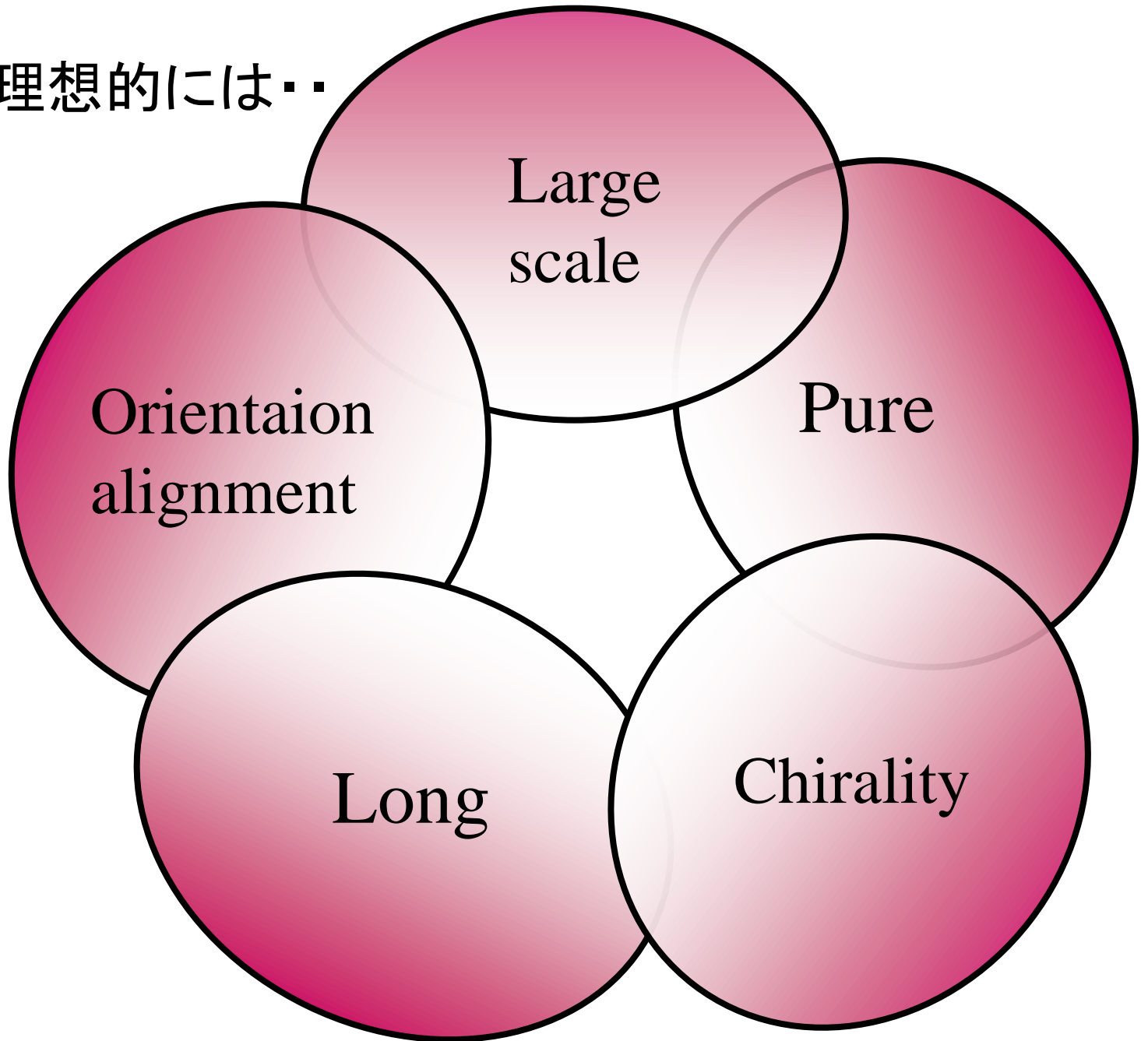
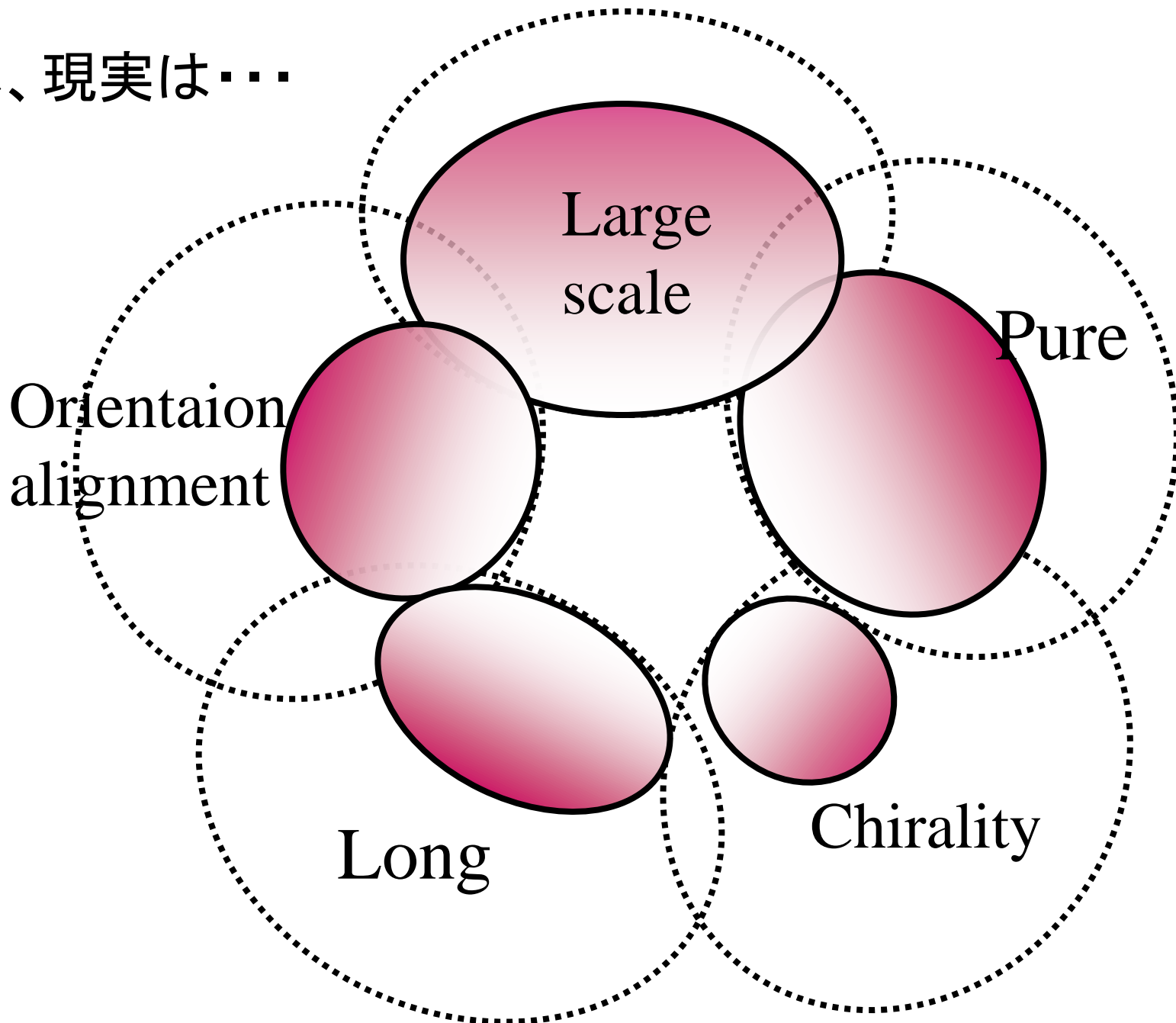


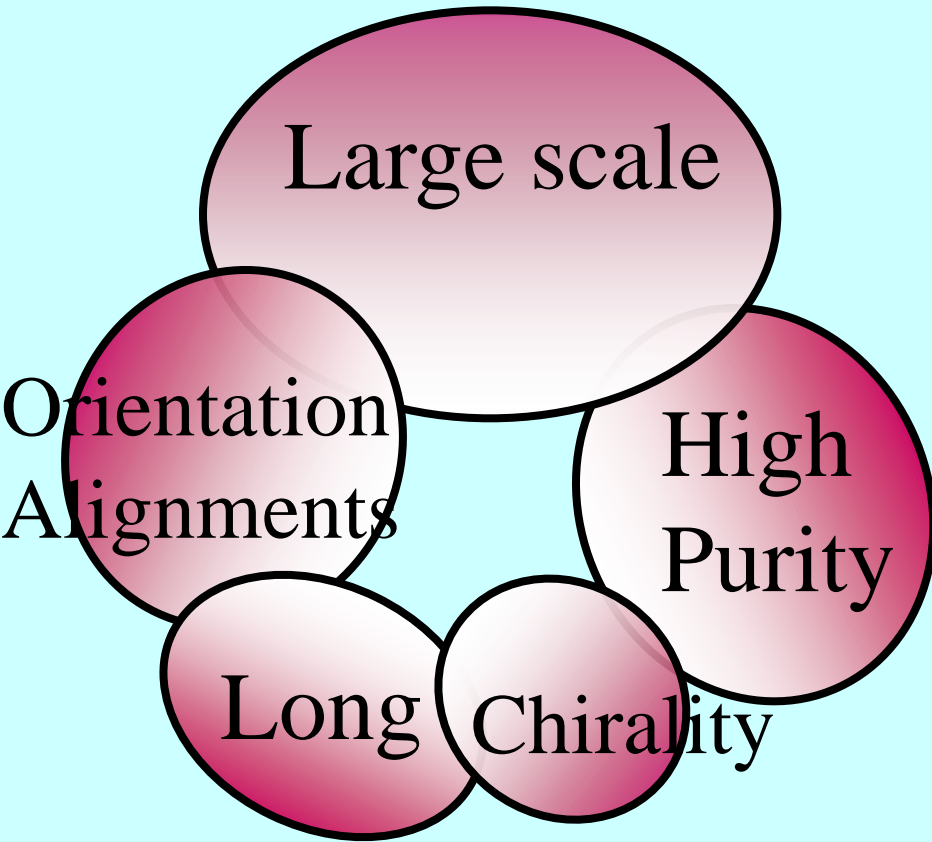
SWNT: 理想的には・・・



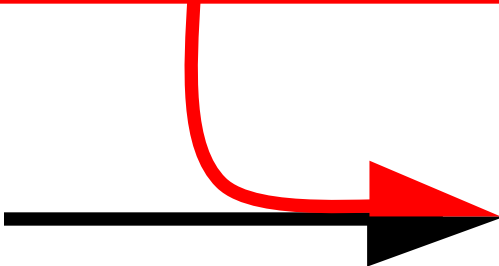
しかし、現実には...



SWNT Production: Current Status

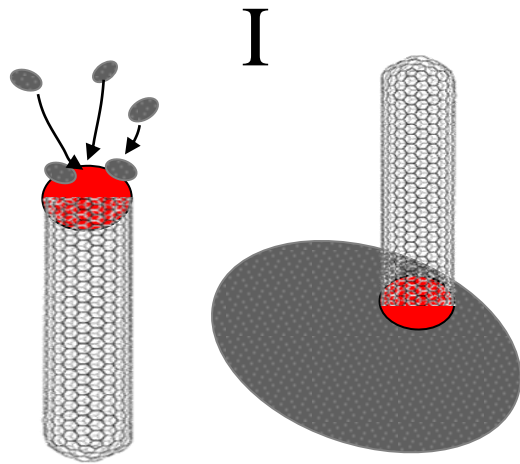


Understanding
Formation mechanism



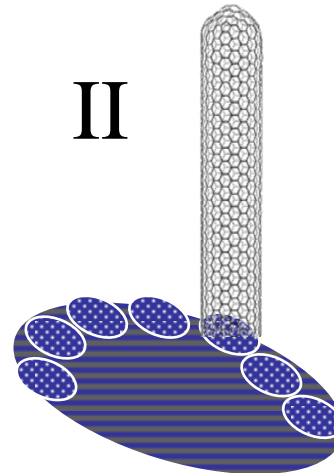
*Ideal
Status
of
Production*

Growth Mechanism of SWNTs



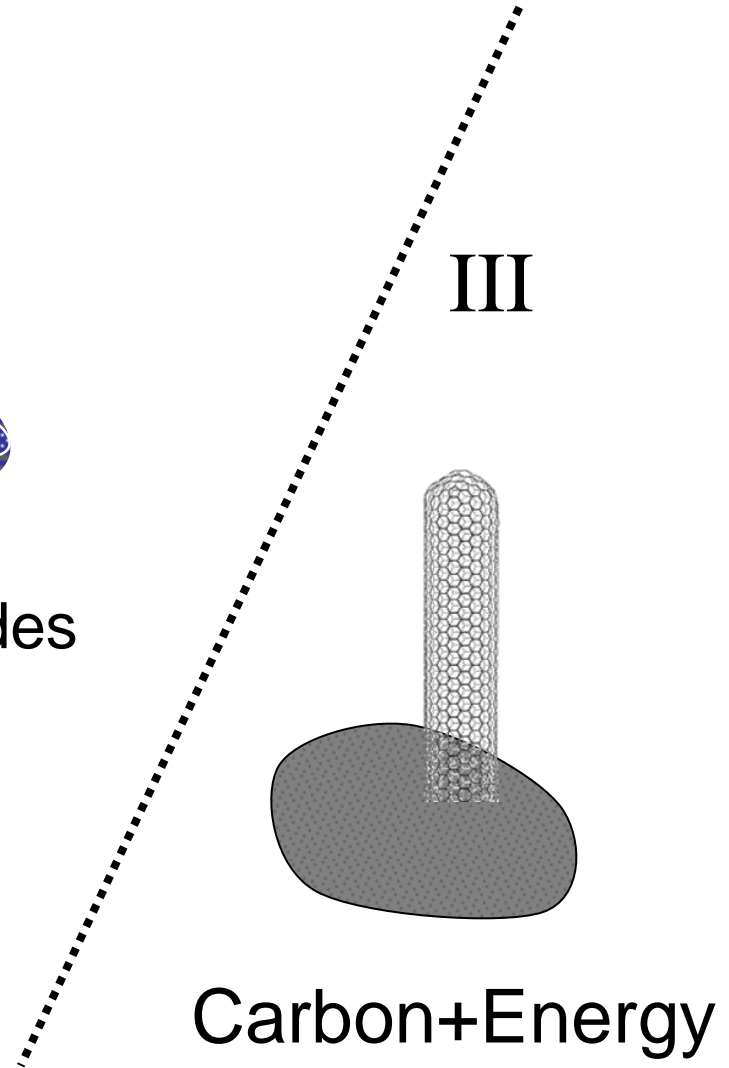
Metals

Catalysts
(~nm)

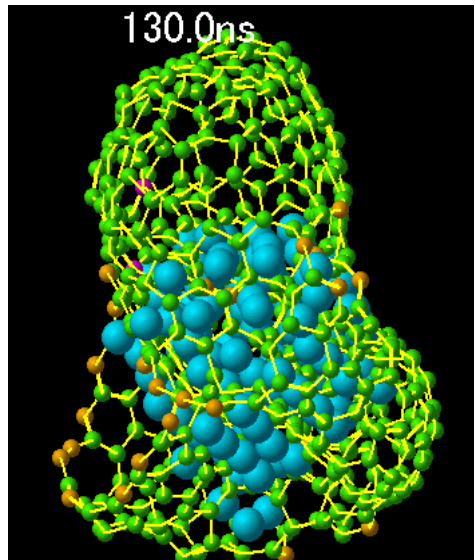
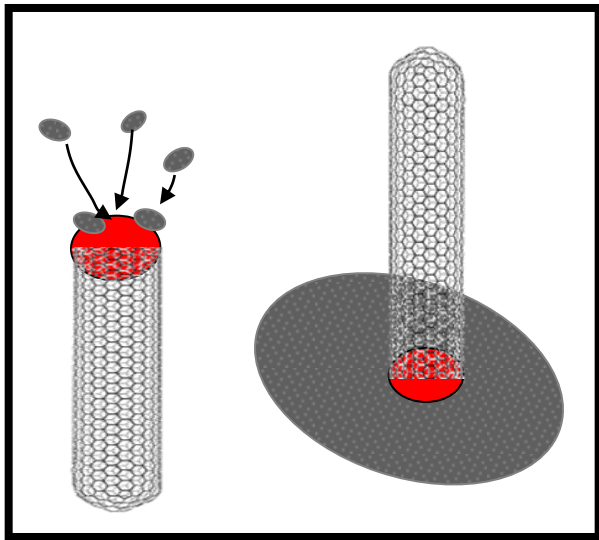


