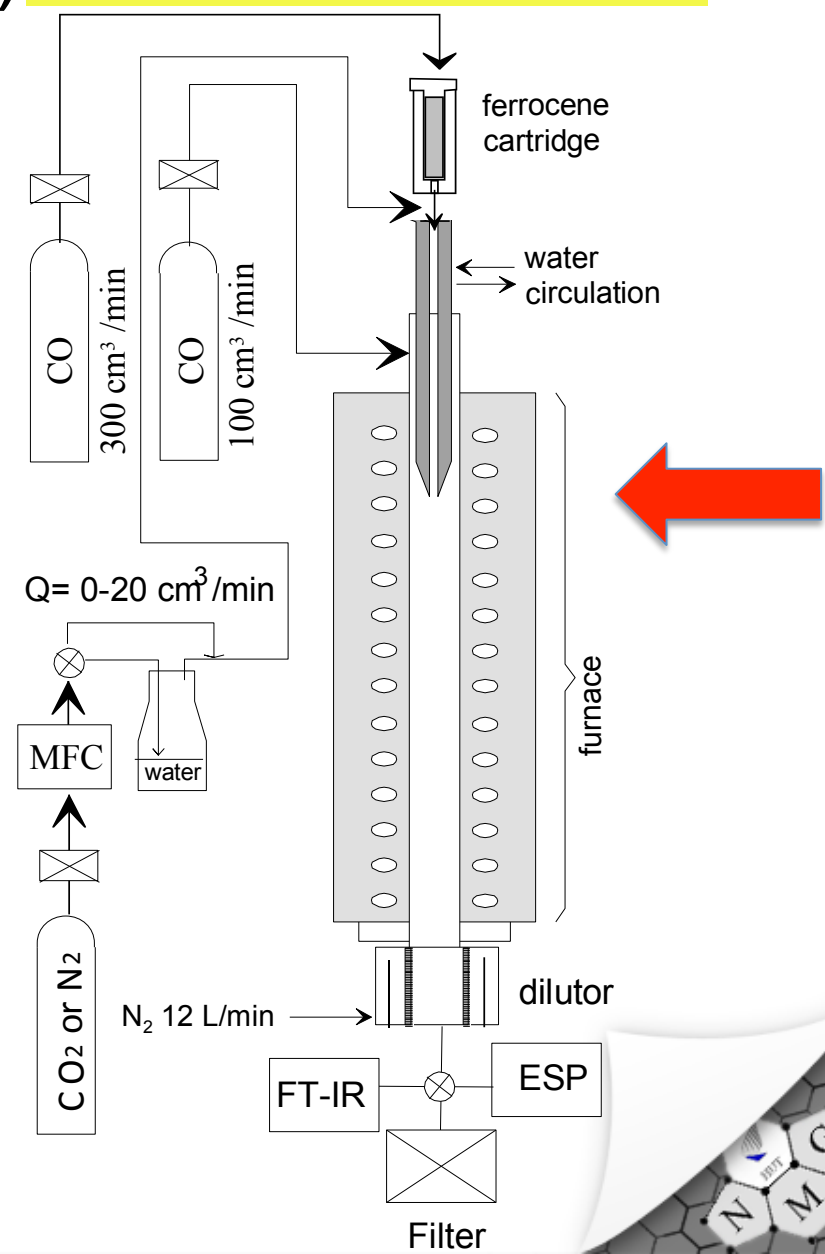


# Floating Catalyst Methods for CNT Synthesis

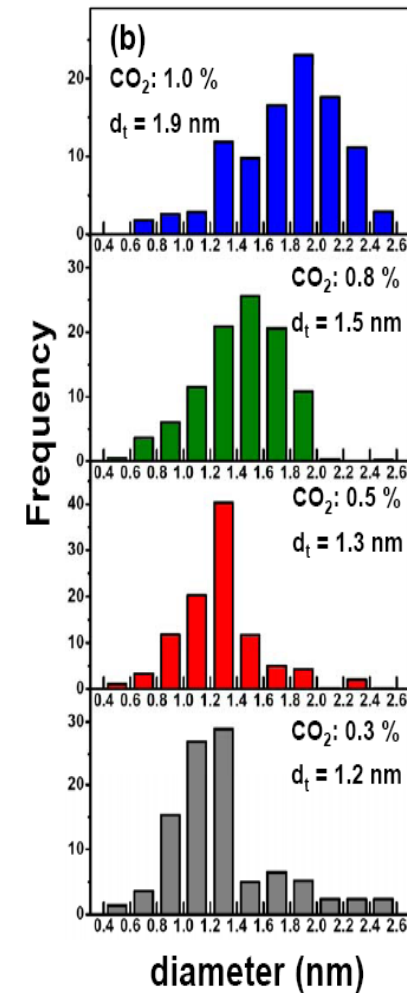
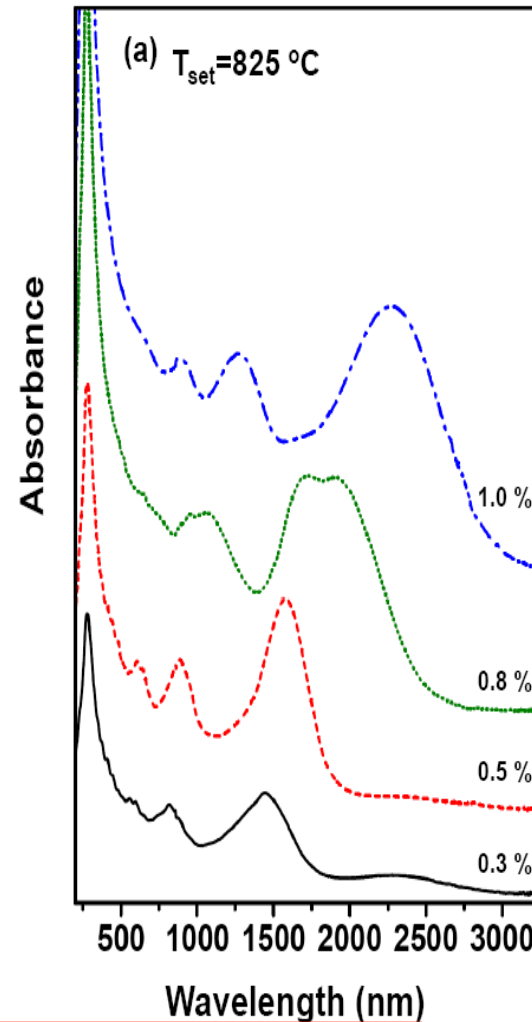
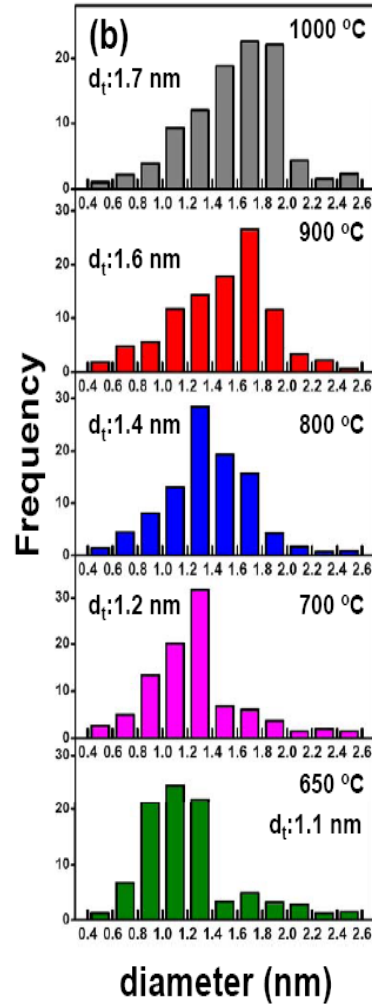
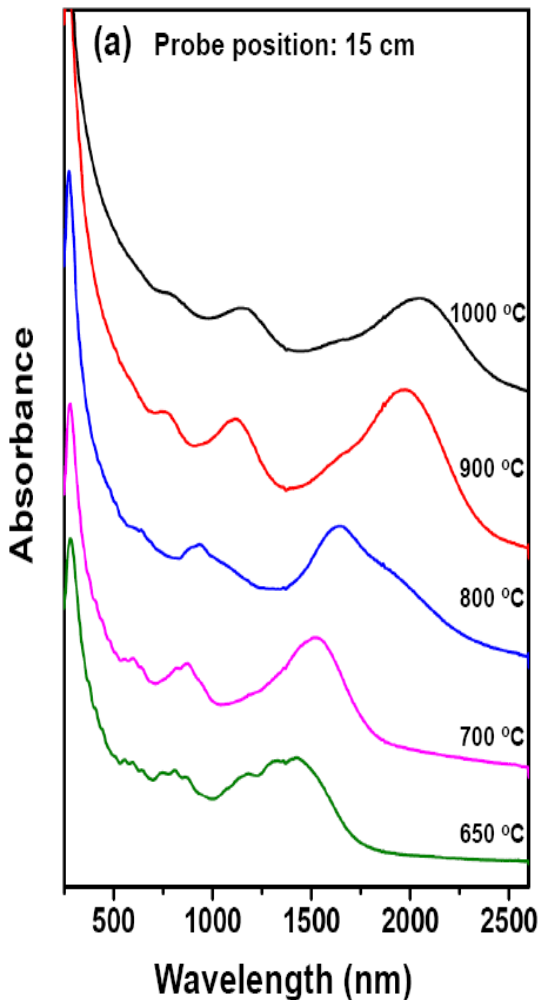
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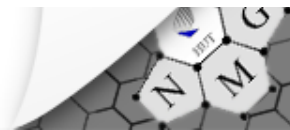
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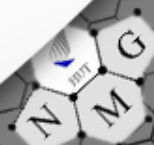
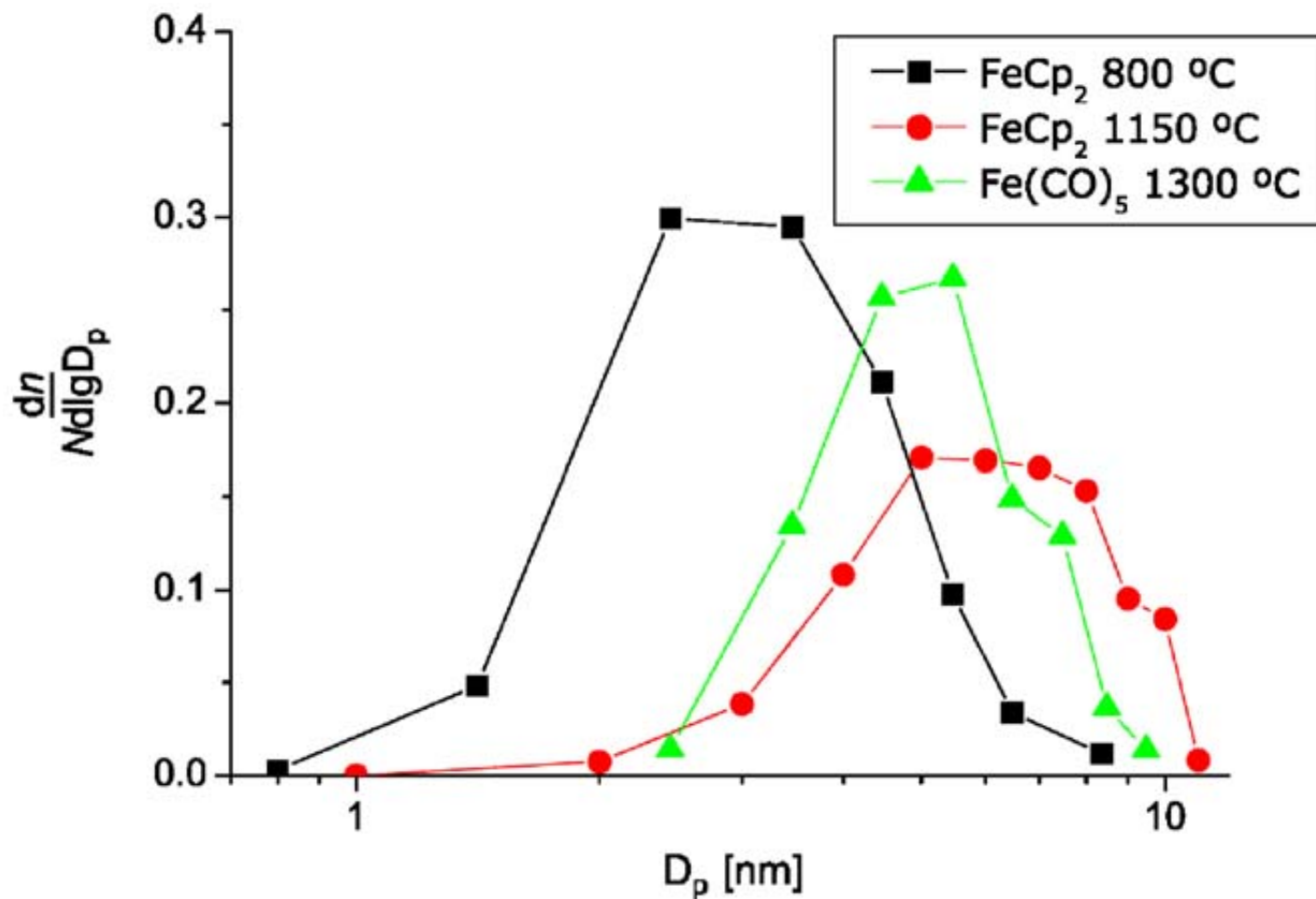
# Optical absorption derived SWCNT diameter distributions vs. wall temperature and vs. CO<sub>2</sub> concentration