Templates

- Metal Carbides
- Diamond
- Pores

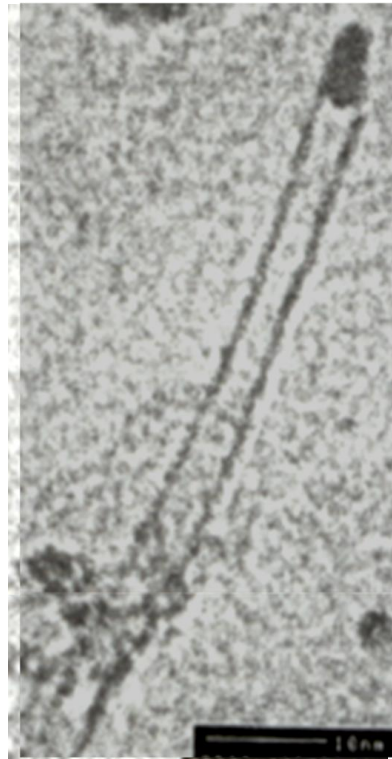


Carbon+Energy

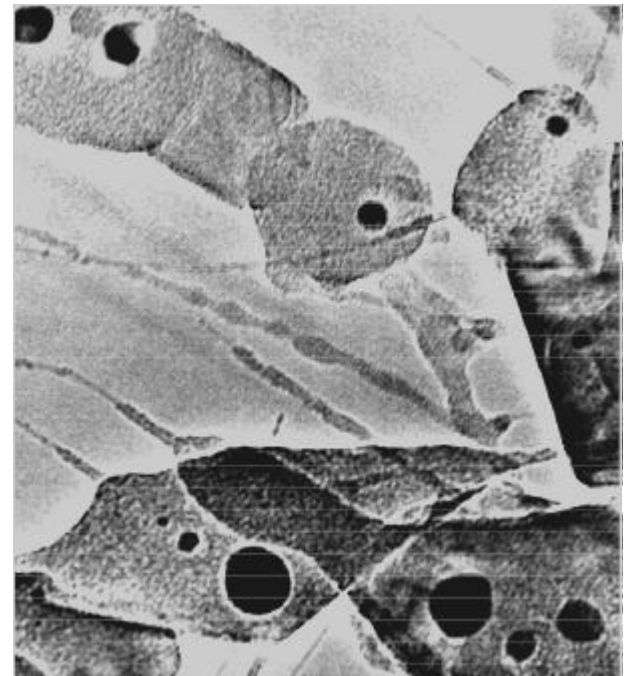


Shibuta, Maruyama (2002)

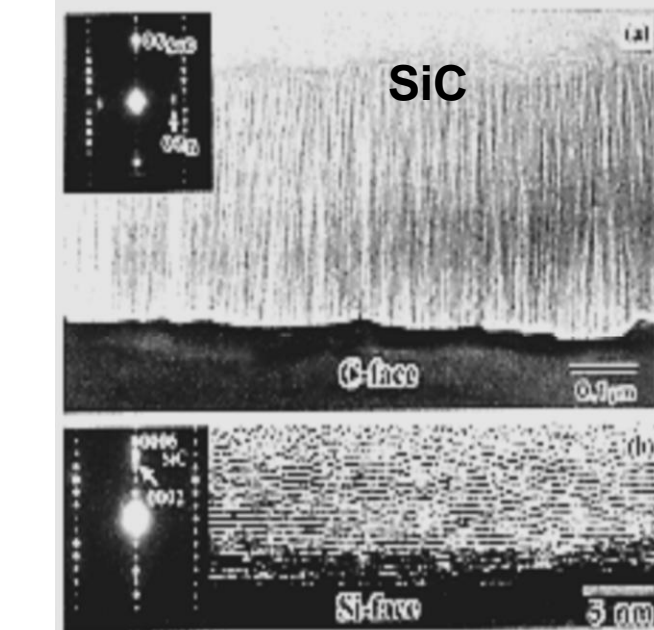
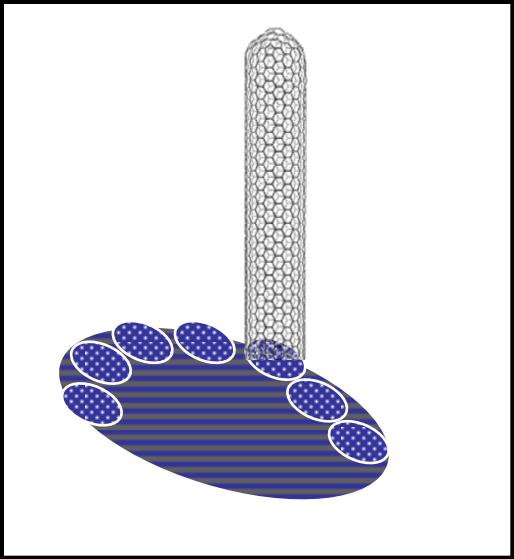
500 Carbon
&
 Ni_{108} : 2500K



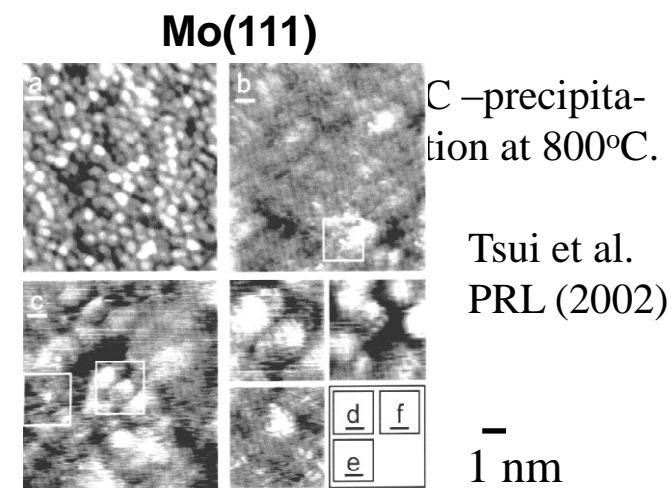
H.Dai et al.
Chem. Phys. Lett.
(1996).



Yudasaka et al.
JPC B 1998.



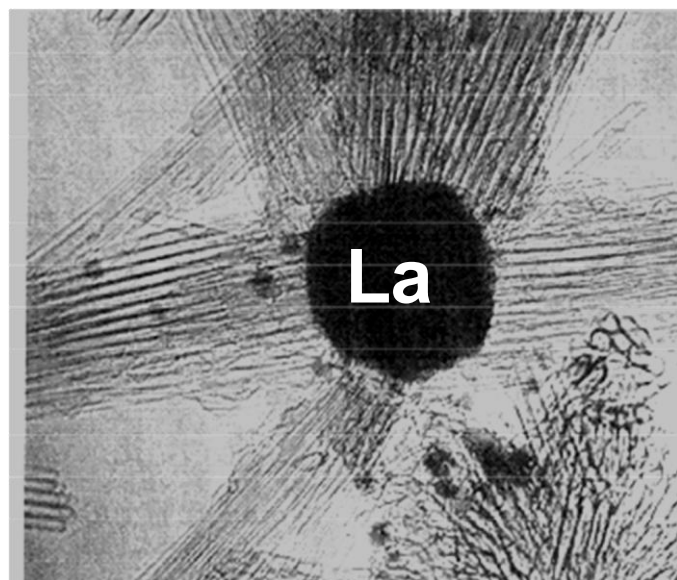
M. Kusunoki et al. Physica B (2002).



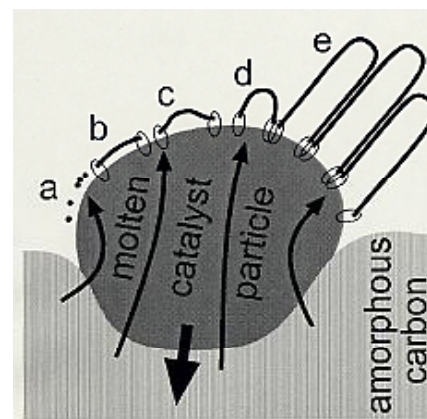
C-precipitation at 800°C.

Tsui et al. PRL (2002)

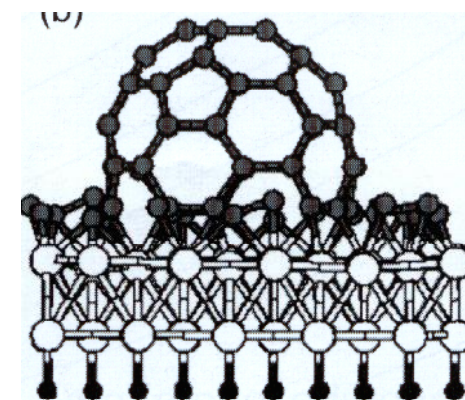
1 nm



Y. Saito et al. Chem. Phys. Lett. (1995).



Gorbunov et al, Carbon (2002).



J. Gavillet et al. Carbon (2002).

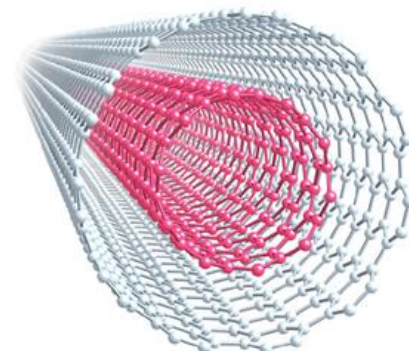
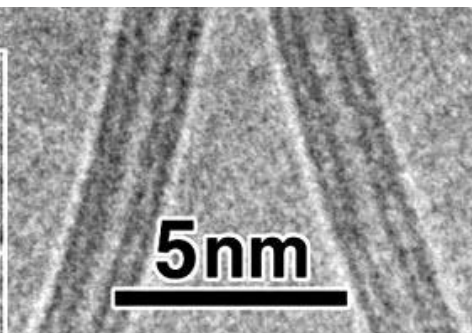
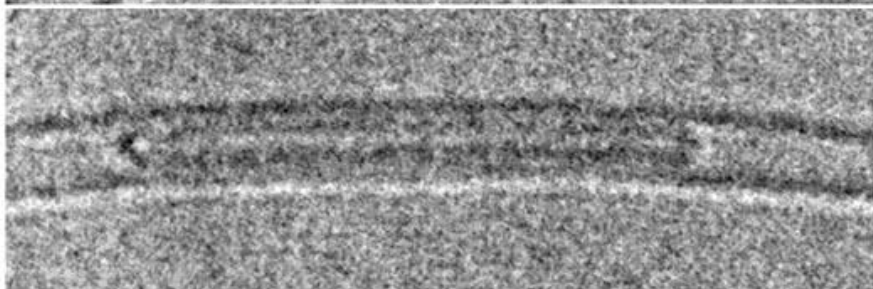
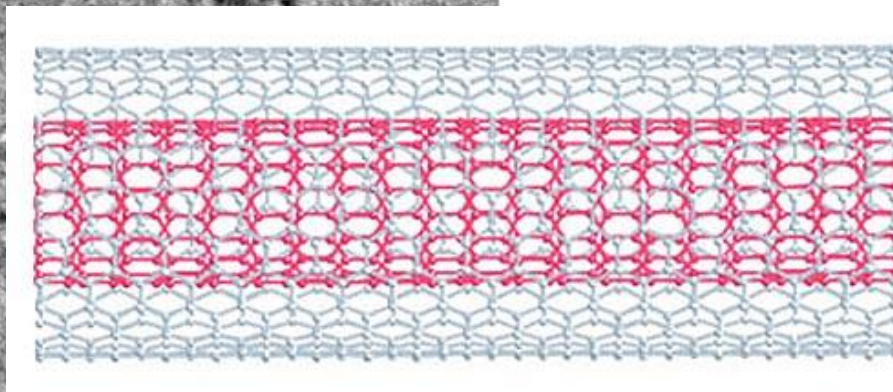
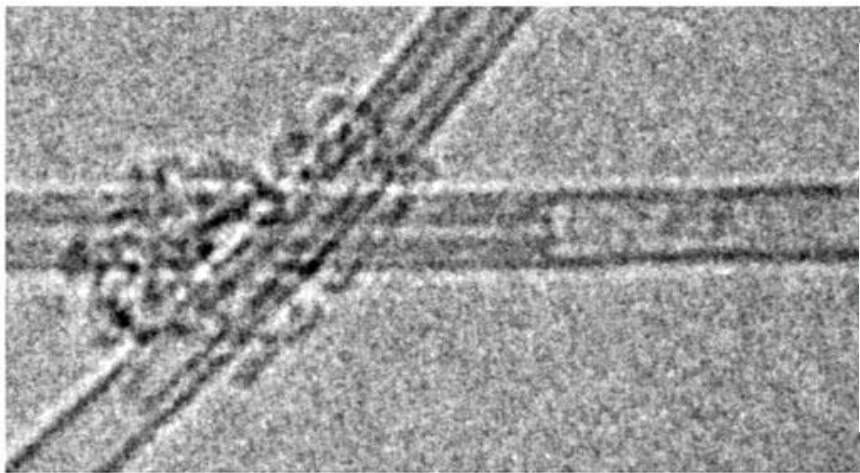
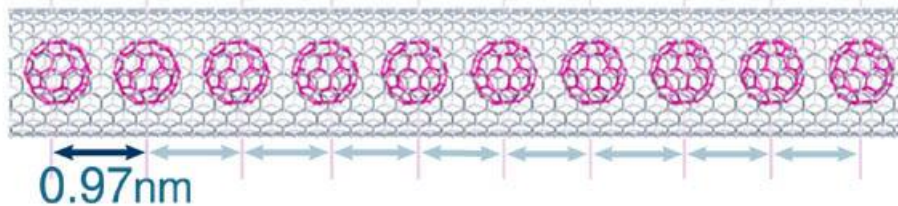
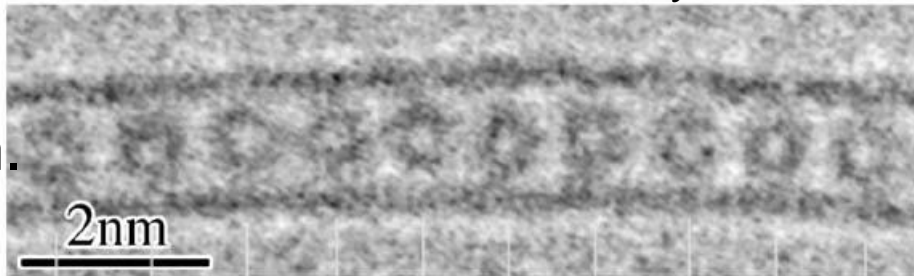
Growth Mechanism

Bandow et al. Chem. Phys. Lett. 2001

Evidence of Template Growth.