Ying et al. (2011) Carbon



Catalyst particle size distribution in SWNTs produced with ferrocene reactor.  
SWNT mean diameter about 1.3 nm.

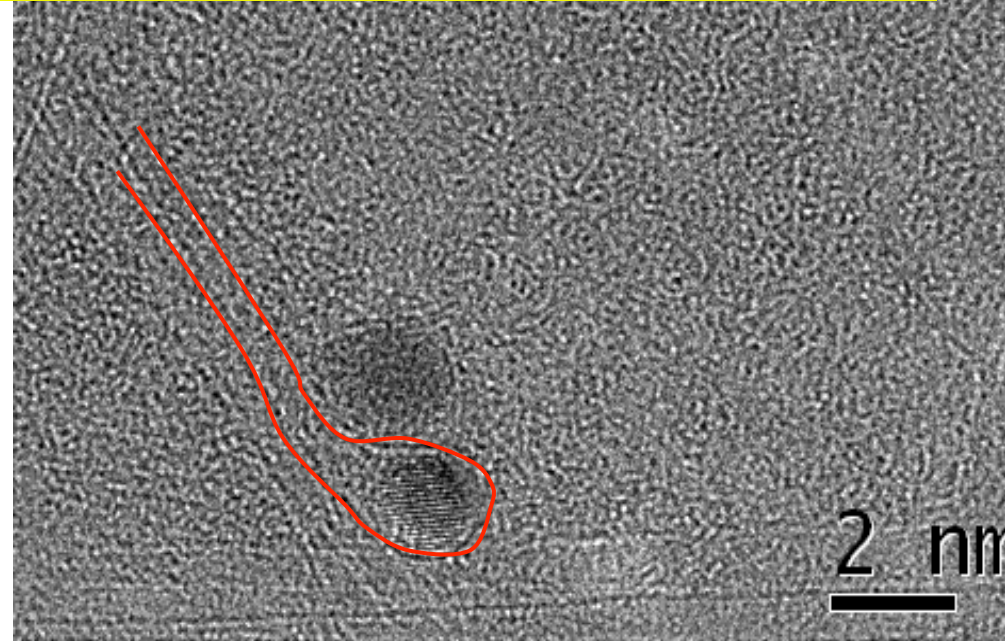
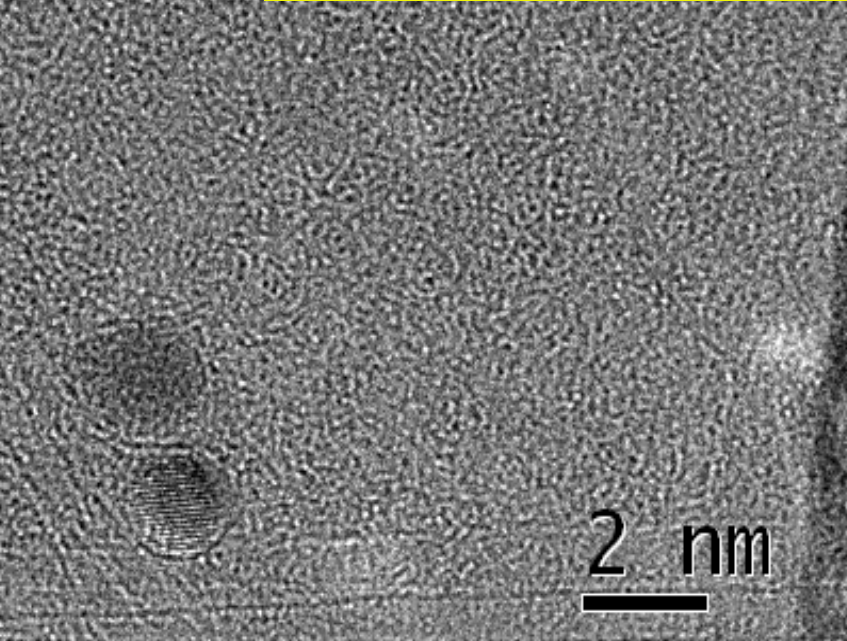


# Cs-corrected TEM images of SWCNTs

Sample: **Ferrozen**  
1000 C with CO + H<sub>2</sub>O  
Collected on TEM grid  
via ESP filter

Microscope: **FEI Titan™ 80-300**  
High tension (HV): **80 kV**

*Active catalyst is larger than the tube*



# Cs-corrected TEM images of SWCNTs

Microscope: **FEI Titan™ 80-300**

High tension (HV): **80 kV**

Carbon atoms

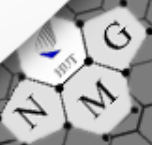
*Large  
Non-active  
Catalyst  
Particles*

Sample: **Ferrozene Reactor**

1000 C with CO + H<sub>2</sub>O

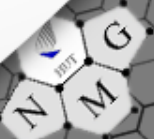
Collected to TEM grid via ESP filter

5 nm

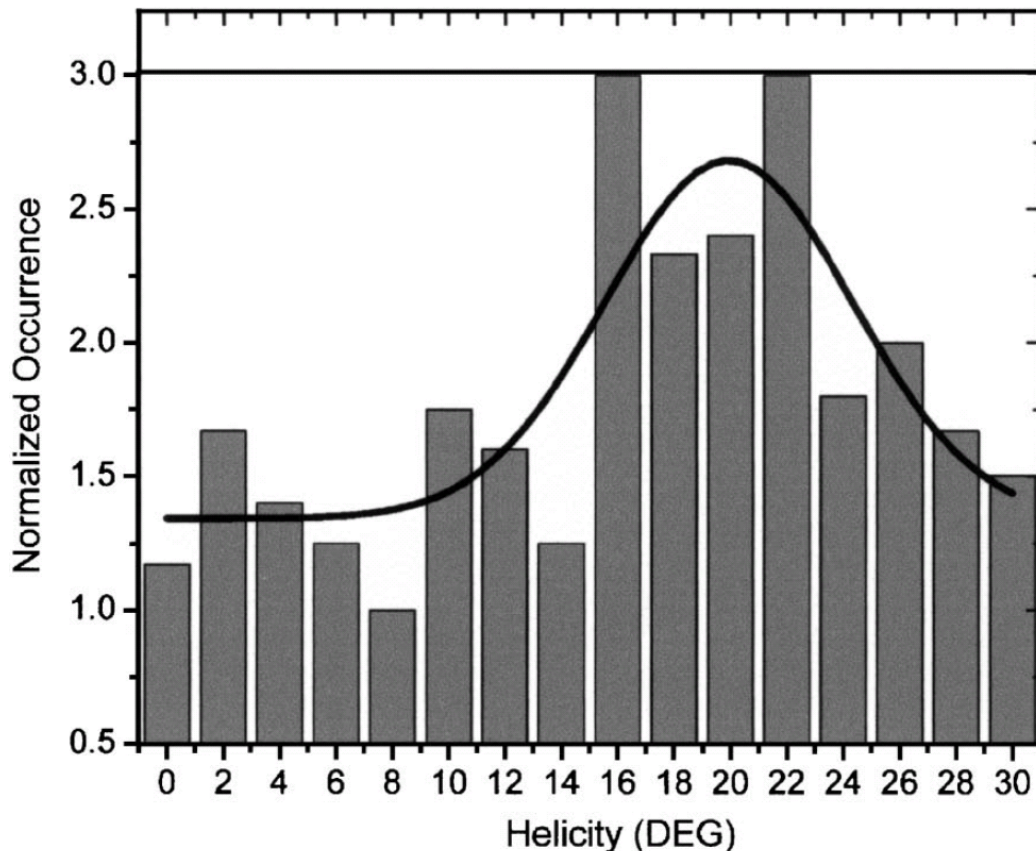


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# Chiral angle distributions for a SWCNT sample produced by arc discharge<sup>1</sup>

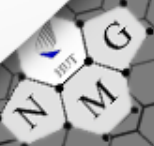


Chiral angle distribution of 124 individual SWNTs characterized experimentally by nano-beam **electron diffraction** analysis.

The histogram shows normalized number (normalized by dividing the observed number of nanotubes by the number of nanotube species at each helicity) of characterized individual nanotubes with different helicities.