- ◆ *Phase diagram?*
- ◆ *Catalyst?*
- ◆ *New type?*

HT 1200°C

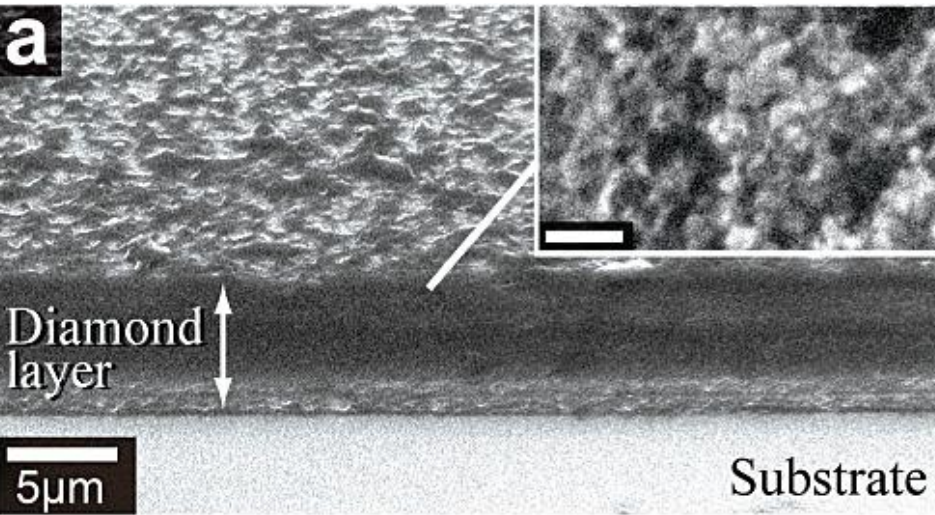


Carbon Nanotube Growth from Diamond

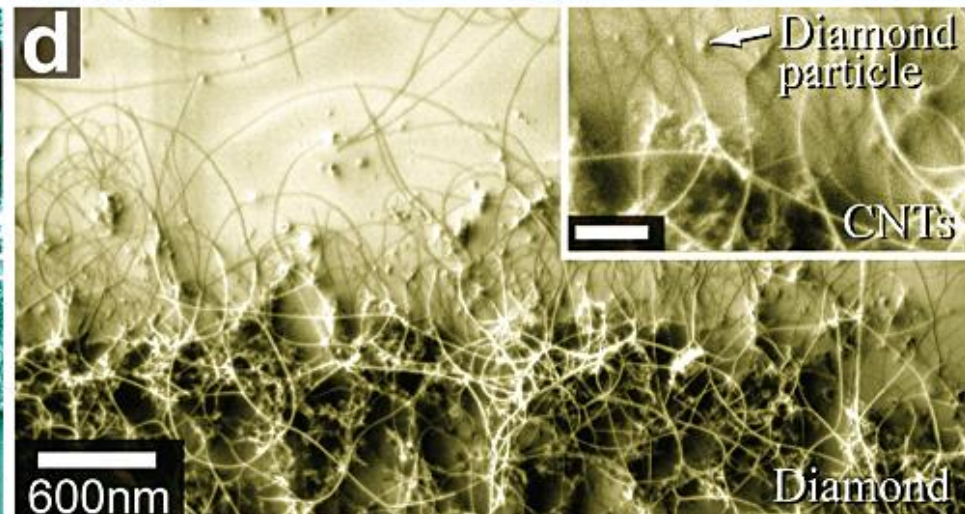
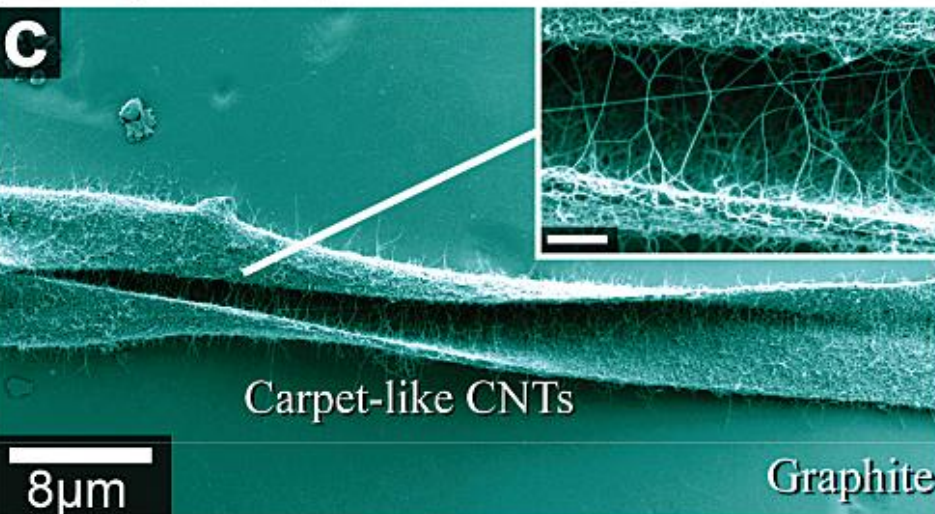
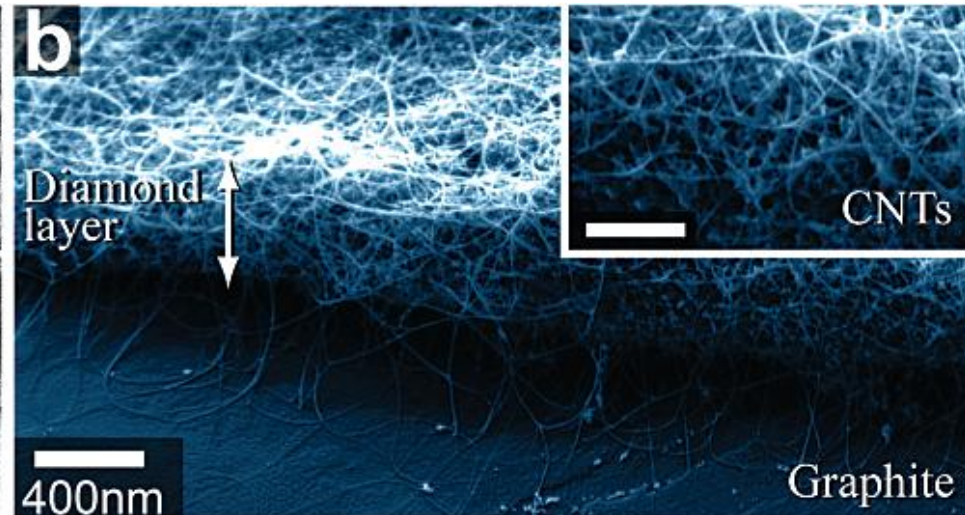
Takagi, Kobayashi, Homma, JACS 2009

EtOH CVD 850°C

Nanodiamond



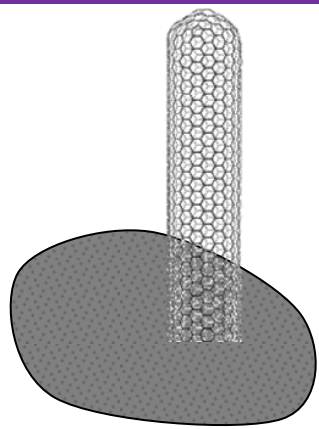
CNT/Nanodiamond/Graphite



CNT/graphite

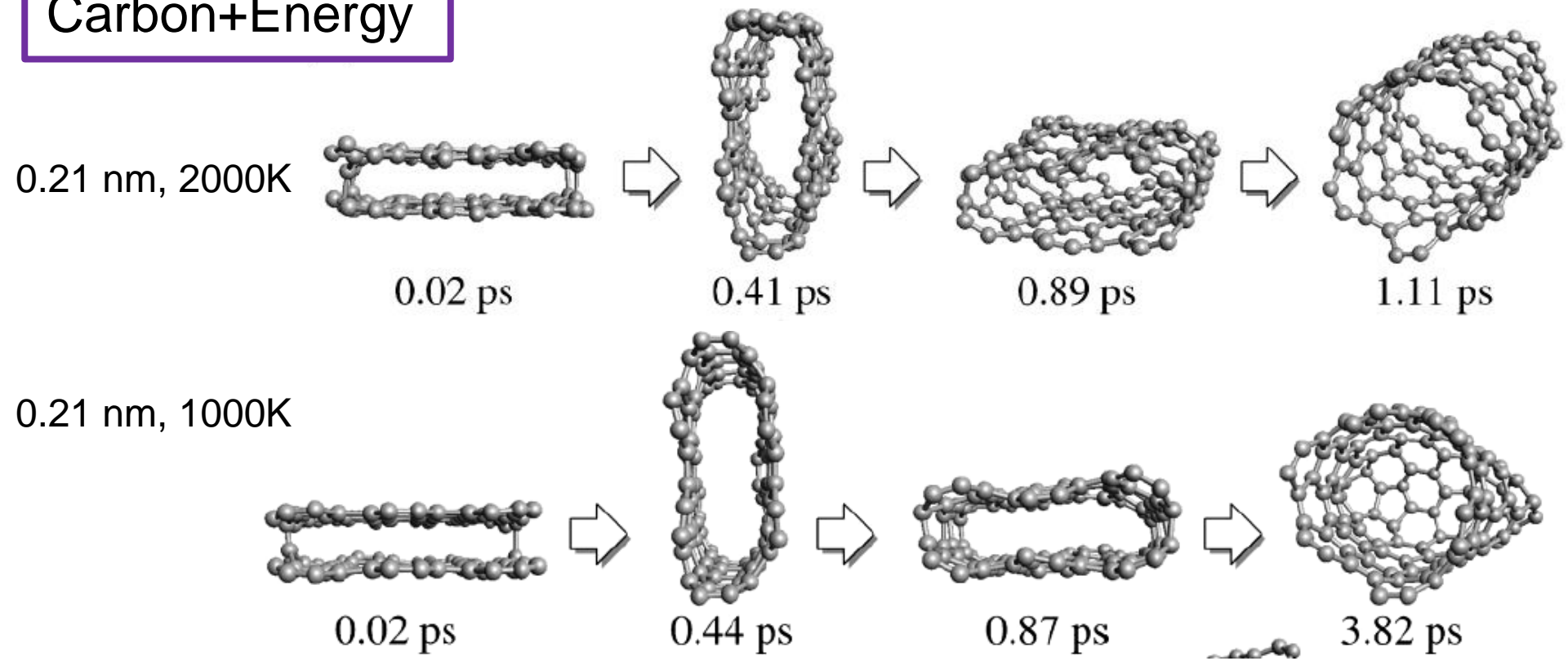
CNT/Nanodiamond/diamond

Nucleation of SWNT Growth Without Metal Catalysts and Templates.



Carbon+Energy

Kawai et al. *Phys. Rev. B* **66**(2002)033404



Cap formation probability is about 50% at 1000K.

It decreases at higher temperature.

*Zhang & Crespi *PRB* **45**(1992)12227

SWNTを製造するためには、

- 炭素(源)を供給する。
 - 反応エネルギーを与える。
-
- 金属触媒の粒子を作る(溶解・析出)
SWNT 直径1~2 nm, (MWNT; 直径3~20 nm)
 - 鑄型を使う(表面同士の相互作用)
Channels in AlPO_4-5 (Z.K. Tang et al, Nature 408(2000)50
Appl. Phys. Lett. 73(1998)2287.) *Metal free growth : Pyrolyzing tripropylamine in the channels of the zeolite crystals. 0.4-nm SWNTs are formed which show superconductivity.*
SiC, SWNT, Diamond, et al

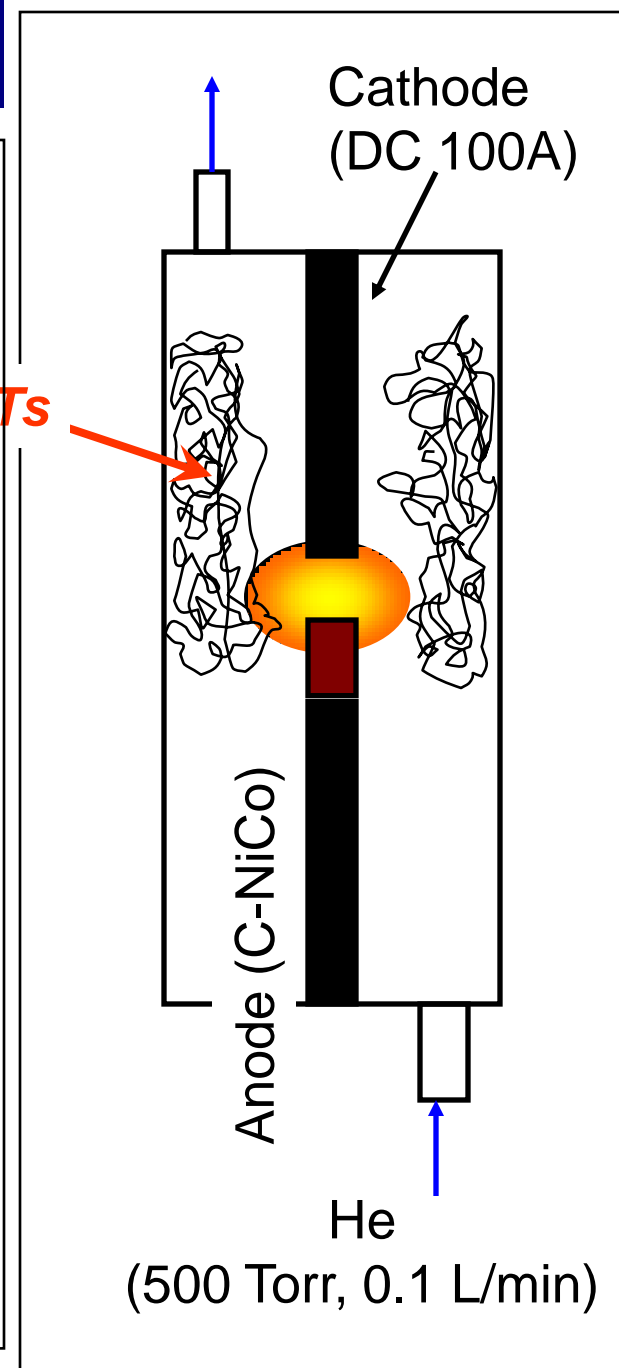
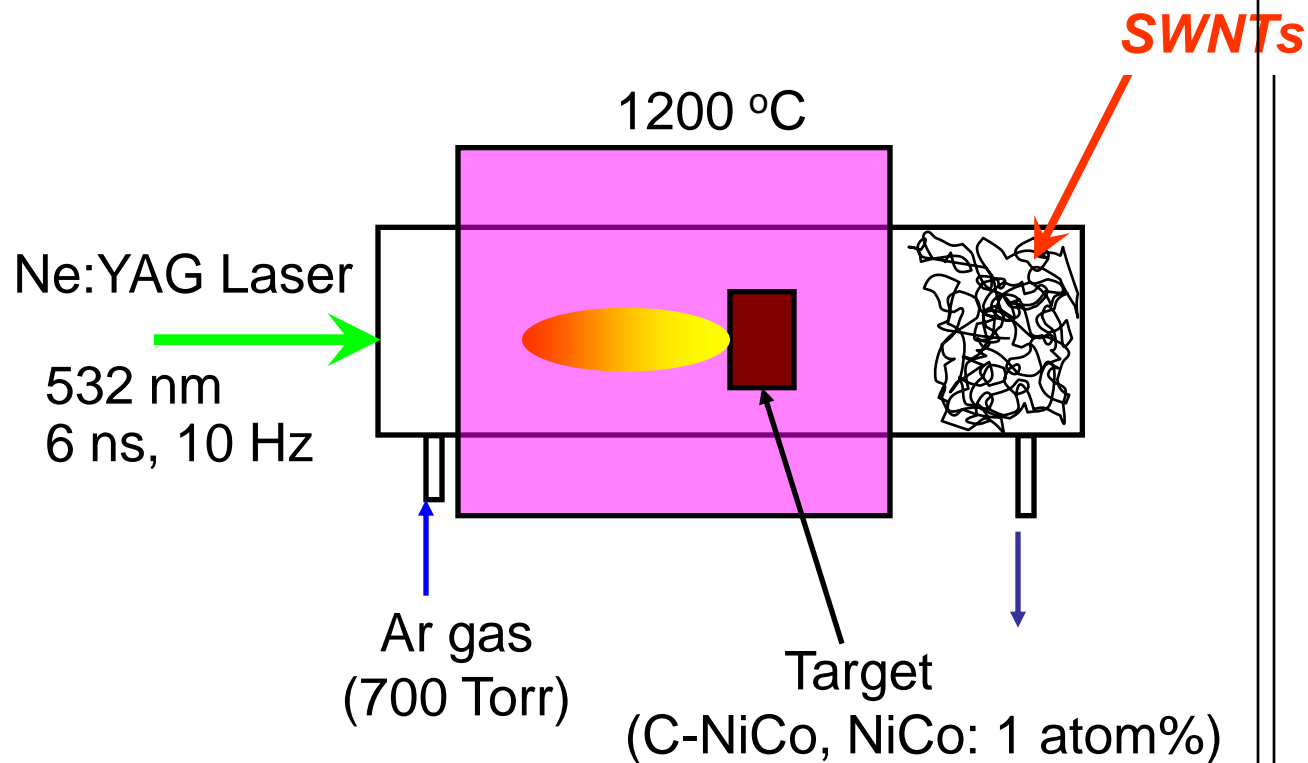
レーザーアブレーション法
アーク放電法
CVD法

Formation of Single-Wall Carbon Nanotubes by Laser Ablation and Arc Discharge

Production rate: ~50 mg/h

Purity : 20~30%

Tube diameter : 1.2~1.5 nm



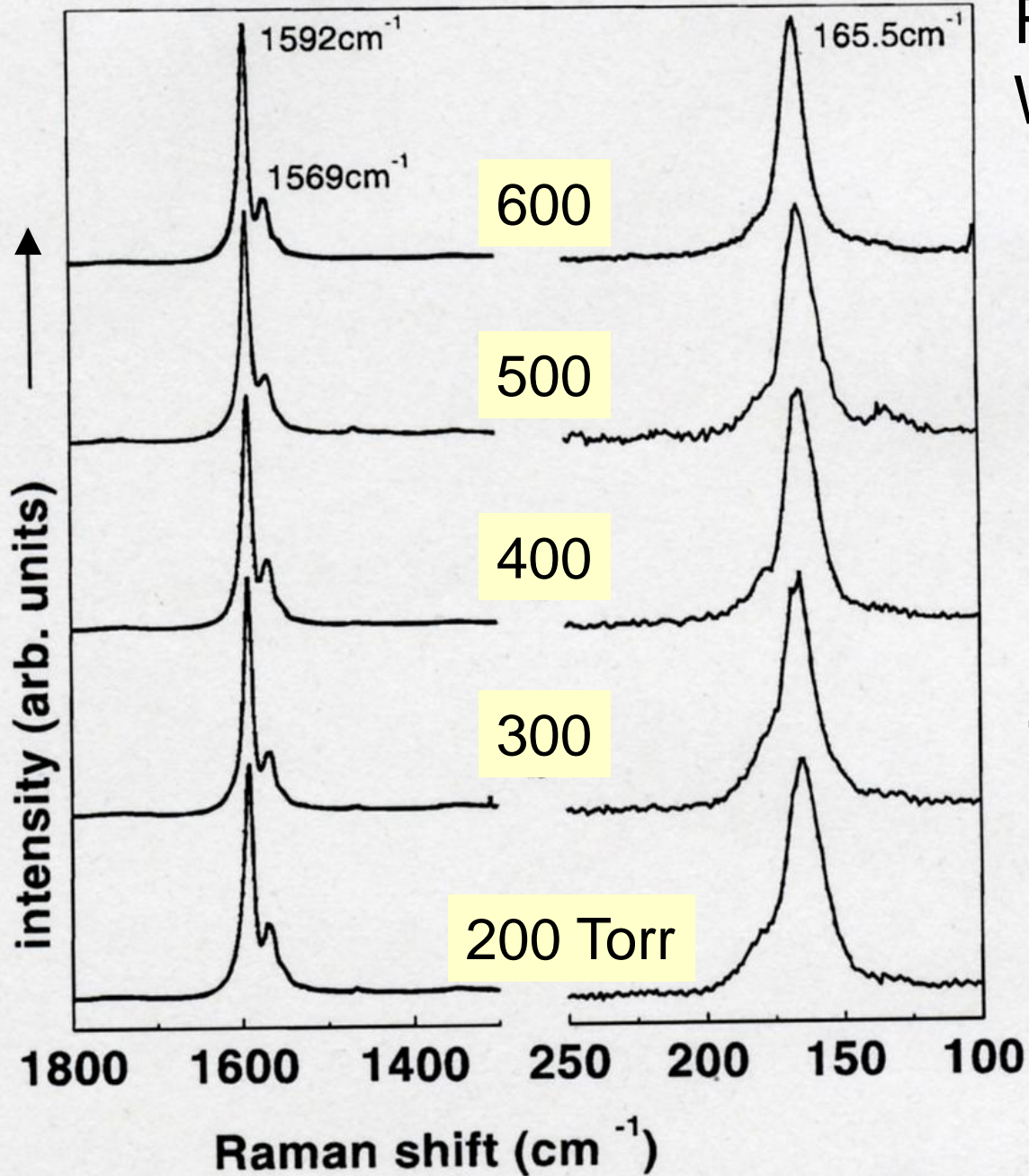
Guo et al. *Chem. Phys. Lett.* 243, 49, **1995**.