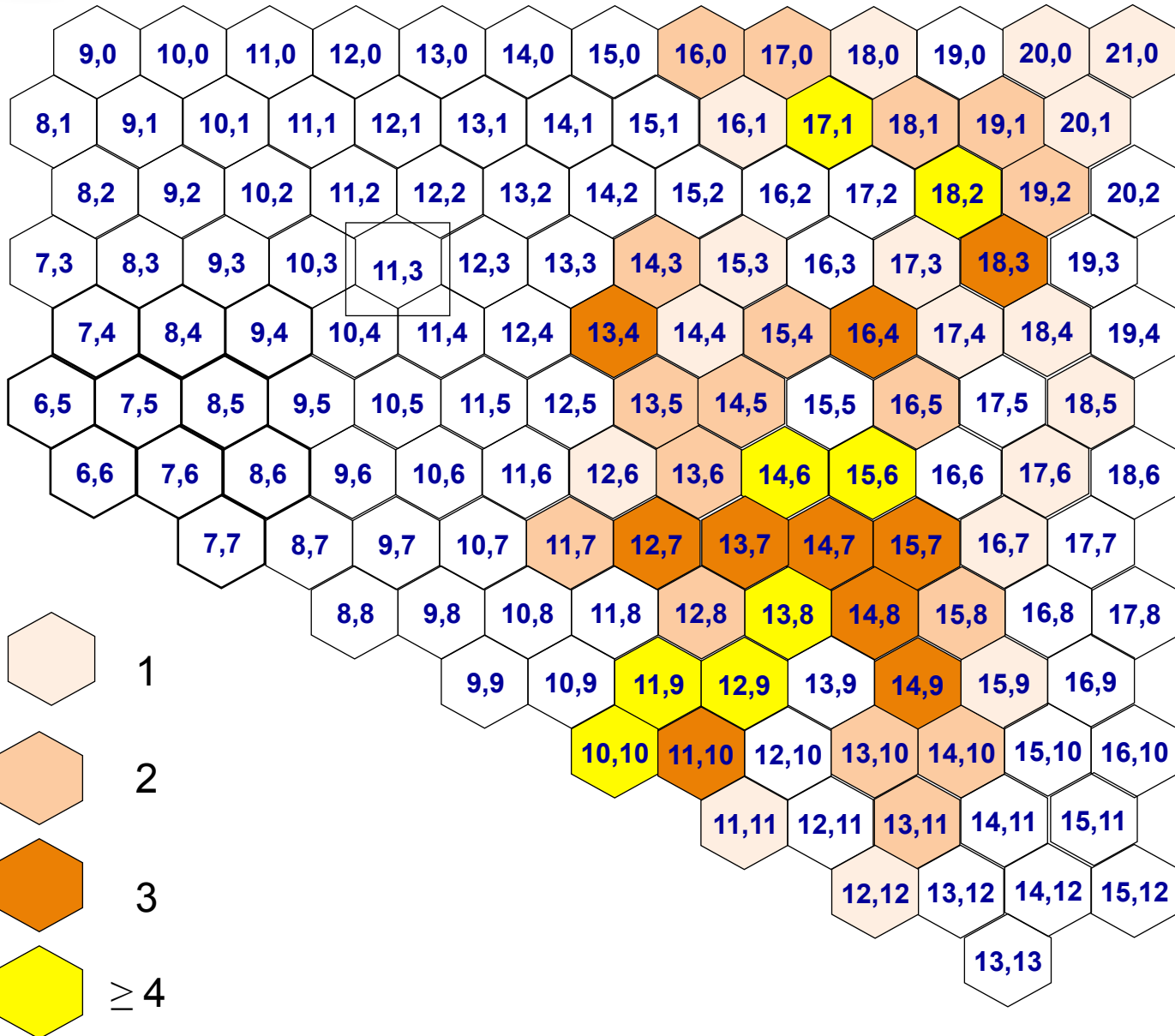
-- Qin et al., *Phys. Rev. B* **71**, 245413 (2005)

1. C. Journet, W. K. Maser, P. Bernier, A. Loiseau, M. Lamy de la Chappelle, S. Lefrant, P. Deniard, R. Lee, and J. E. Fischer, *Nature (London)* **388**, 756 (1997).

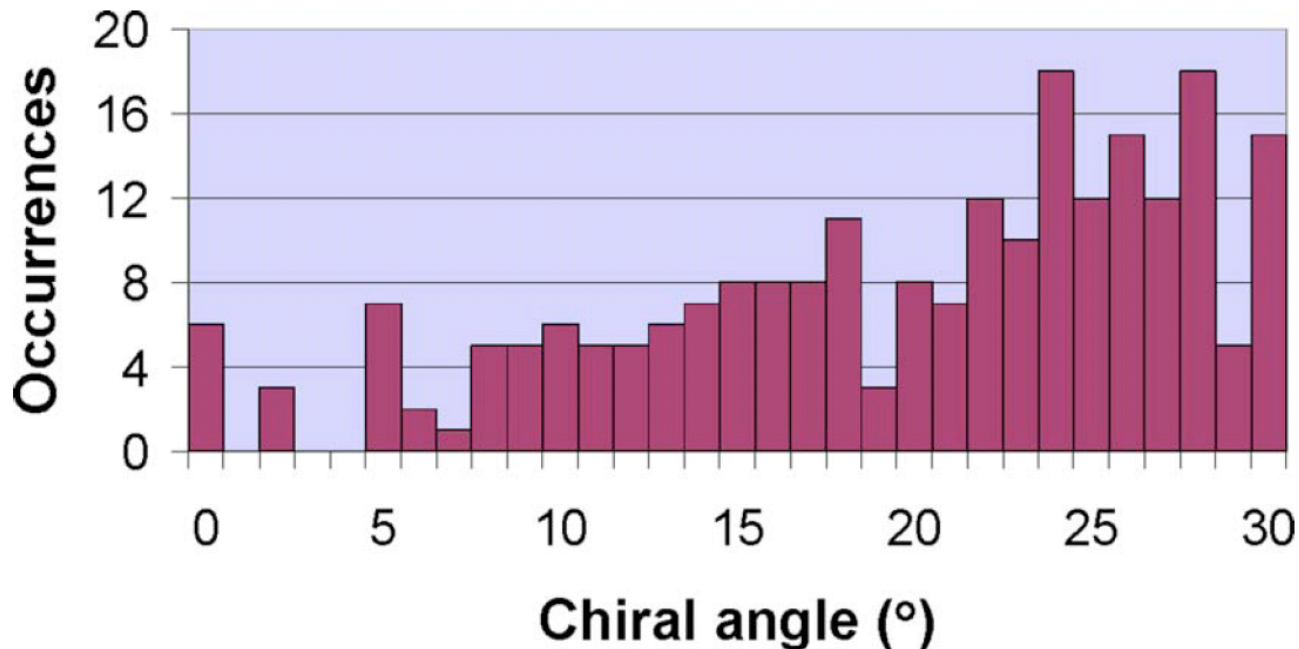




# Arc SWNTs



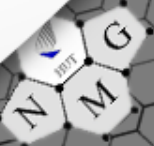
# Chiral angle distributions for a SWCNT sample produced by laser ablation<sup>2</sup>



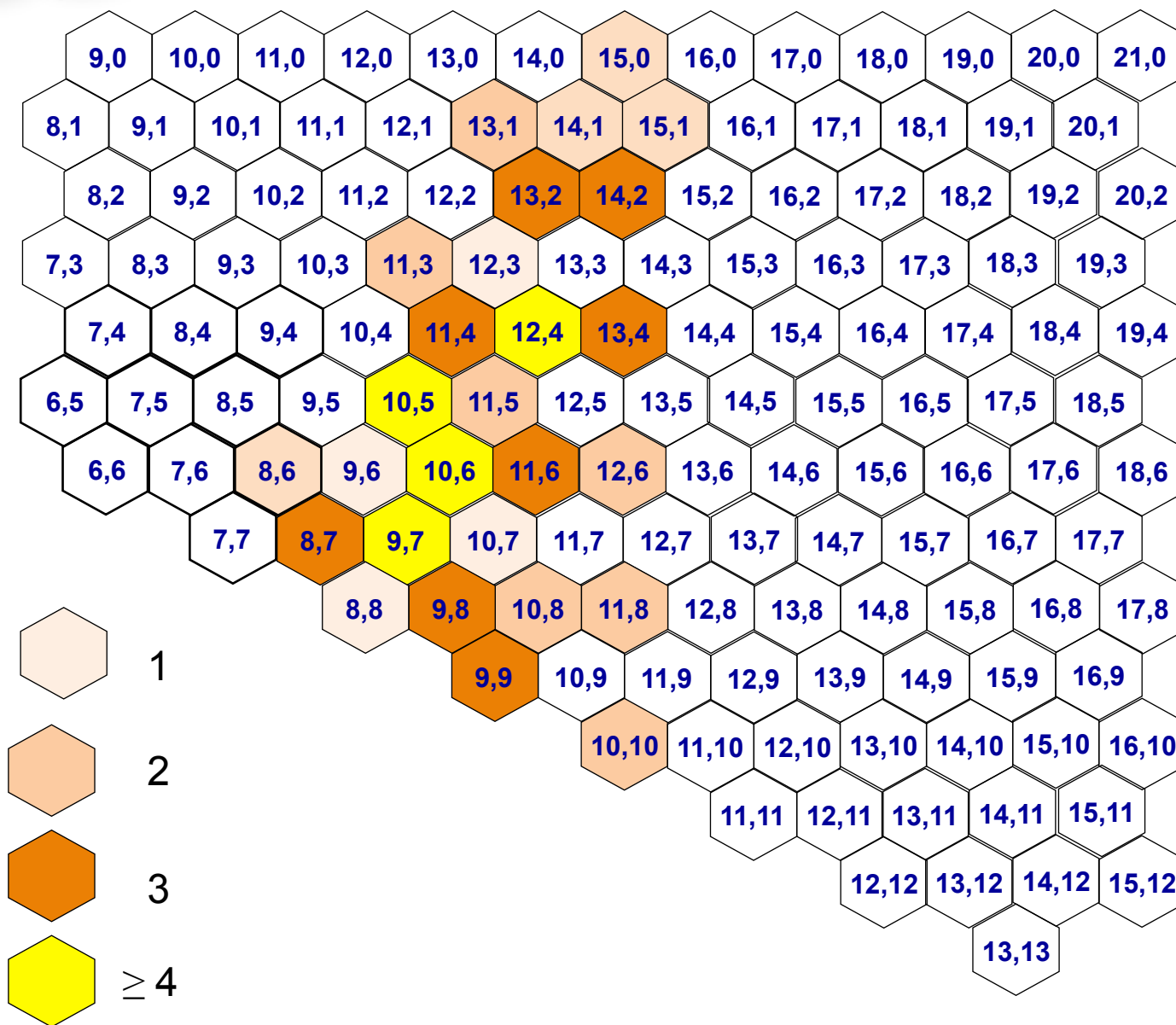
Chiral angle distribution for a laser-ablation SWCNT sample. The measurement was based on **electron diffraction analysis** of 48 SWCNT bundles and 10 individual SWCNTs which results in a total occurrence of 228 helicities for statistic.

-- Jiang et al., *Appl. Phys. Lett.* **93**, 141903 (2008)

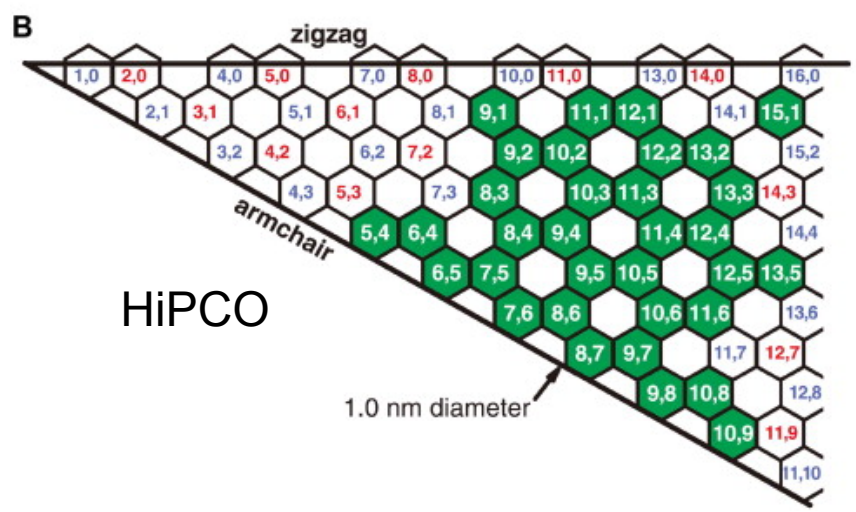
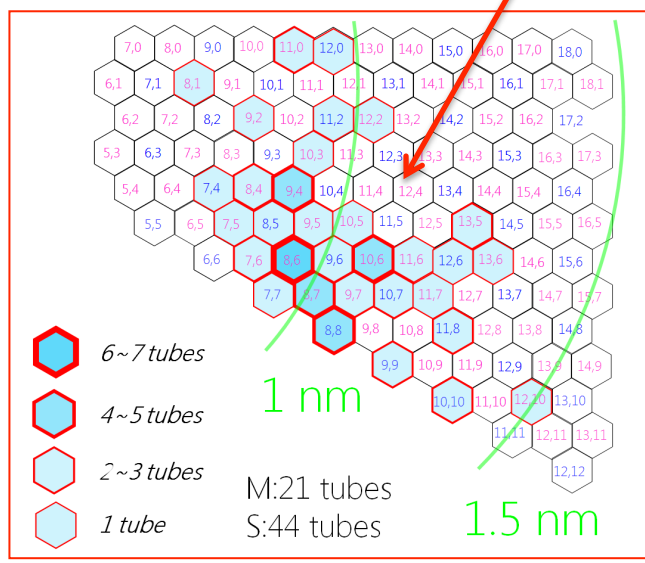
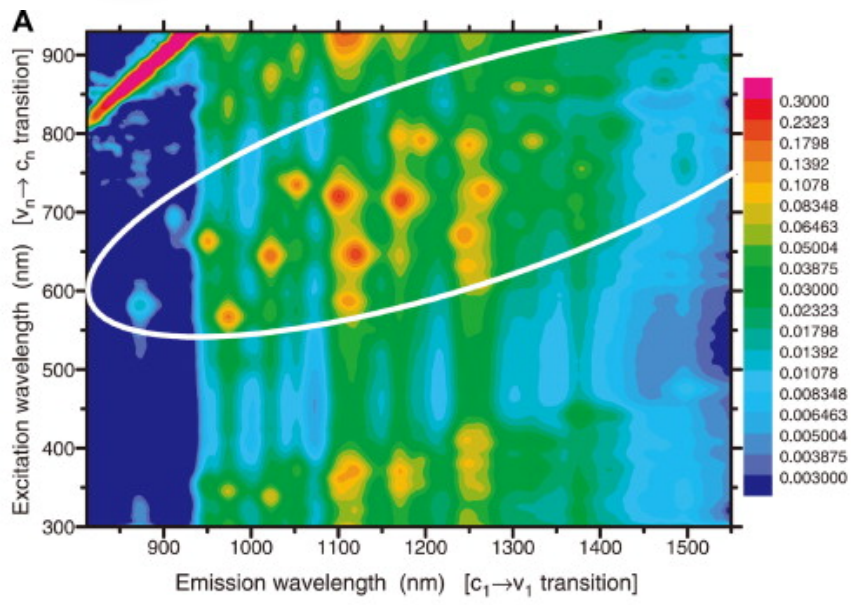
2. T. Guo, P. Nikolaev, A. Thess, D. T. Colbert, and R. E. Smalley, *Chem. Phys. Lett.* **243**, 49 (1995).



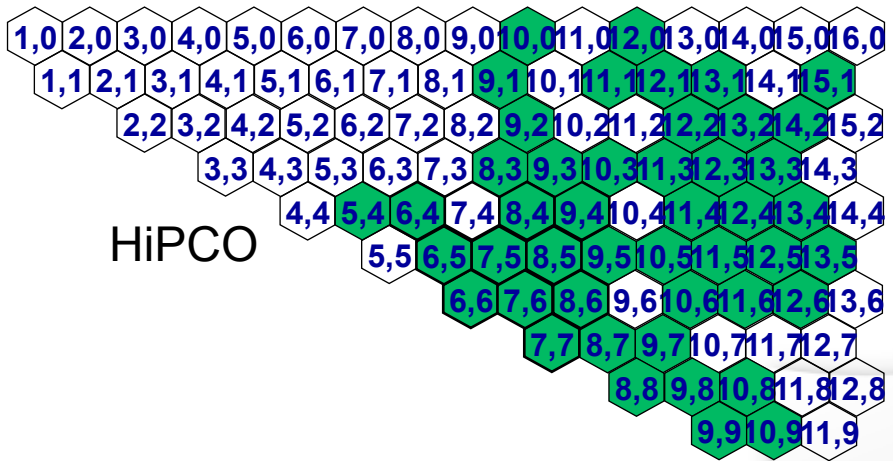
# Laser SWNTs (optical)



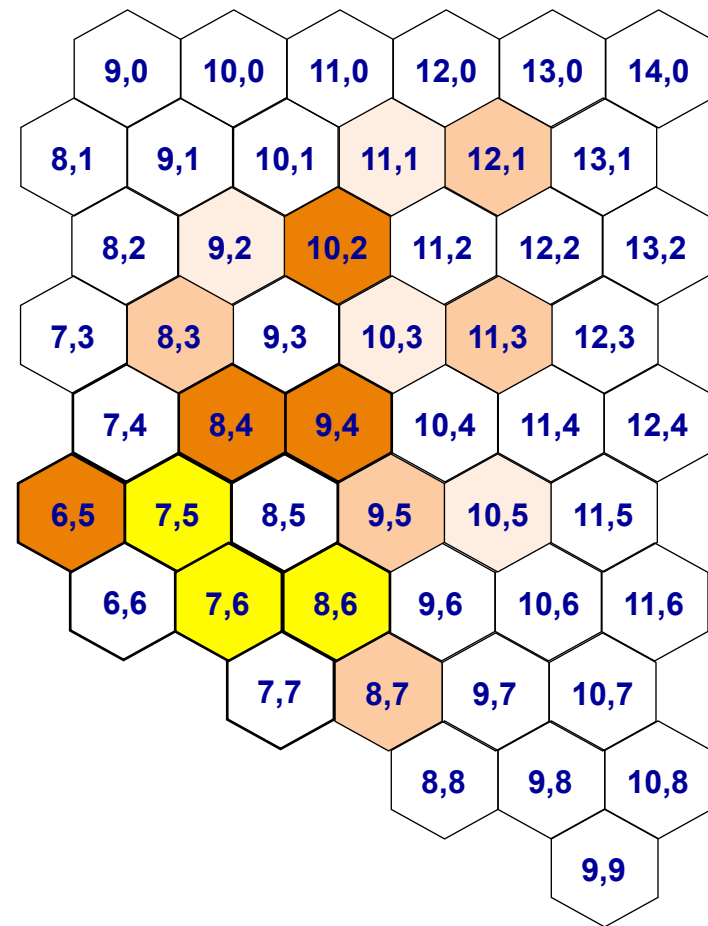
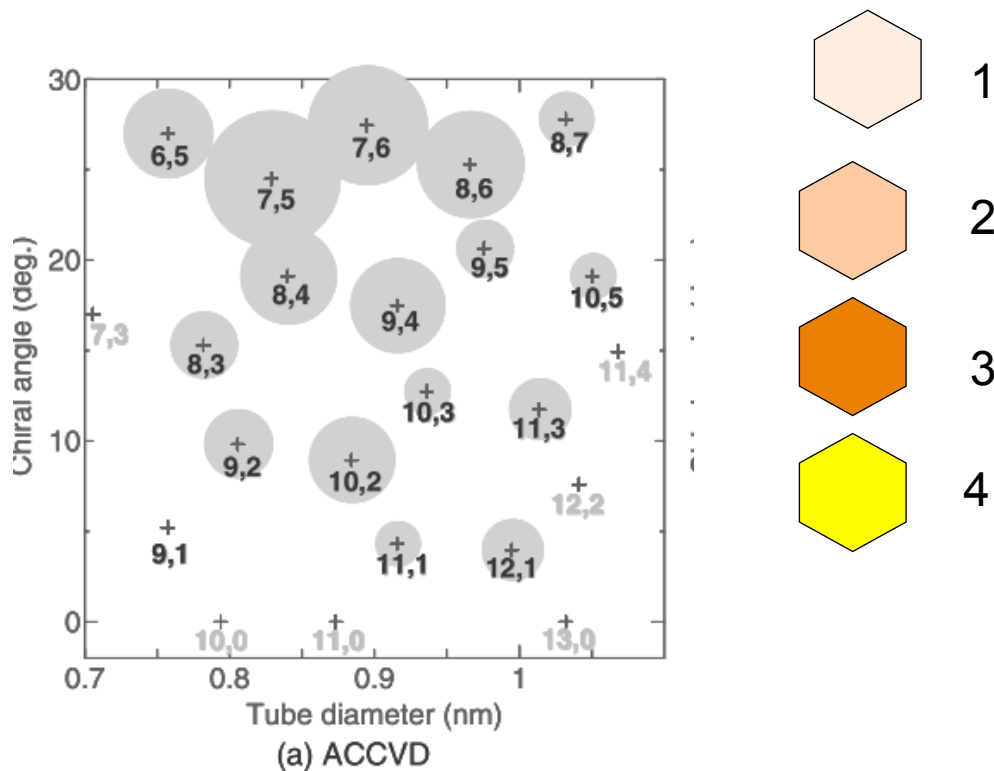
# High pressure CO FC-CVD SWNTs (HiPCO; optical data) – very broad vs. chiral angle when compared to ambient pressure FC-CVD



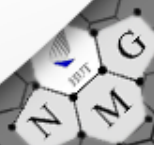
Include metallic tubes



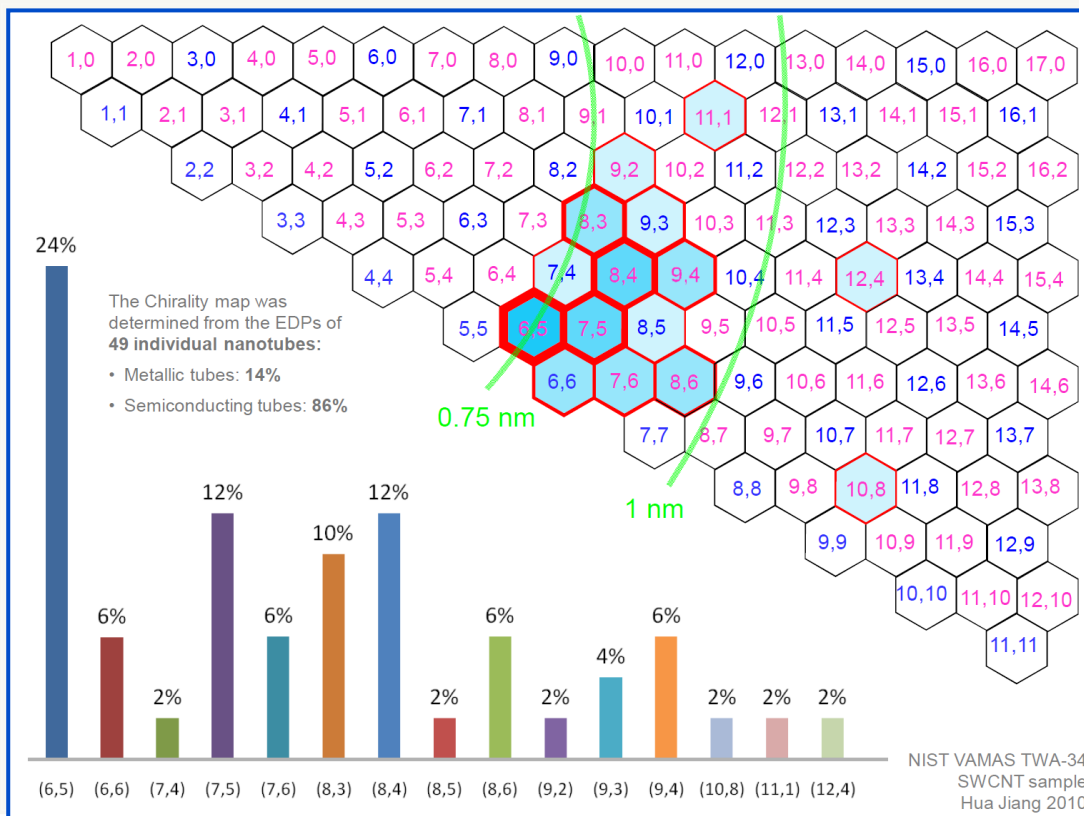
# Alcohol SWNTs (PL – did not find ED)