Thess et al. *Science* 273, 483, **1996**.



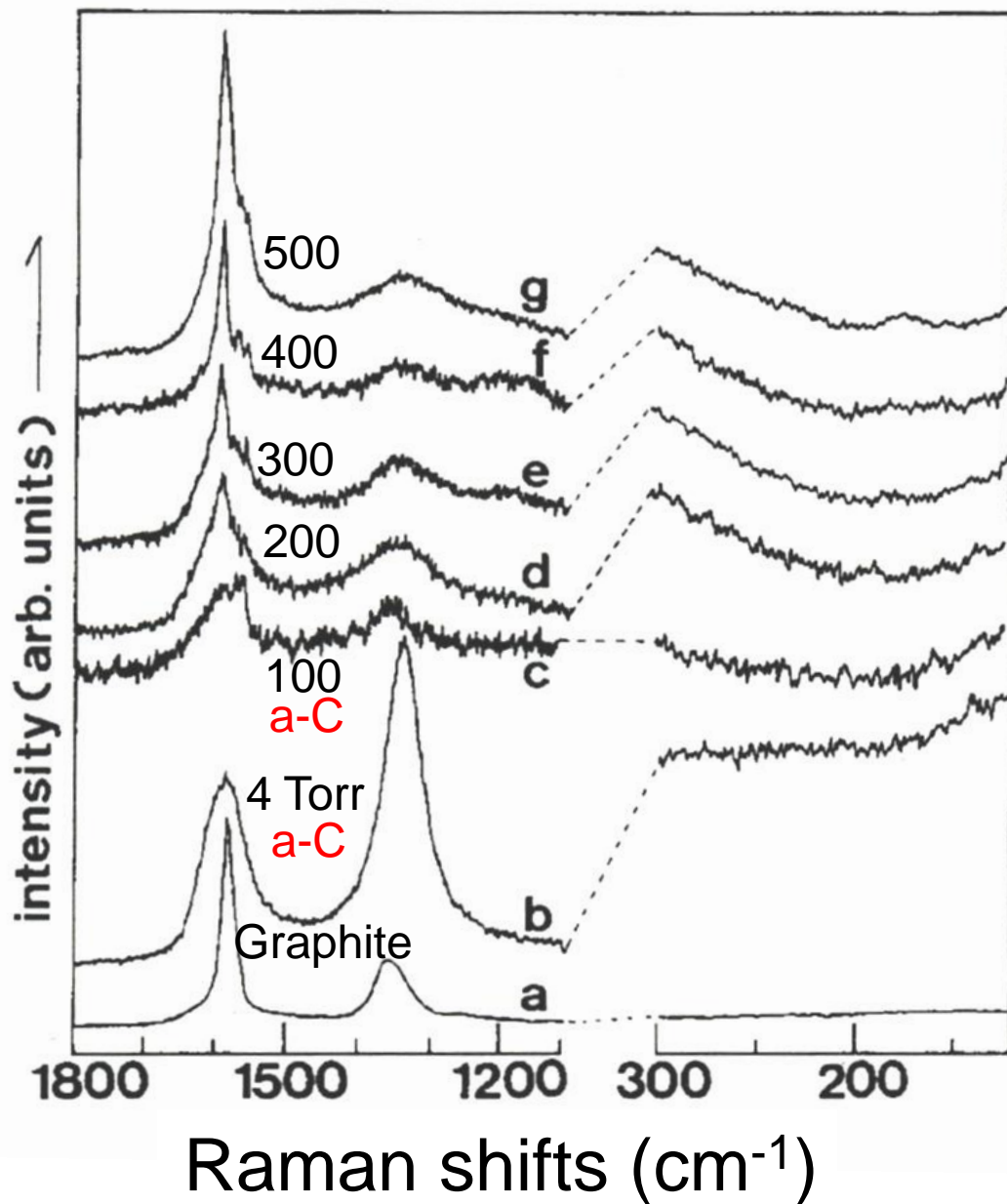
H.Dai et al.
Chem. Phys. Lett. 260(1996)471.

Raman spectra of Web-like deposits

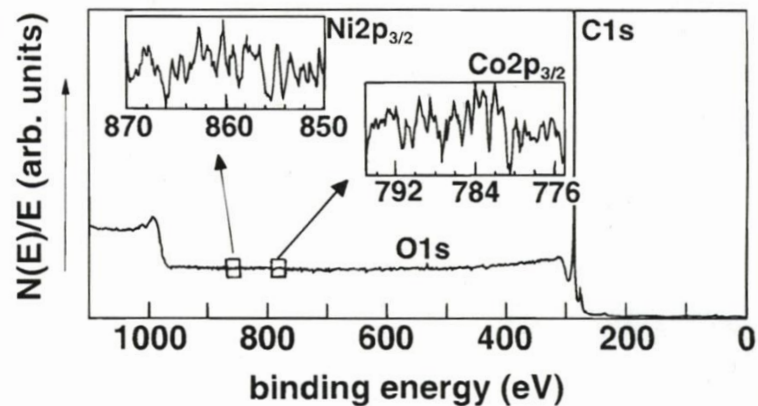


4 and 100 Torr
No SWNTs.
?

Deposits in front of targets

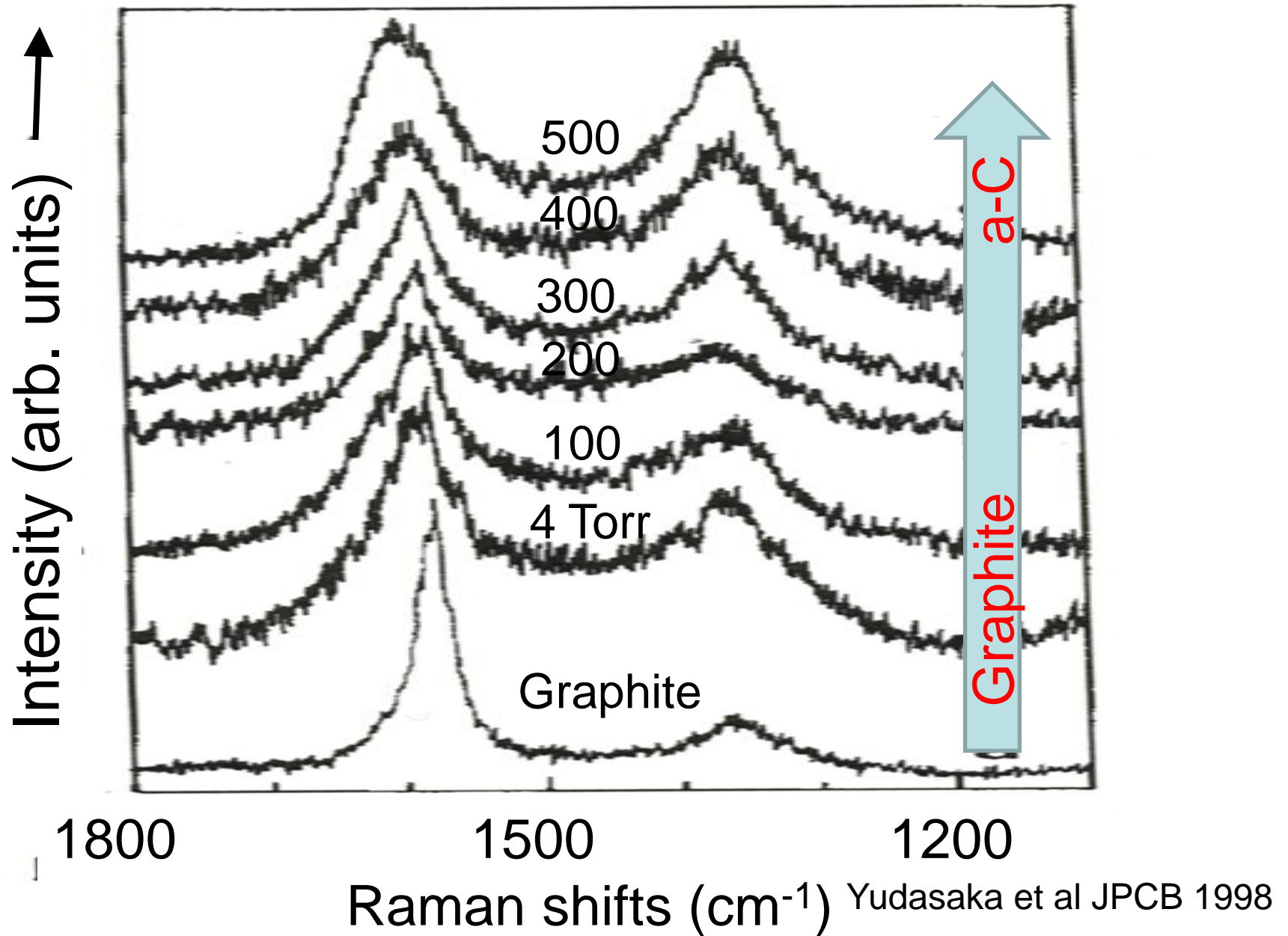


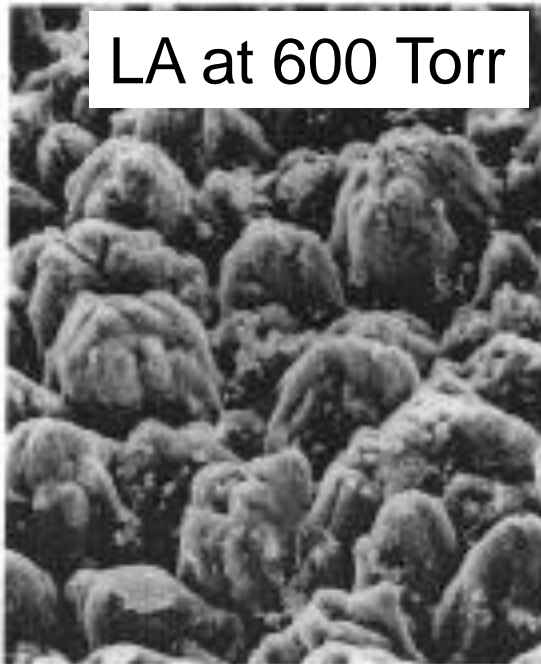
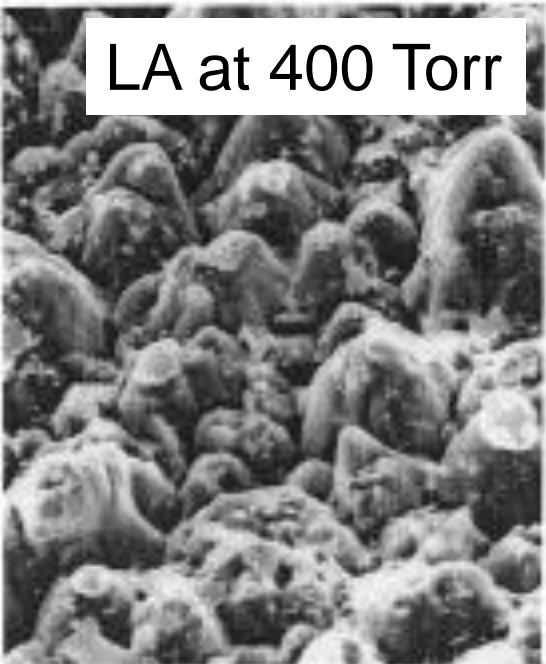
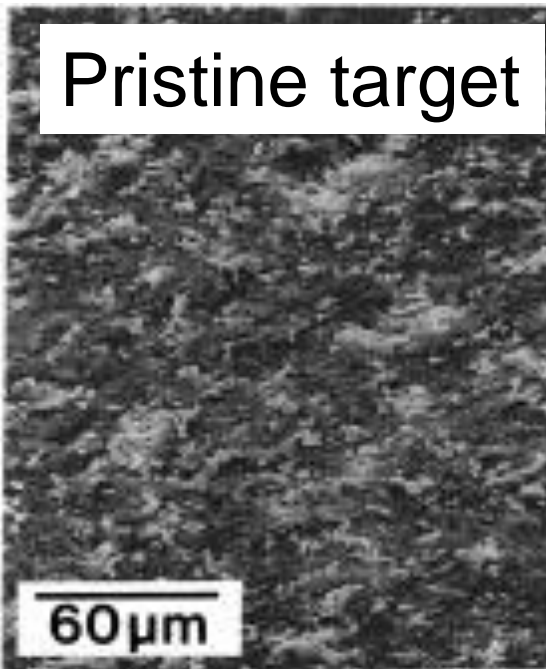
XPS: Ni, Co not exist on
Deposits at 4 Torr