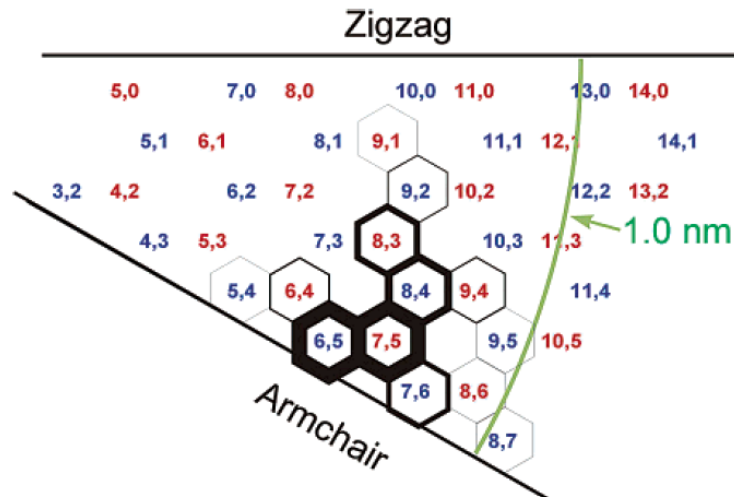
*New Journal of Physics* 5 (2003) 149.1–149.12



# CoMoCat – comparison of TEM/ED and PL



TEM/ED individual tube data -  
Jiang et al. unpublished

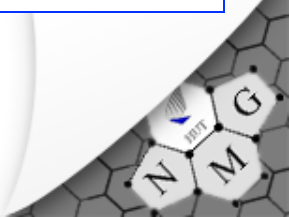
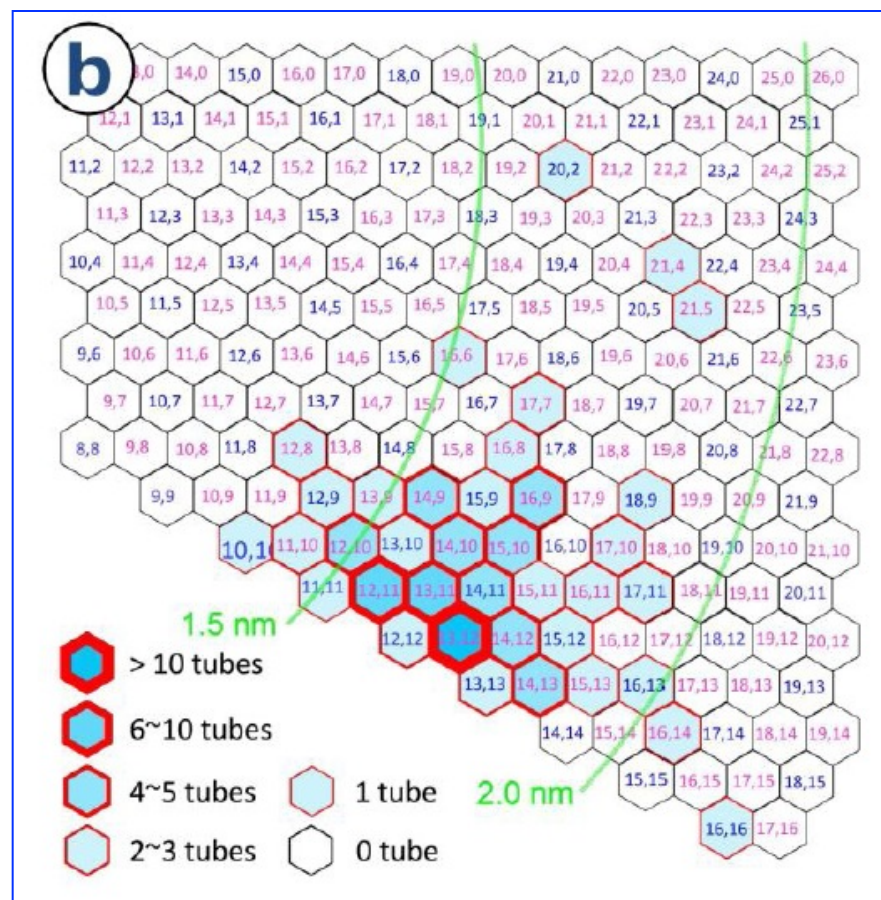
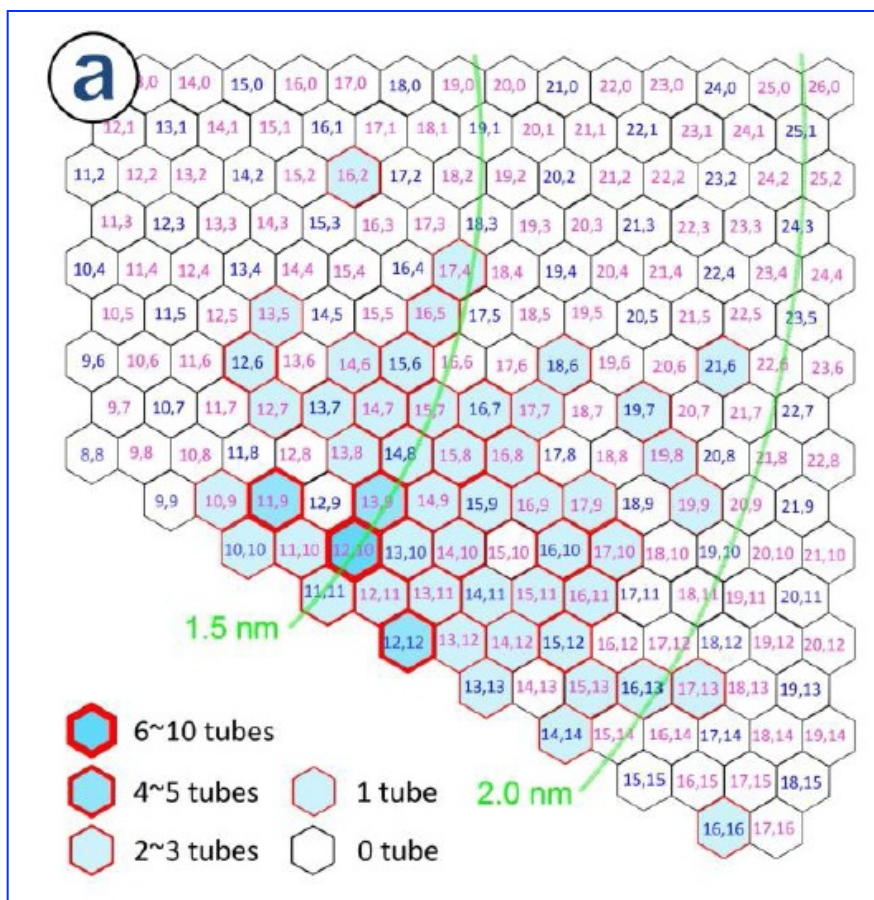


**Table 1.**  $(n,m)$ -Resolved Spectral Intensities from SWNT Samples

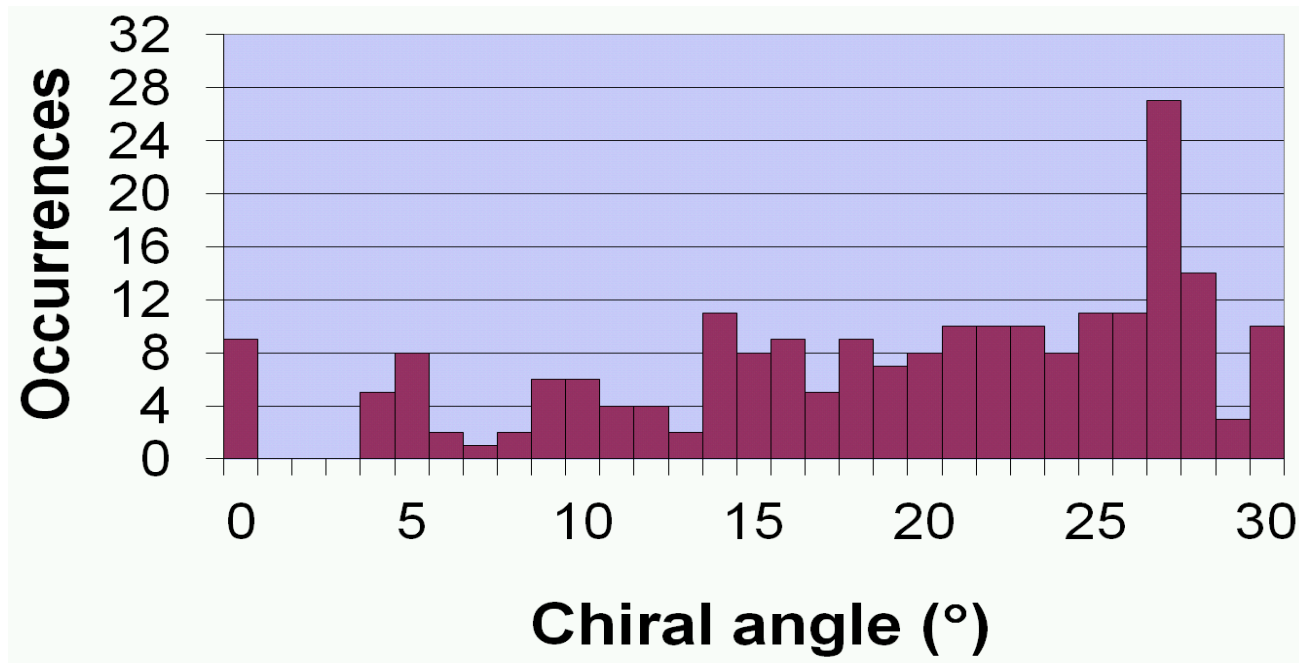
$n,m$	diameter (nm)	chiral angle (deg)	fractional intensity (%), CoMoCAT	fractional intensity (%), HiPco
5,4	0.620	26.3	0.3	0.0
6,4	0.692	23.4	2.8	0.3
9,1	0.757	5.2	0.8	0.2
6,5	0.757	27.0	28	3.7
8,3	0.782	15.3	11	2.9
9,2	0.806	9.8	1.7	0.4
7,5	0.829	24.5	28	4.9
8,4	0.840	19.1	14	4.2
10,2	0.884	9.0	0.0	4.5
7,6	0.895	27.5	8.5	7.1
9,4	0.916	17.5	2.3	7.6
10,3	0.936	12.7	0.0	4.3
8,6	0.966	25.3	0.8	8.3
9,5	0.976	20.6	0.3	5.7
9,5	0.976	20.6	0.0	5.7
12,1	0.995	4.0	0.0	3.8
11,3	1.014	11.7	0.0	4.6
8,7	1.032	27.8	0.3	5.6
10,5	1.050	19.1	0.0	4.6

Sergei M. Bachilo, Leandro Balzano, Jose E. Herrera, Francisco Pompeo, Daniel E. Resasco and R. Bruce Weisman  
**J. AM. CHEM. SOC.** 2003, 125, 11186-11187.