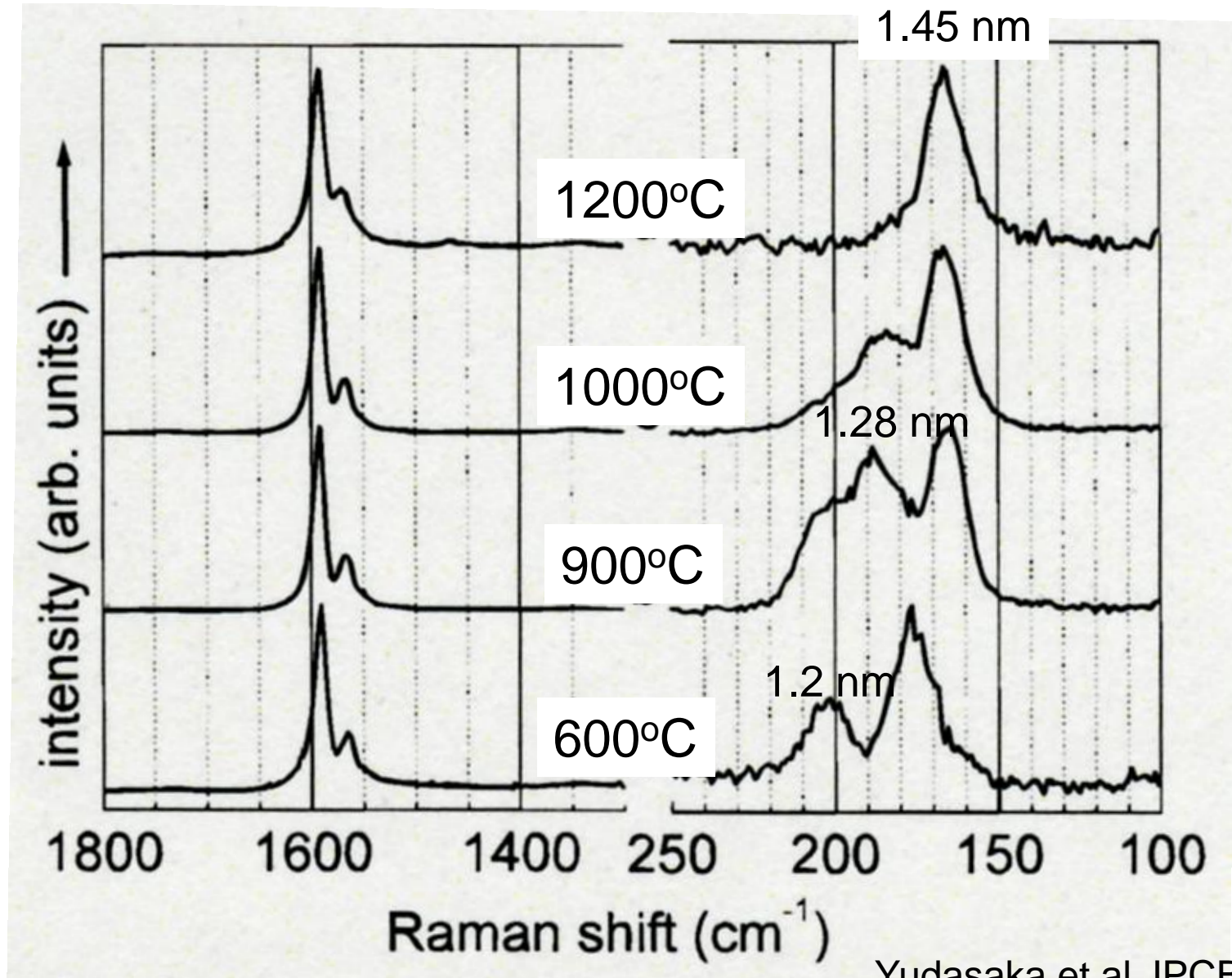
Yudasaka et al JPCB 1998

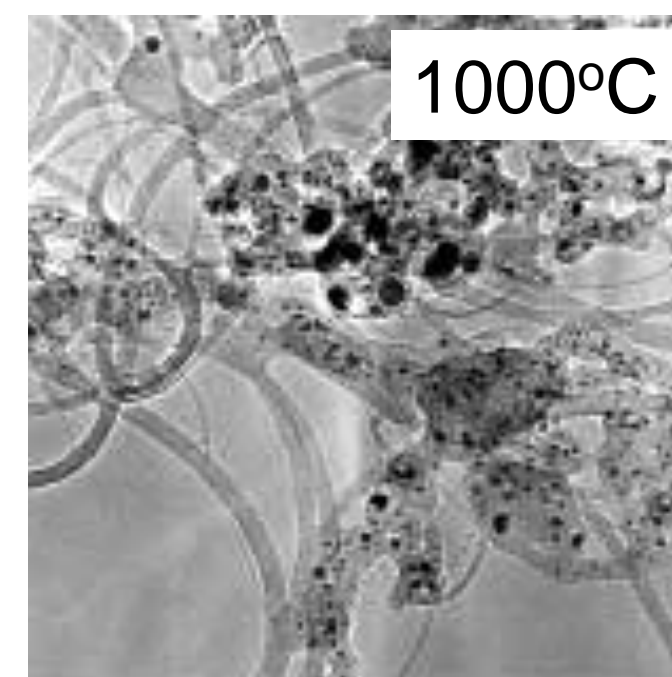
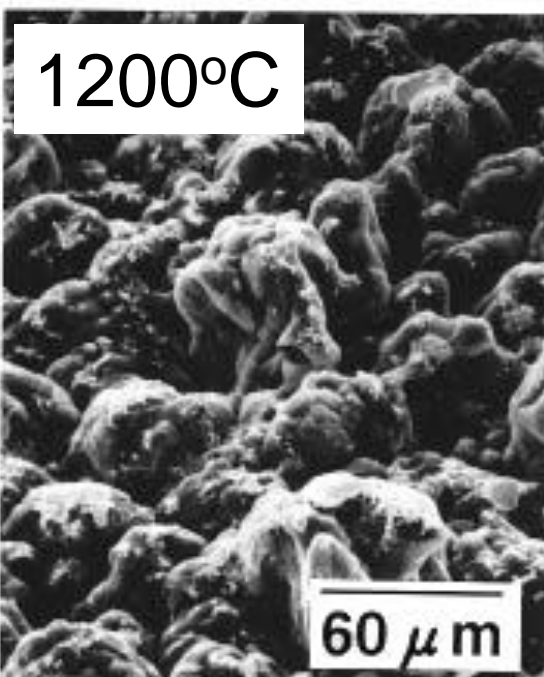
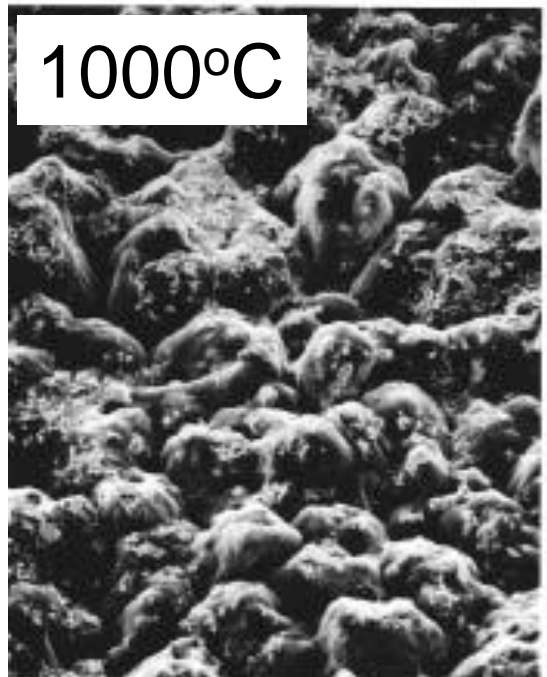
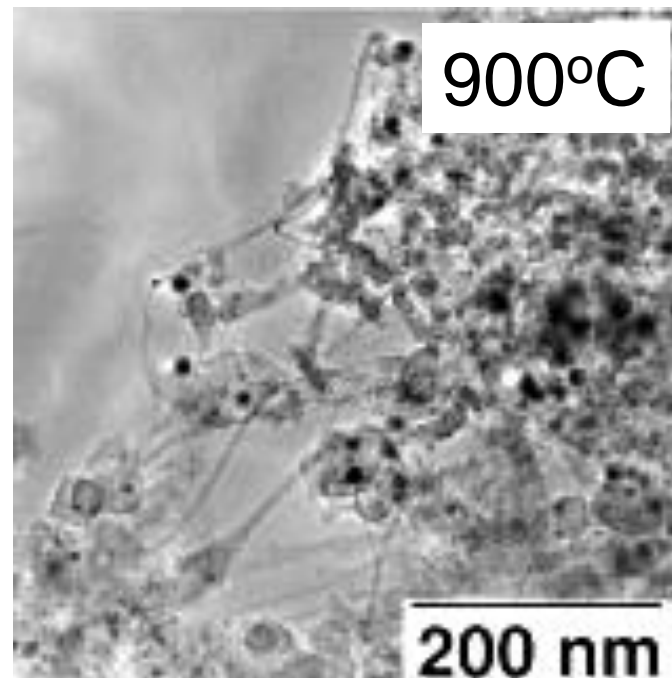
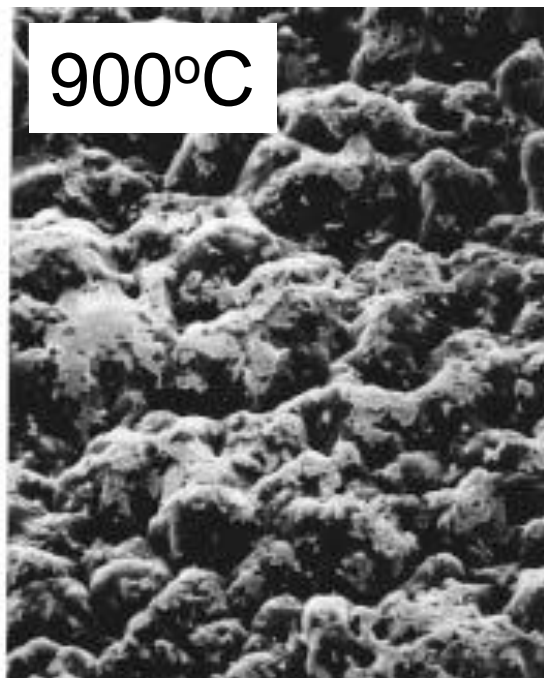
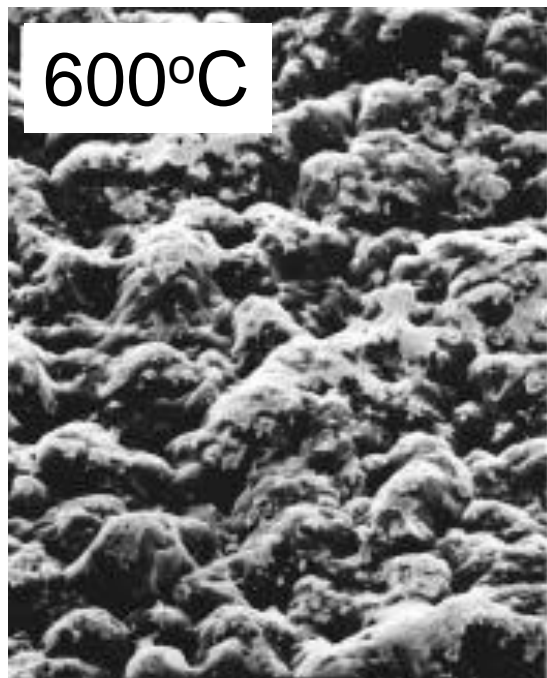
Raman spectra of graphite targets



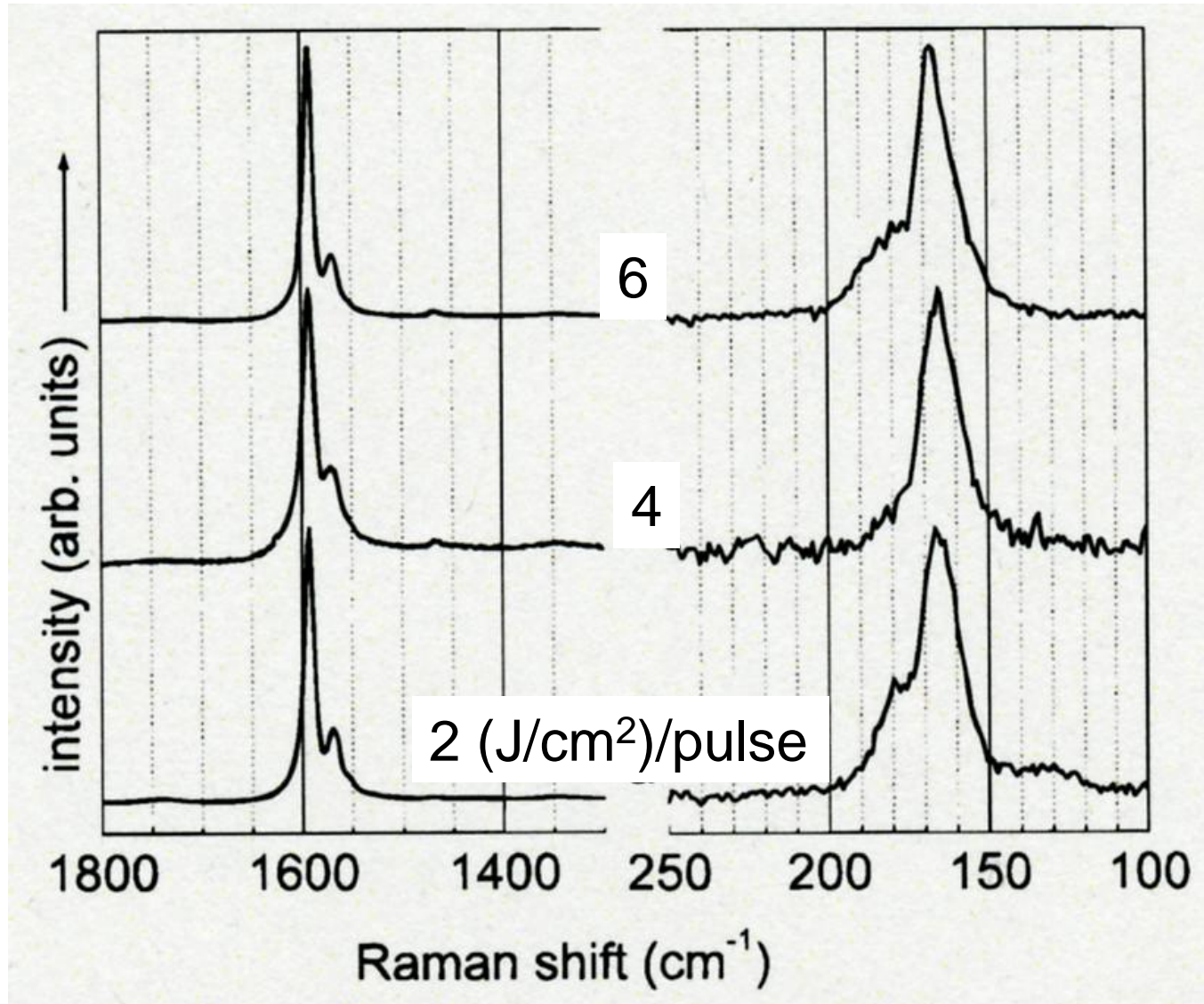


SWNT diameter decreased with ambient temperature.

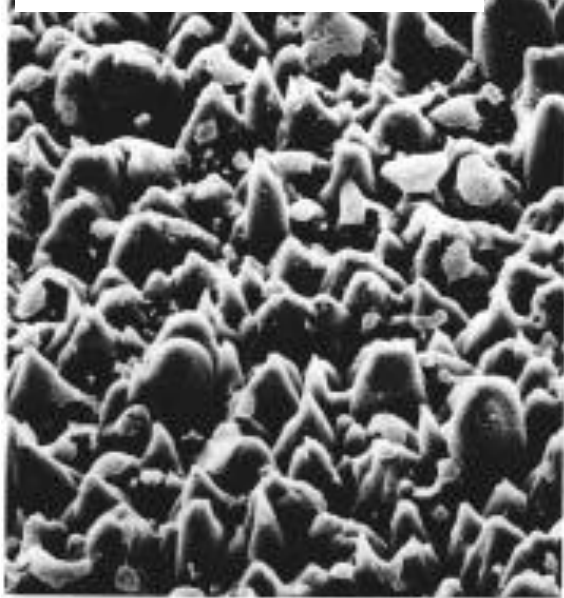




SWNT diameters did not depend
on laser power intensity



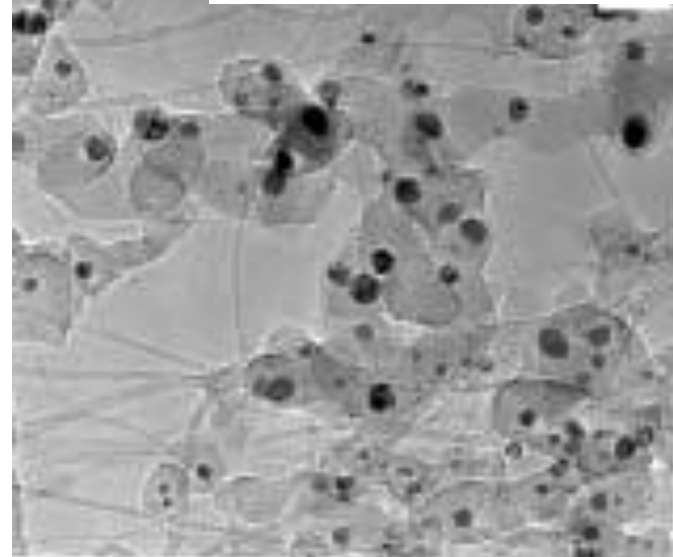
1 (J/cm²)/pulse



2 (J/cm²)/pulse



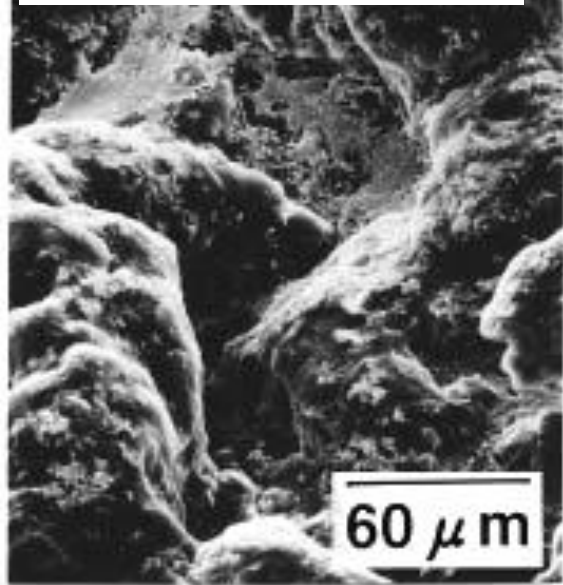
2 (J/cm²)/pulse



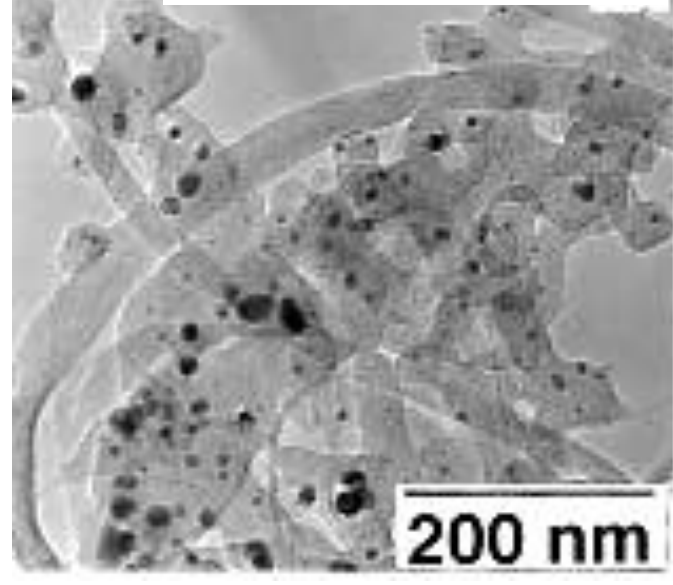
4 (J/cm²)/pulse



12 (J/cm²)/pulse



10 (J/cm²)/pulse

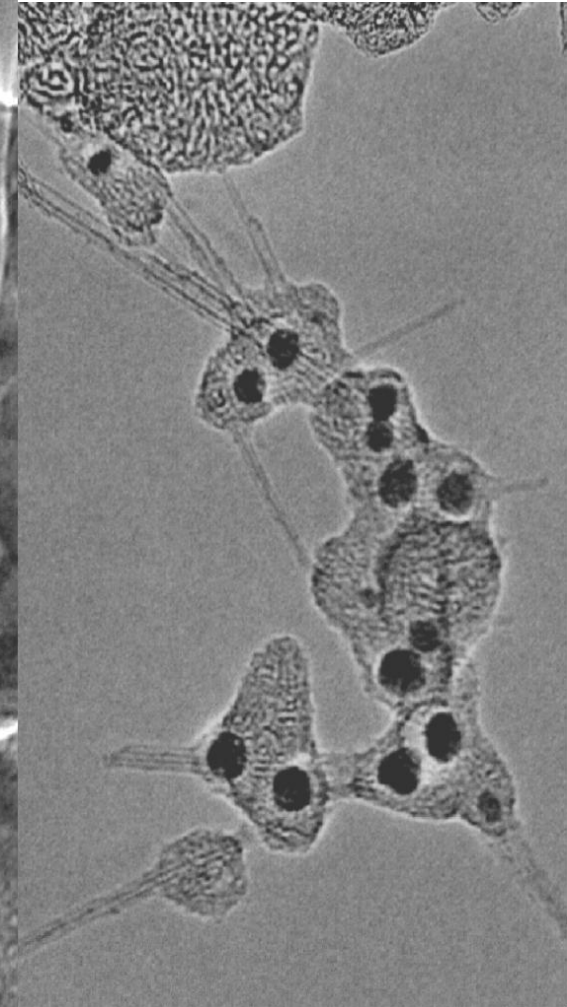
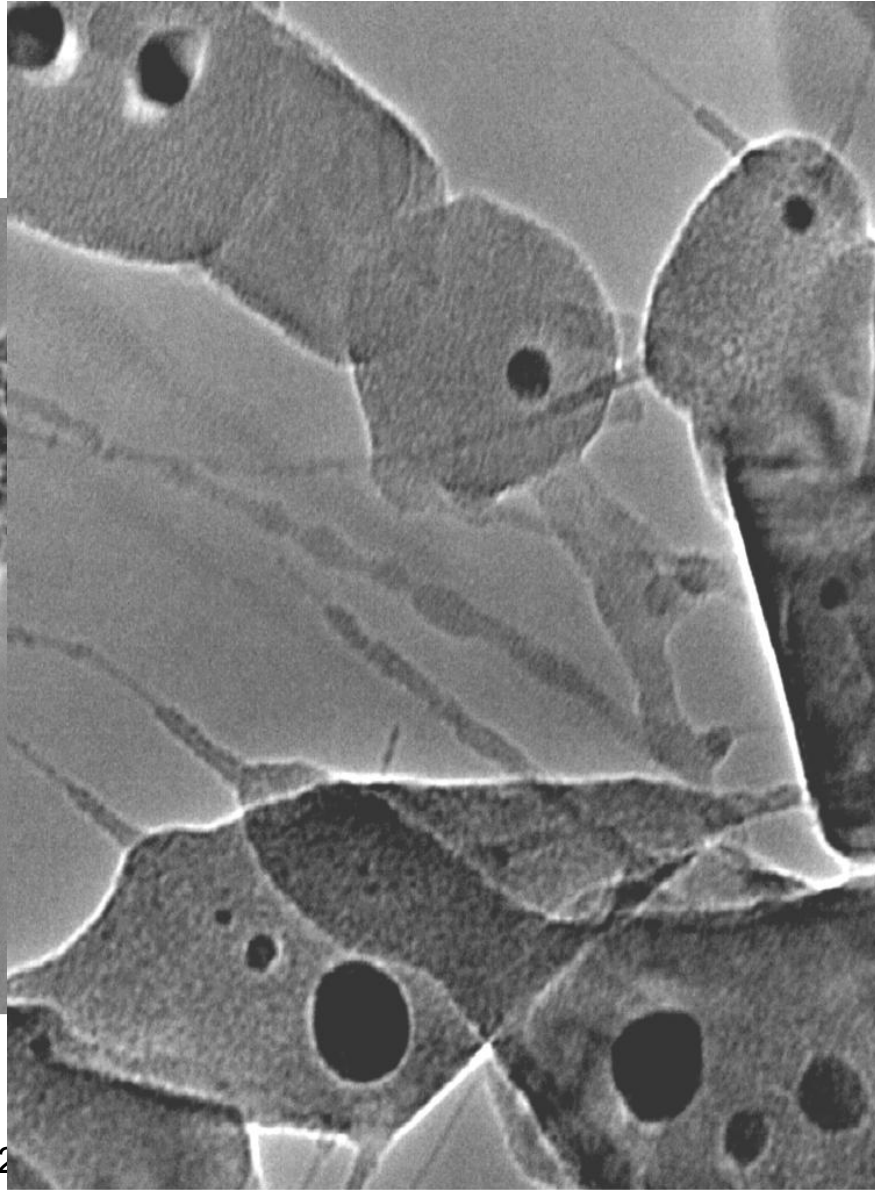
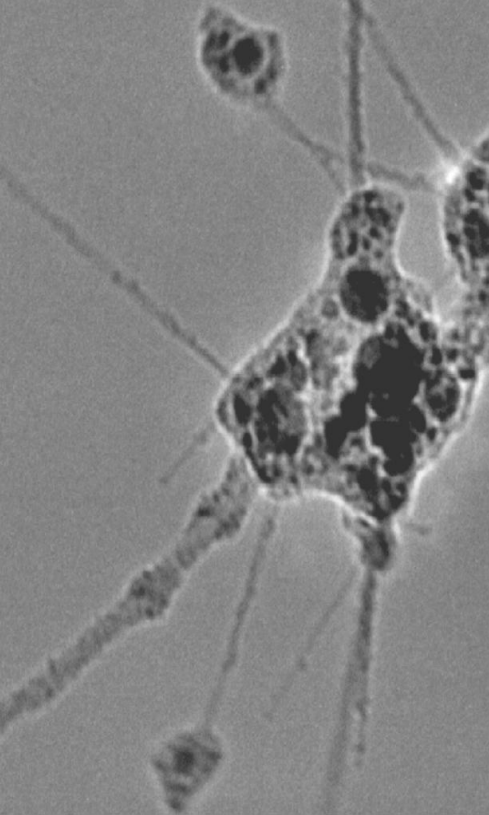


Nucleation and Growth of SWNTs
Individual SWNTs Start to Grow from a-C Having Metal Particles Inside

Nd:YAG laser ablation

CO₂ laser ablation

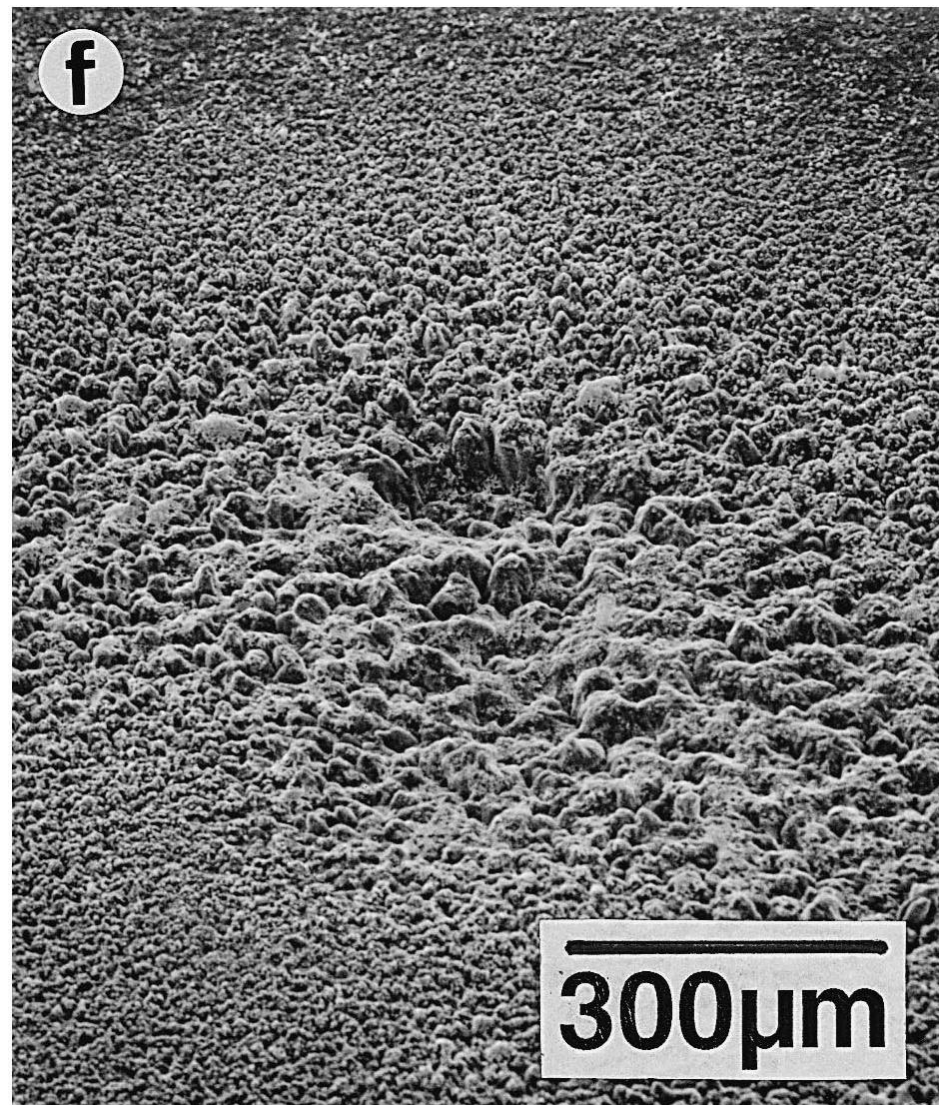
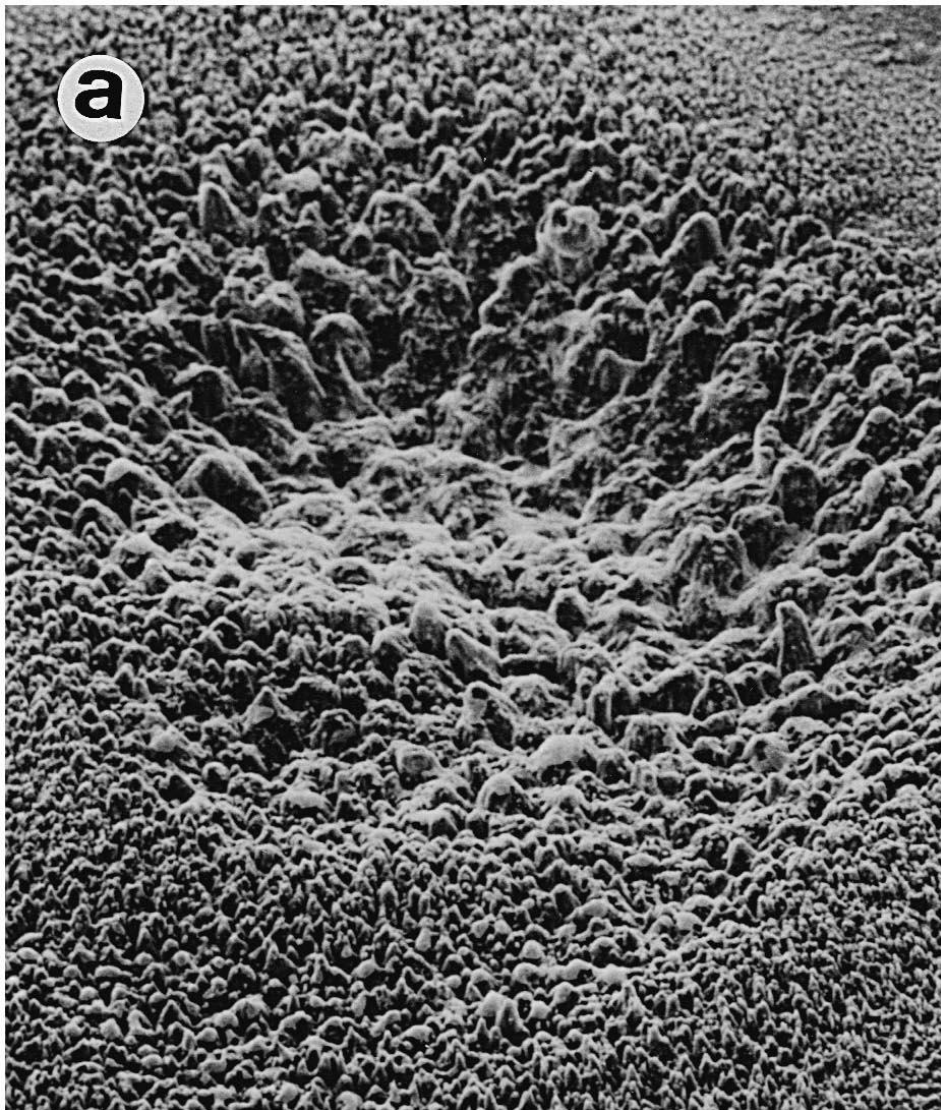
Arc discharge



Pulse-pulse intervals (Total 150 pulses)

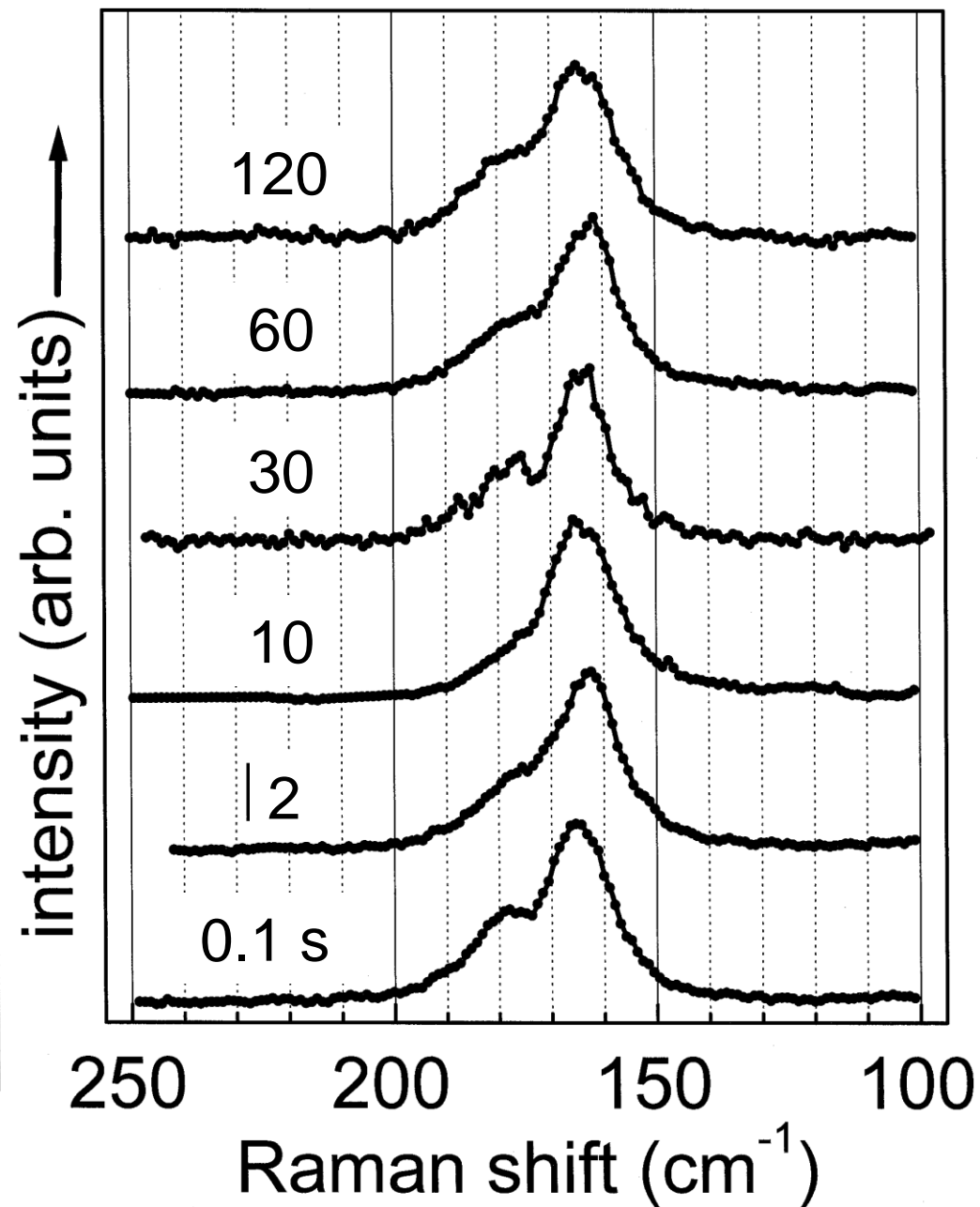
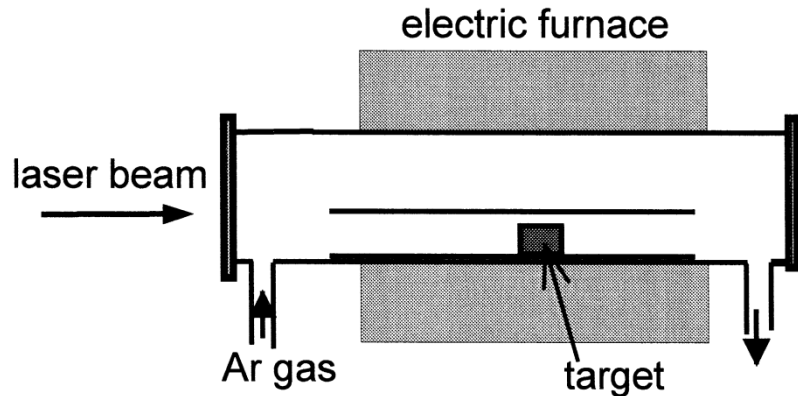
0.1 s