Chirality i.e.  $(n,m)$  maps for samples produced with (a) 0 ppm NH<sub>3</sub>, and (b) 500 ppm NH<sub>3</sub> as determined with electron diffraction of individual SWCNTs – *large chiral angle due to enhanced etching of low chiral angle tubes*

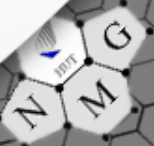


# Chiral angle distributions for the NIST VAMAS TWA-34 SWCNT sample (CoMoCAT)<sup>3</sup>



Chiral angle distribution for a CoMoCAT SWCNT sample. The measurement was based on **electron diffraction analysis** of 44 SWCNT bundles and 49 individual SWCNTs which results in a total occurrence of 269 helicities for statistic.

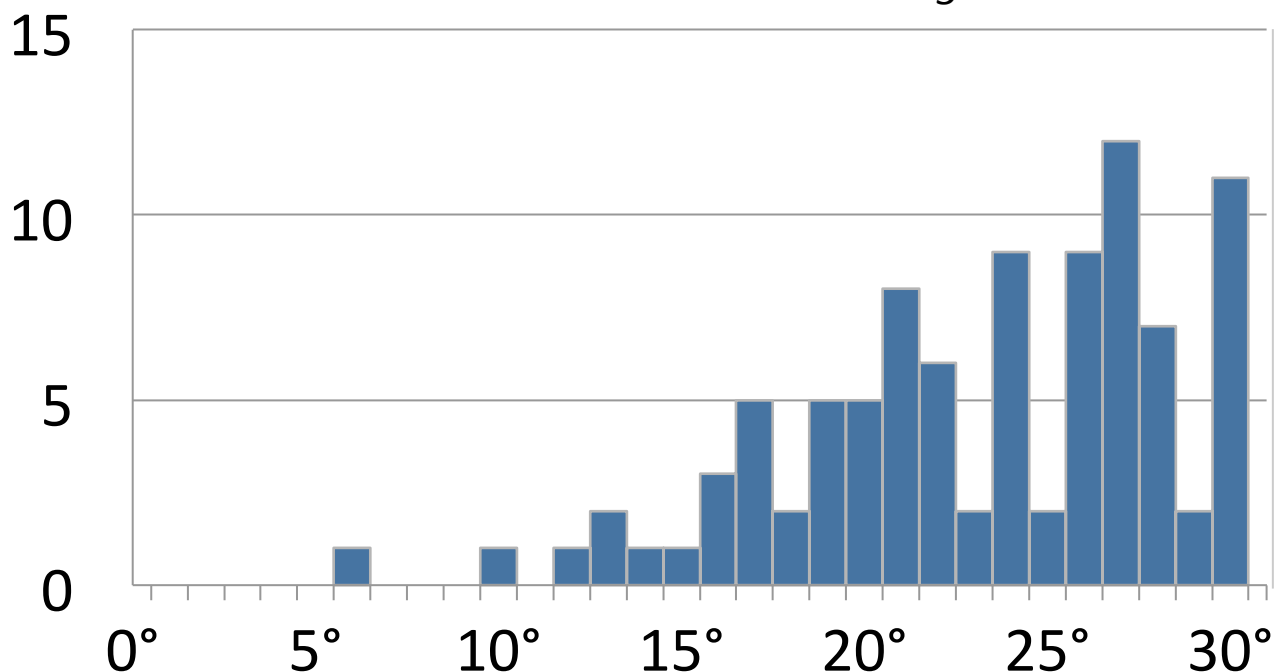
-- Jiang *et al.*, unpublished (2010)





# Chiral angle distributions for a SWCNT sample produced from CO by a floating catalyst (aerosol) CVD process

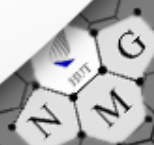
*introduced with 0 ppm NH<sub>3</sub> (blue)*



Chiral angle distribution for a SWCNT sample produced without NH<sub>3</sub> introduced. The measurement was based on electron diffraction analysis of totally **95** individual SWCNTs.

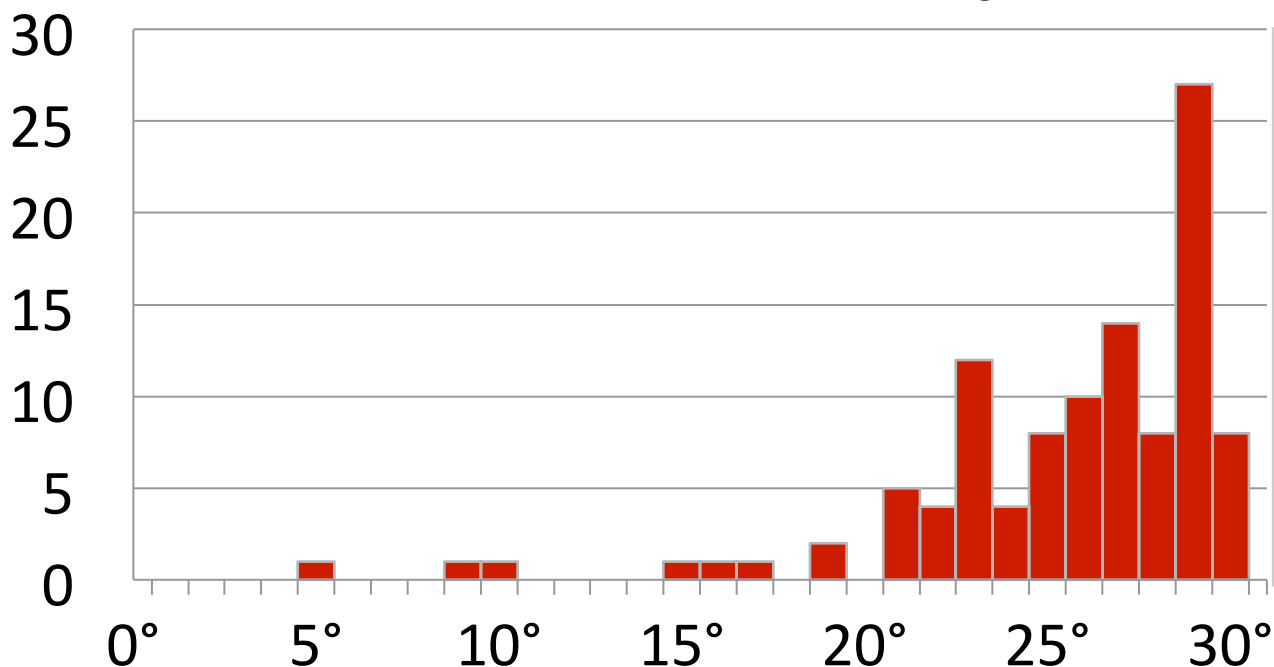
*CO flow rate@400 cm<sup>3</sup>/min, CO<sub>2</sub> flow rate@2 cm<sup>3</sup>/min, Temperature@ 880 °C*

-- Jiang et al., *JACS* **133**, 1224 (2010)



# Chiral angle distributions for a SWCNT sample produced from CO by a floating catalyst (aerosol) CVD process

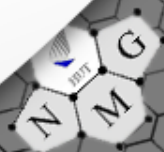
*introduced with 500 ppm NH<sub>3</sub> (Red)*



Chiral angle distribution for a SWCNT sample produced with 500 ppm NH<sub>3</sub> introduced. The measurement was based on electron diffraction analysis of totally **108** individual SWCNTs.

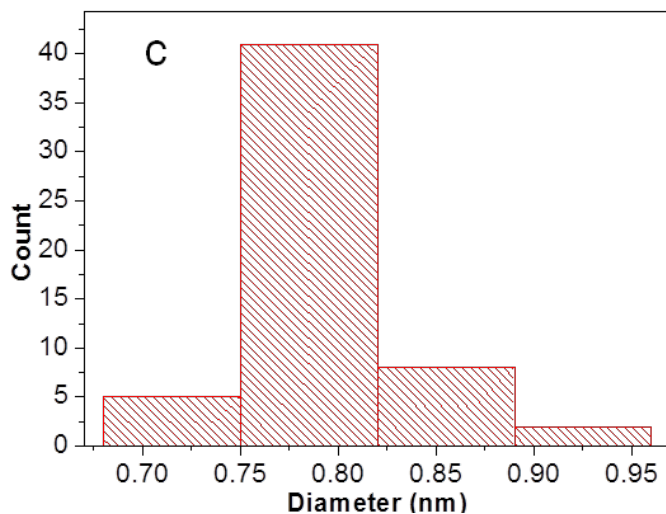
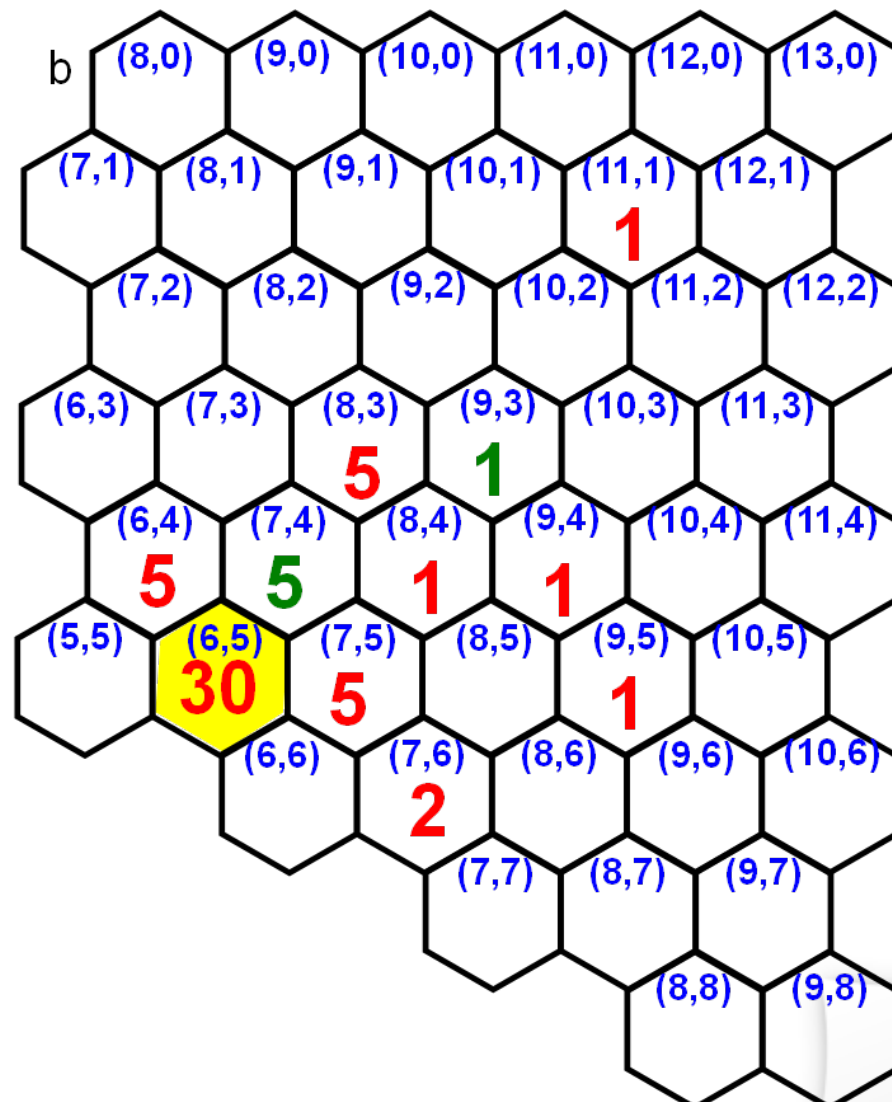
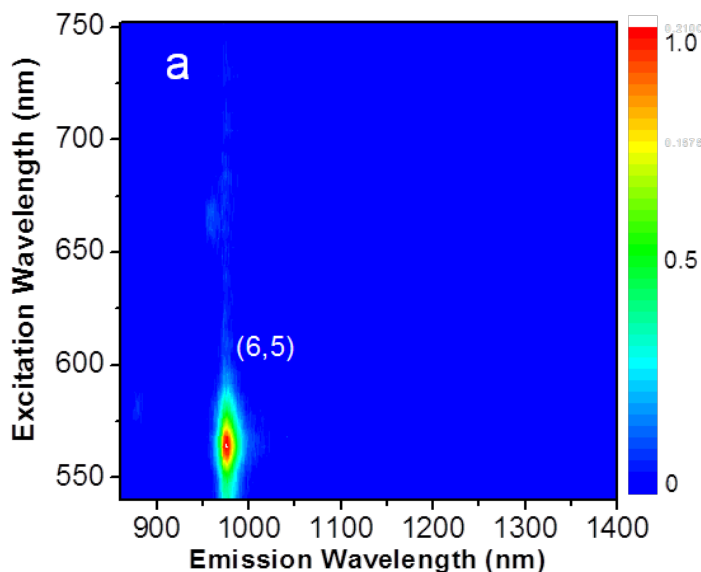
*CO flow rate@400 cm<sup>3</sup>/min, CO<sub>2</sub> flow rate@2 cm<sup>3</sup>/min, Temperature@ 880 °C*