120 s

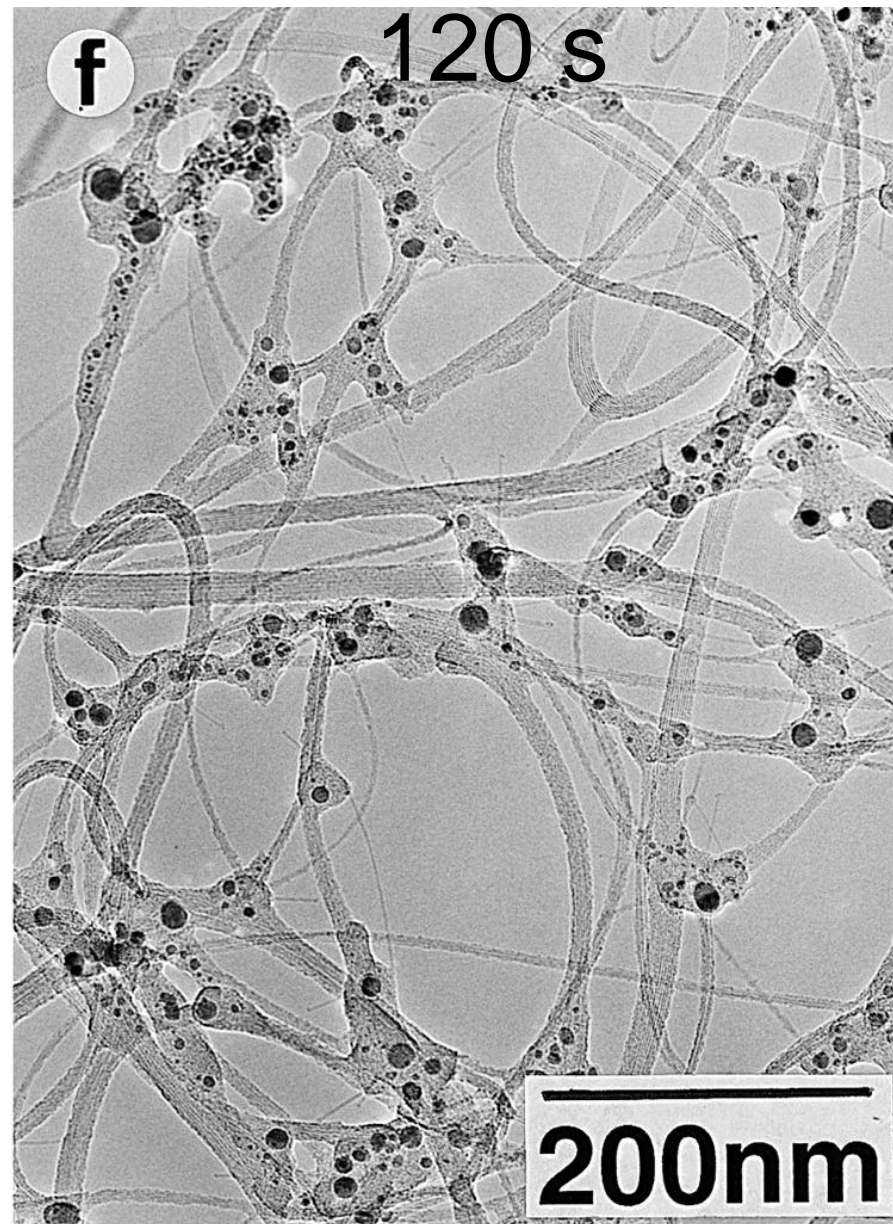
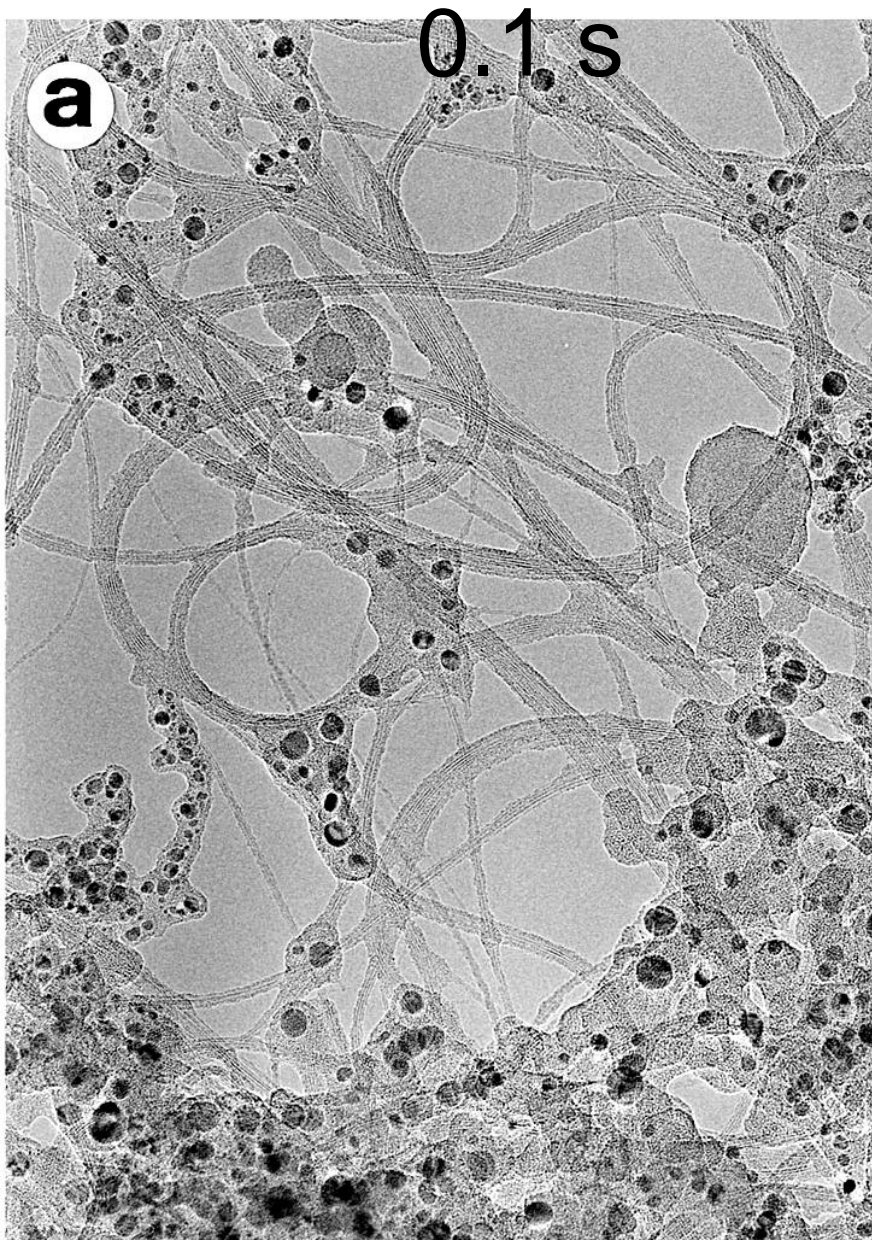


Nd:YAG laser
Wavelength 532 nm
Pulse width 6–7 ns
Frequency 10 Hz

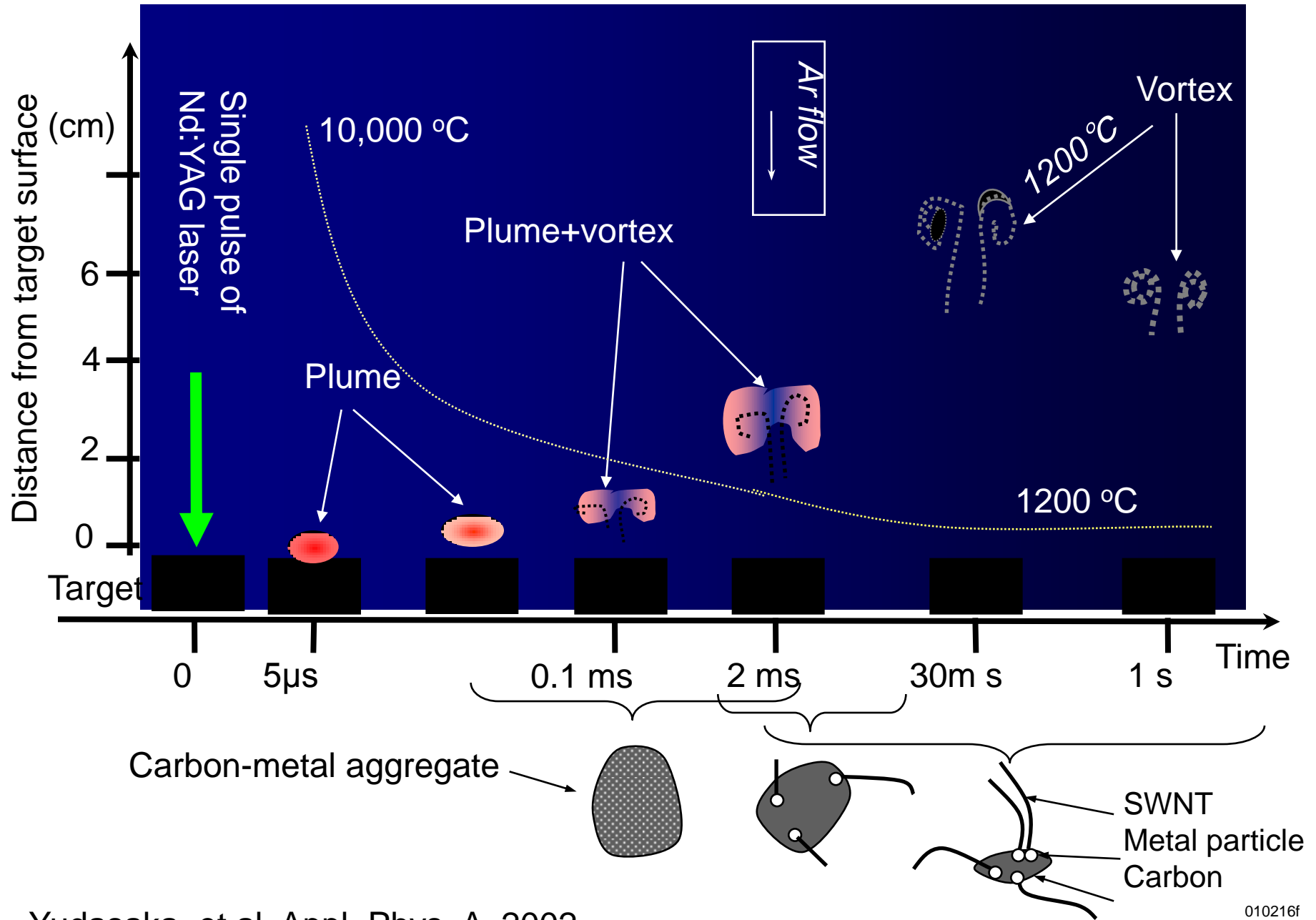
150 pulses
Pulse–pulse intervals
0.1, 2, 10, 30,
60 and 120 s.



Pulse-pulse intervals (Total 150 pulses)

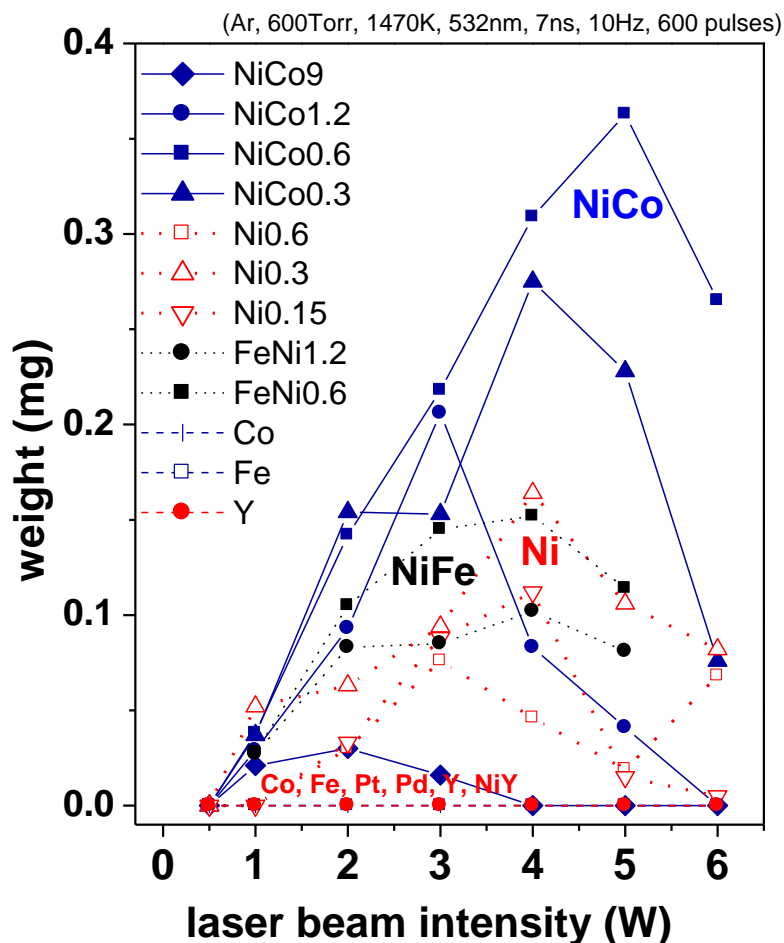


Nucleation and Growth of SWNTs by Single-Pulse of Nd:YAG Laser

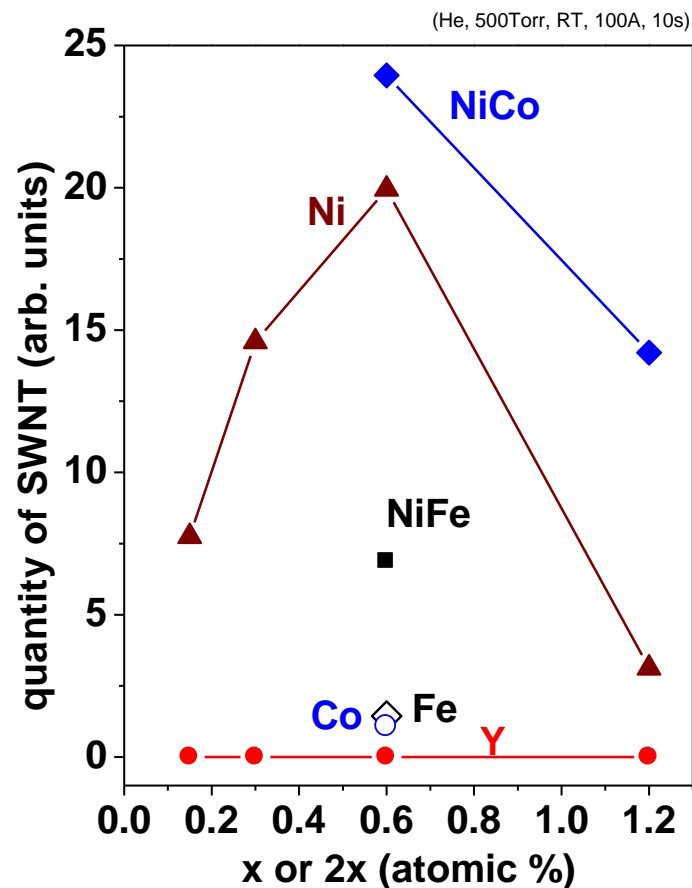
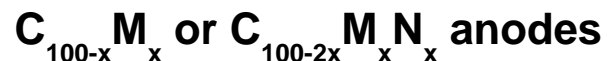


NiCo, Ni, and NiFe Are Effective Catalysts. Co, Fe, and Pt Are Poor Catalysts.

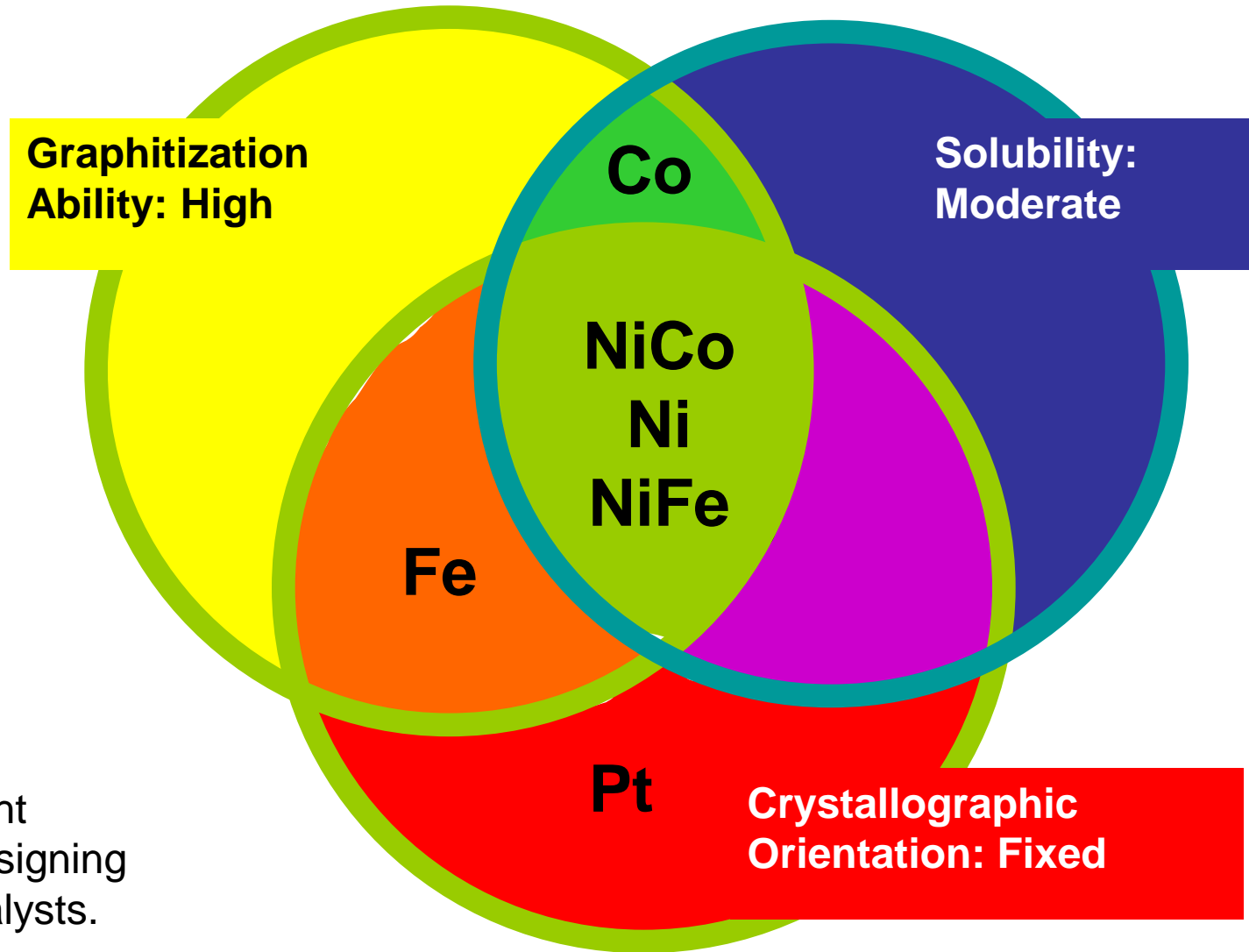
Quantities of web-like deposits formed by **Nd:YAG laser ablation**