-- Jiang et al., *JACS* **133**, 1224 (2010)

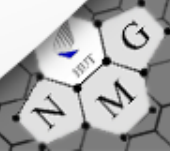


# Characterization of SWNTs grown at 500 C, 1 atm

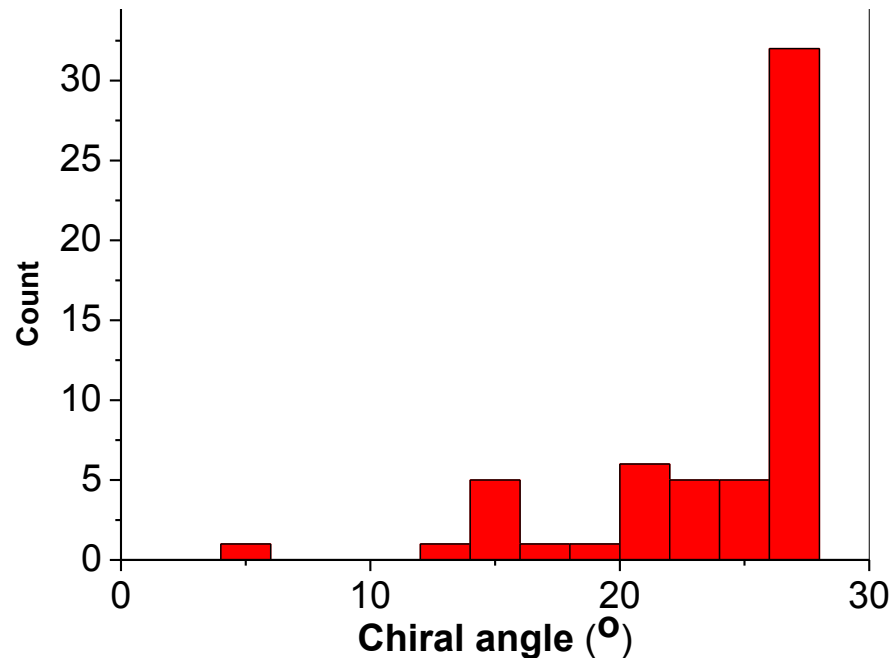
## Co-MgO II by impregnation



Preferential growth of semiconducting SWNTs (~90%)



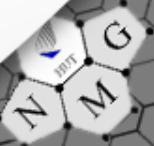
# Chiral angle distributions for a SWCNT sample grown on lattice-mismatched epitaxial Co Nanoparticles



***No armchair tubes !***

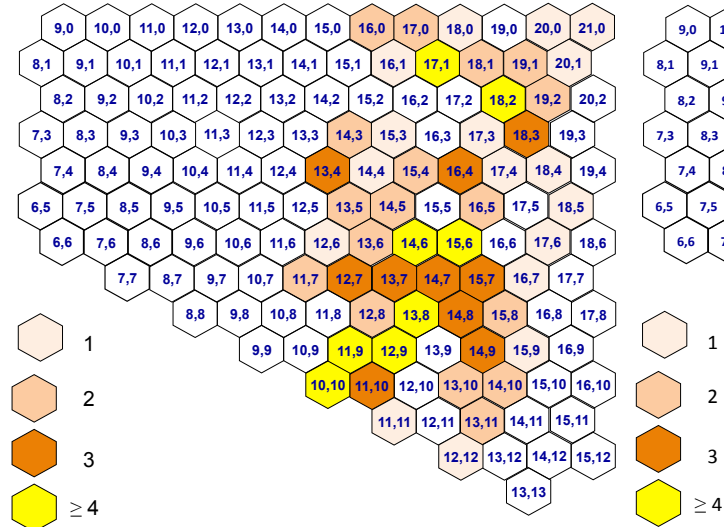
Chiral angle distribution for a Co grown SWCNT sample (500 °C). The measurement was based on **electron diffraction analysis** of 57 individual SWCNTs.

-- He et al., *Scientific Reports* **3**, 1460 (2013)

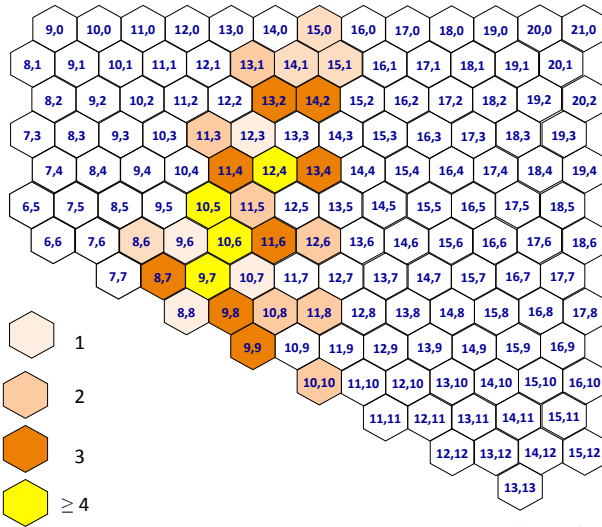


# (n,m) maps of SWNTs via ED

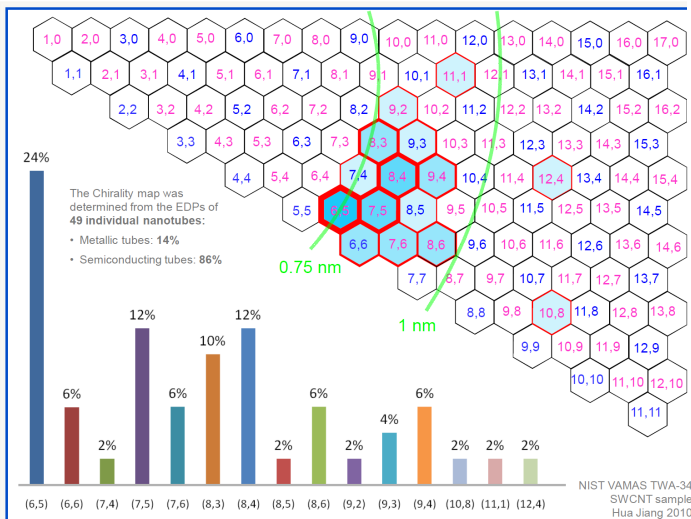
## Arc SWNTs



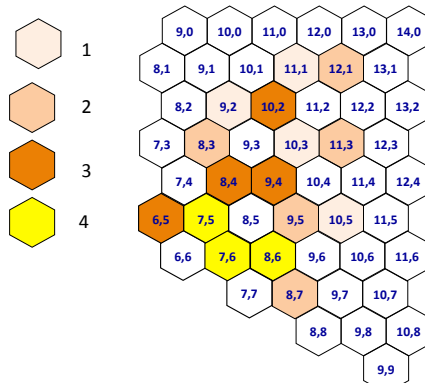
## Laser SWNTs (optical)



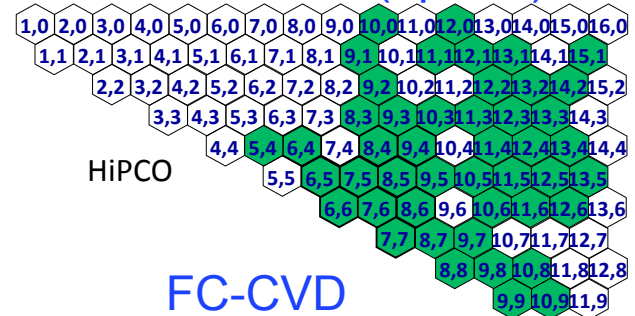
## CoMoCAT



## A-CVD (optical)

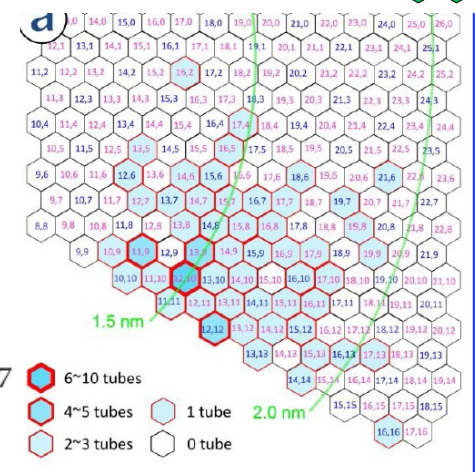


## HiPCO (optical)



## HiPCO

## FC-CVD

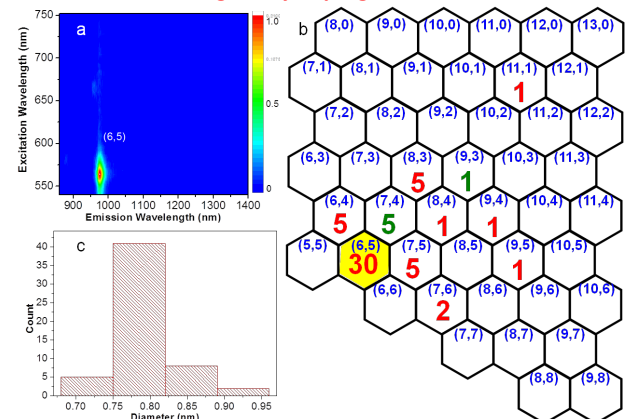


Nano Res (2009) 2: 818-827

- 6~10 tubes
- 4~5 tubes
- 2~3 tubes
- 1 tube
- 0 tube

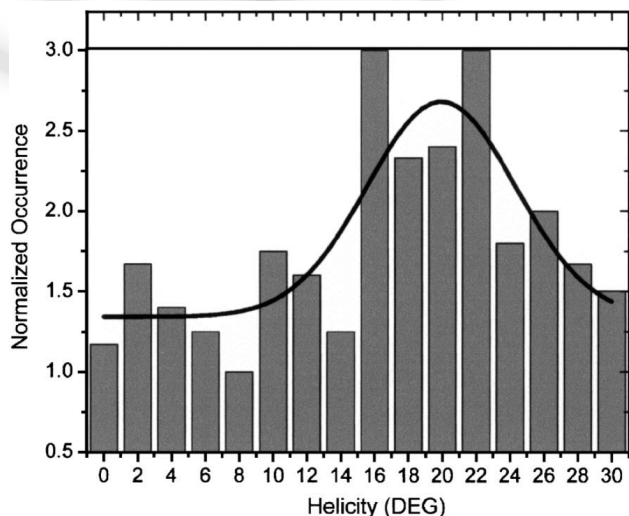
## CVD – "locked" catalyst

Co-MgO II by impregnation

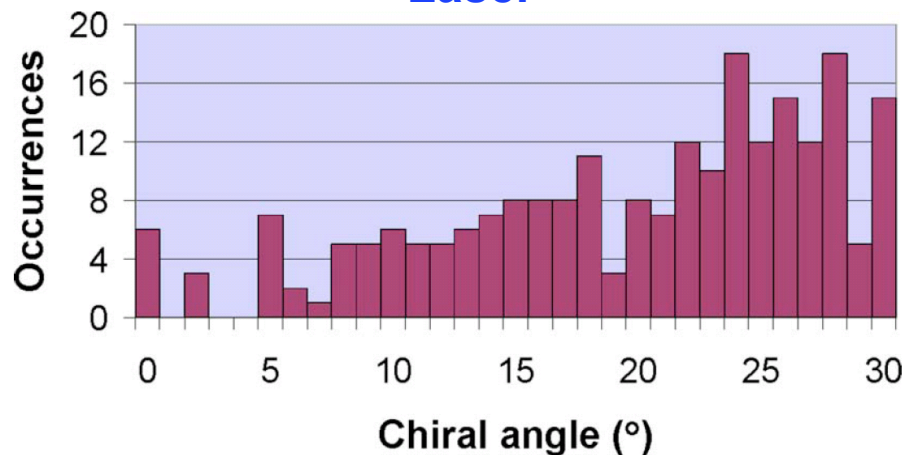


Preferential growth of semiconducting SWNTs (~90%)

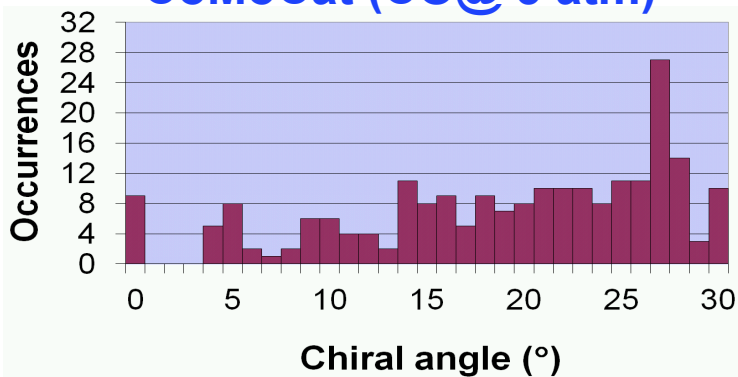
## Arc



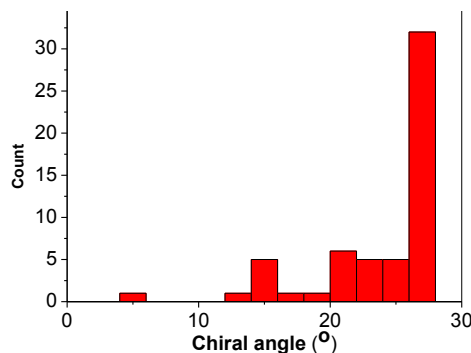
## Laser



## CoMoCat (CO @ 5 atm)

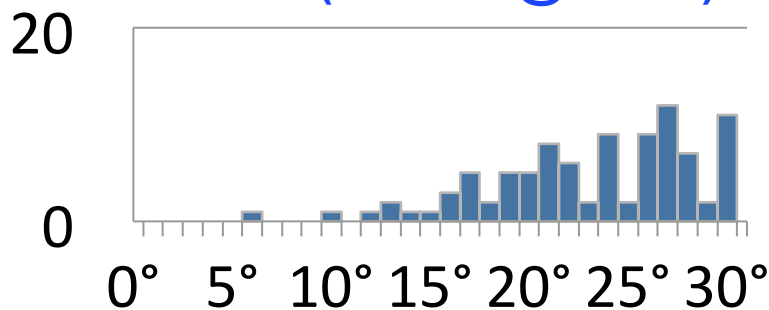


## Lattice-mismatched epitaxial Co on MgO CO @ 1 atm, 500 oC

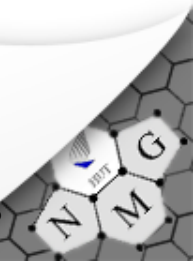
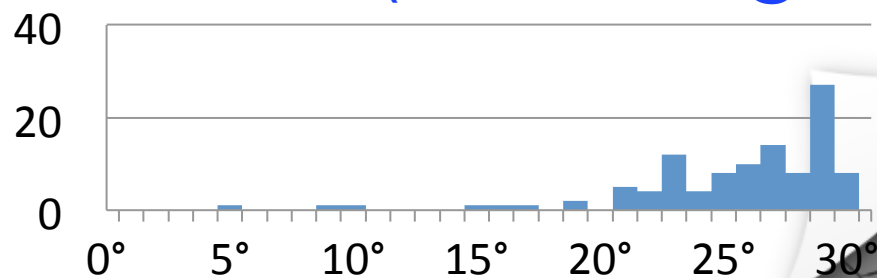


**Peking U. Prof. Y. Li  
with faceted catalyst:  
close to just one (n,m) ?**

## FC-VD (Fe + CO @ 1 atm)



## FC-CVD (Fe + CO + NH3 @ 1 atm)



# New Project



# EU-Japan 2013-17

The University of Tokyo

Aalto University

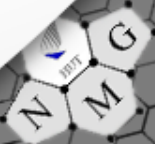
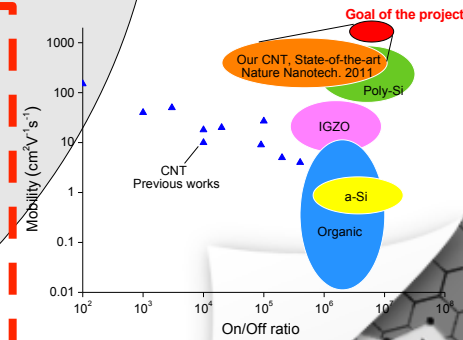
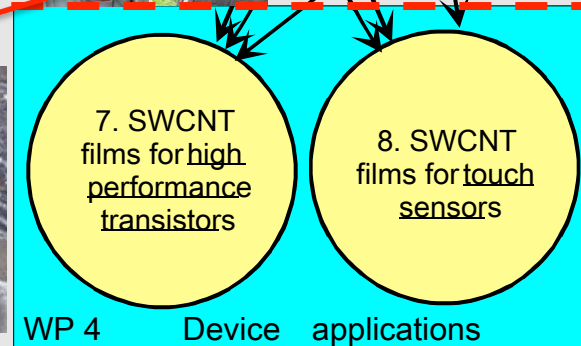
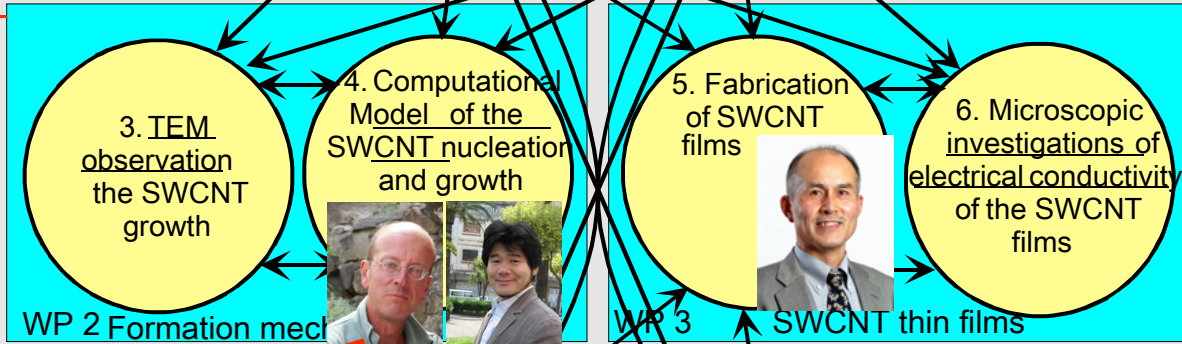
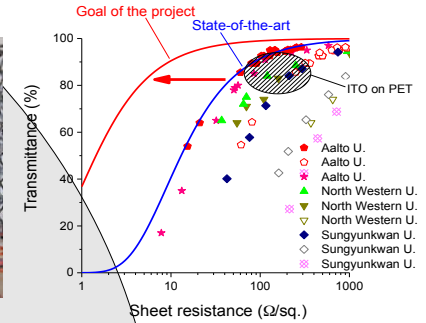
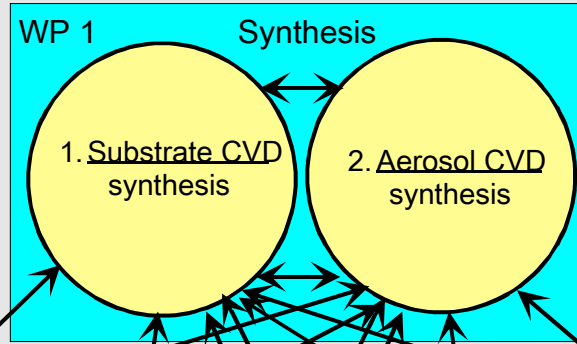
Technical University of Denmark

Dr. J. B. Wagner



CNRS, France  
Dr. C. Bischara

Nagoya University



**OPEN POSITIONS!** Department of Applied Physics **OPEN POSITIONS!**



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\* Aalto MIDE and AEF programs

Prof. Y. Ohno, Nagoya University, Japan