Quantities of soot containing SWNTs formed by **DC arc discharge** using



**Yield of SWNTs Are Controlled
by Three Factors of Metal-Carbon interactions**



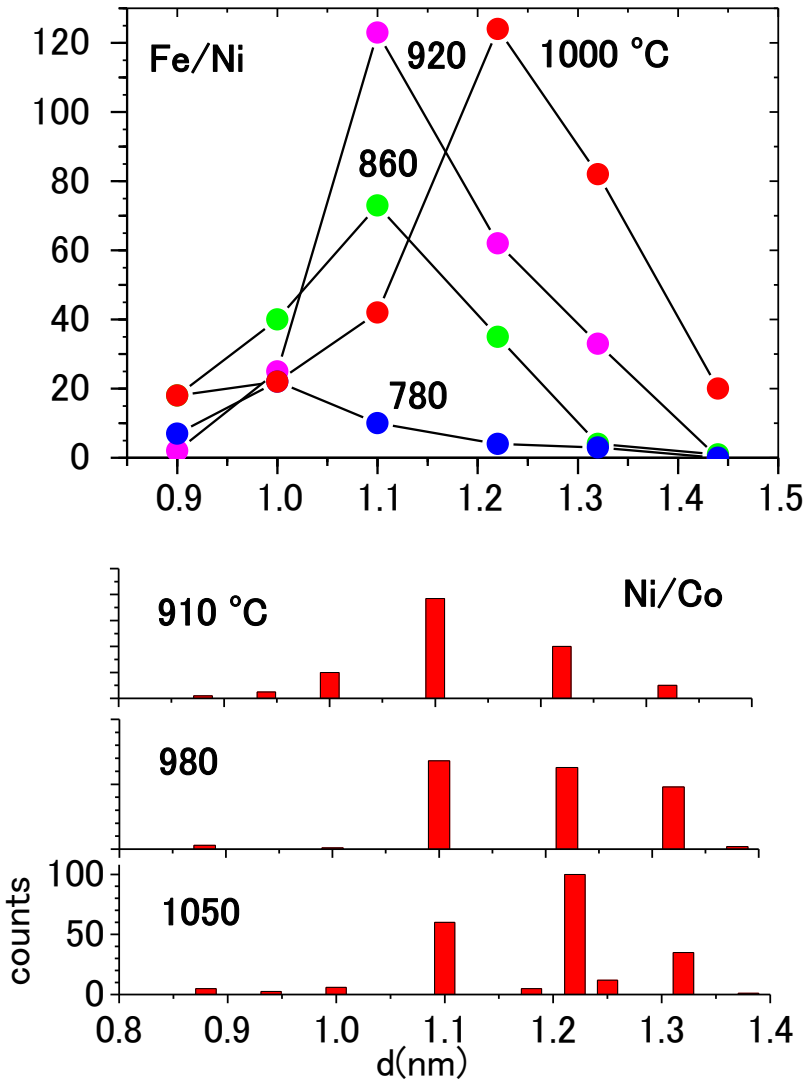
This study might be useful in designing new metal catalysts.

Diameter Control of SWNTs.

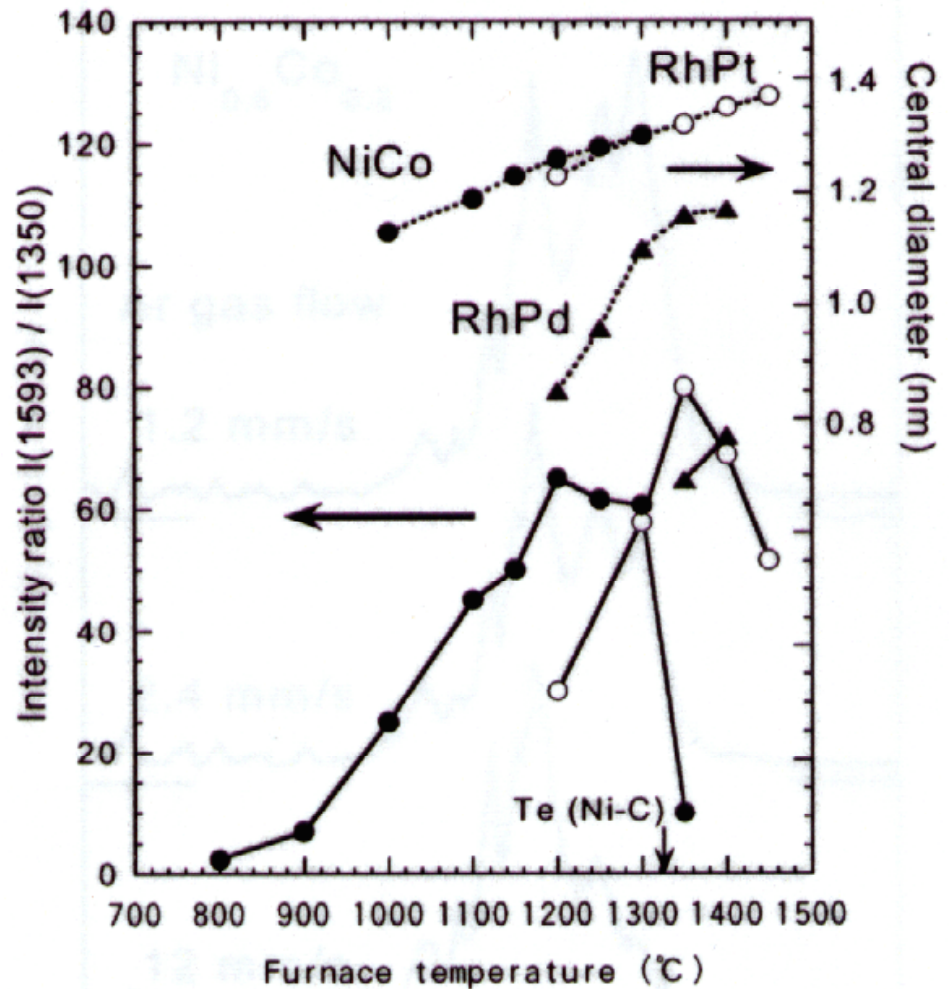
Possible by Choosing Formation Temperature or Metal Catalysts.

S. Bandow et al.

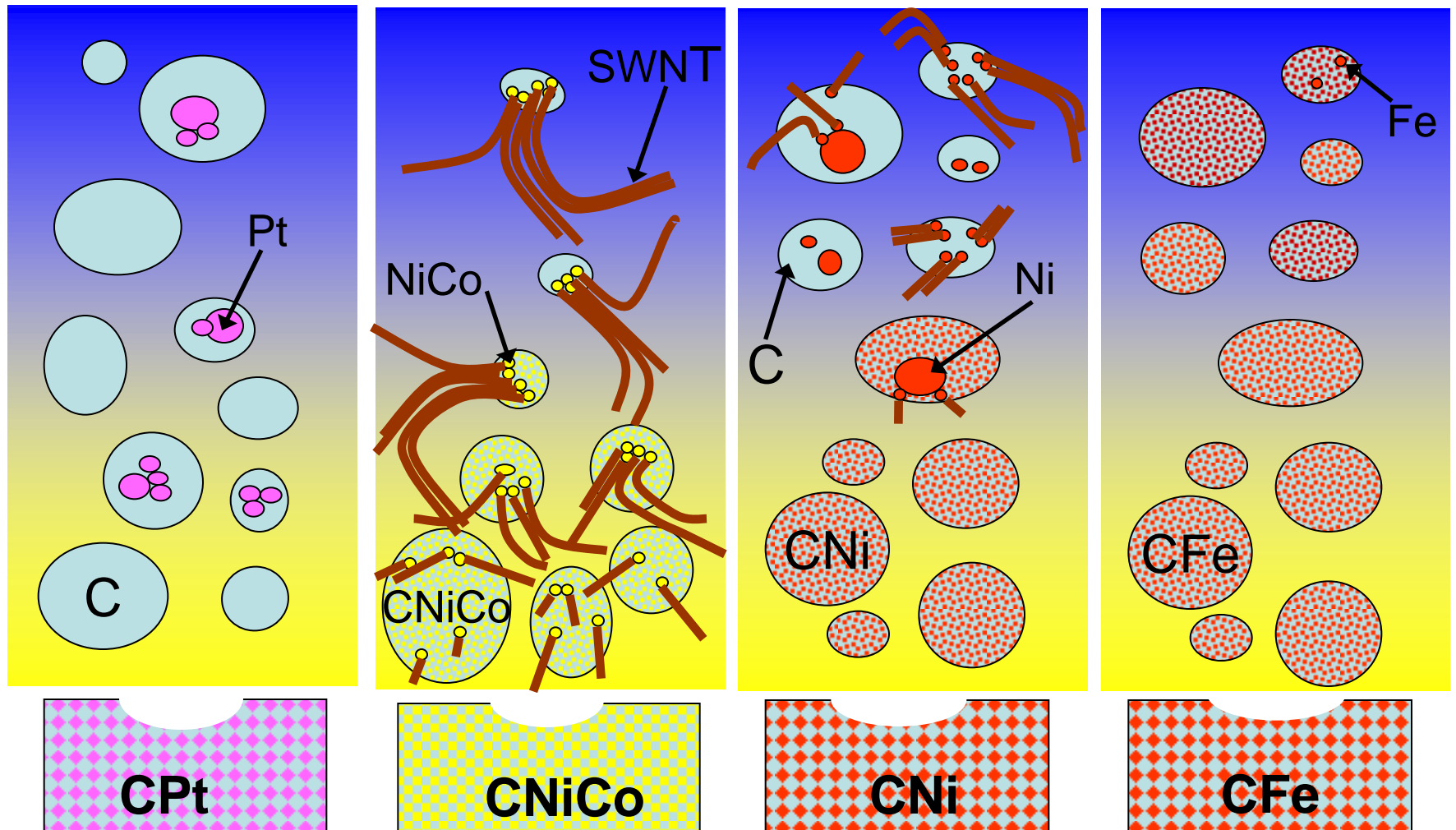
Phys. Rev. B 59 (1999) 2388.



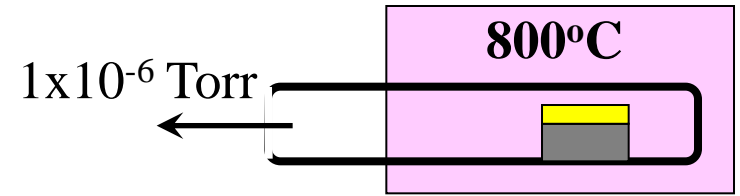
H. Kataura et al. Carbon 38(2000)1691.



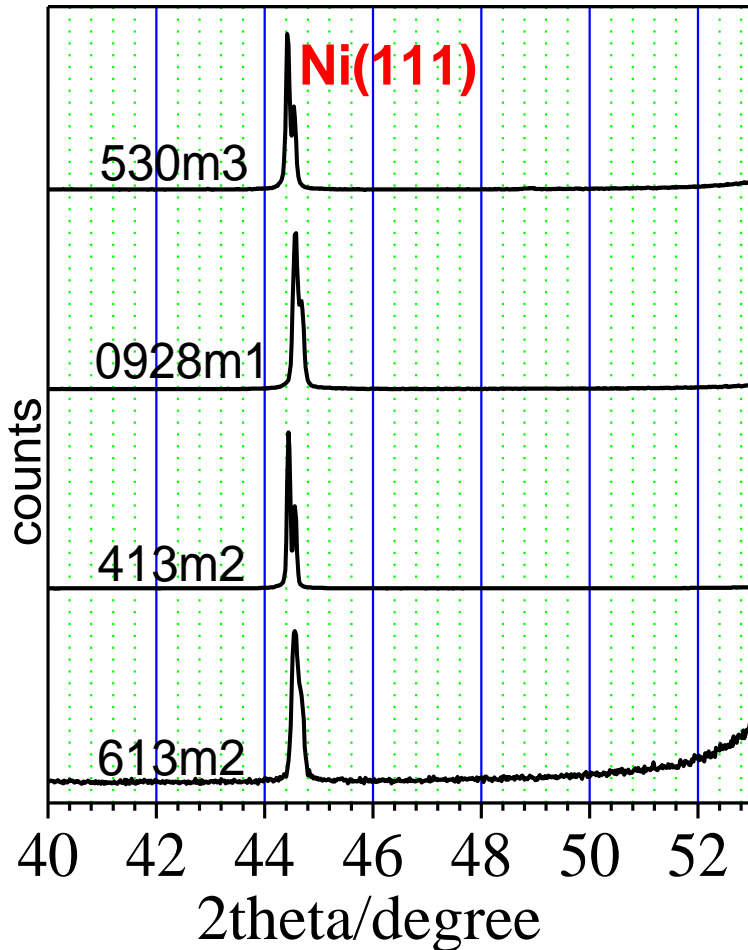
SWNT Growth Depending on Metal Catalysts



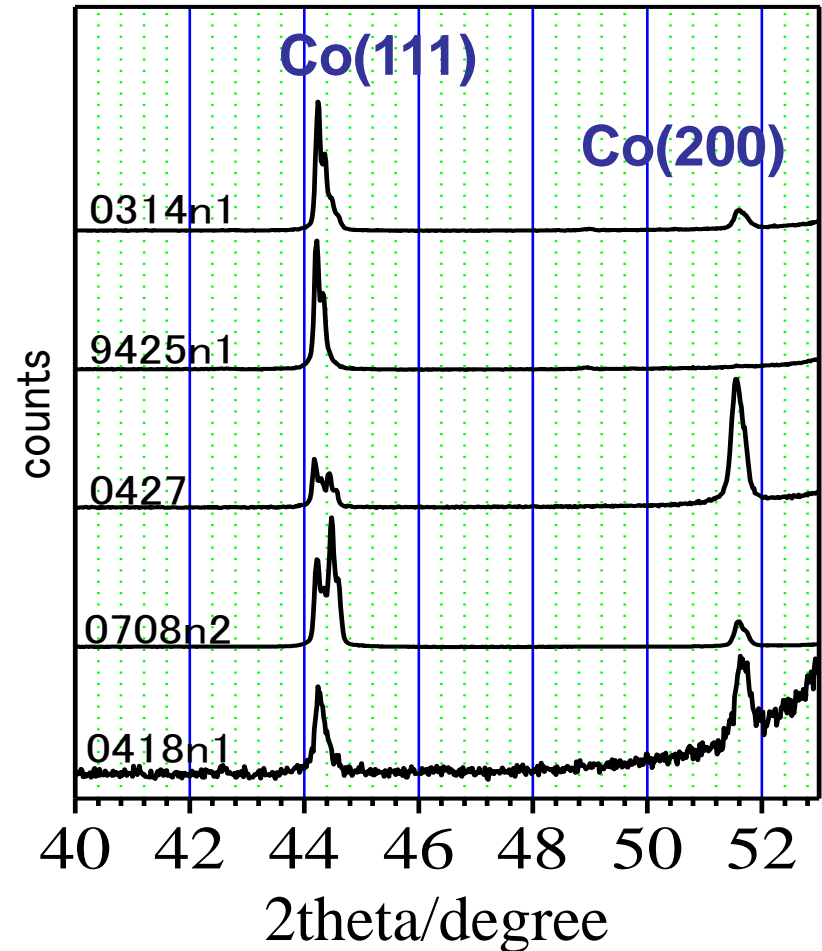
X-ray diffraction of Ni/Graphite and Co/Graphite



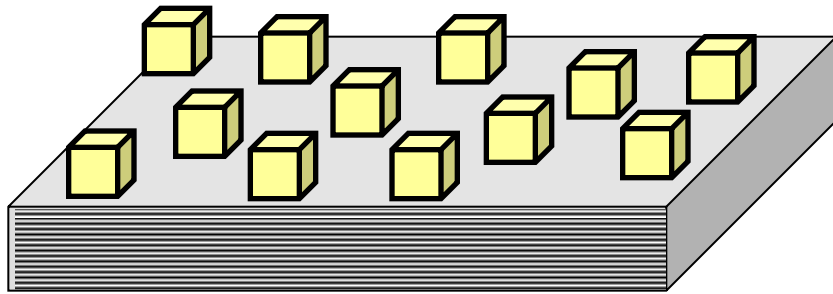
Heat treated at 800°C
REPRODUCIBLE



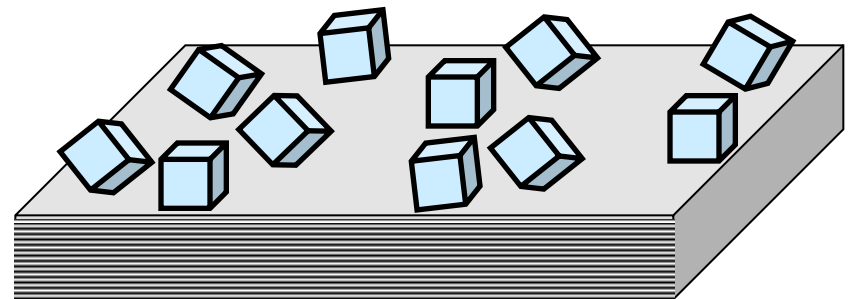
Heat treated at 800°C
NOT REPRODUCIBLE



Crystallographic Orientations of Ni and Co on Graphite Surfaces.

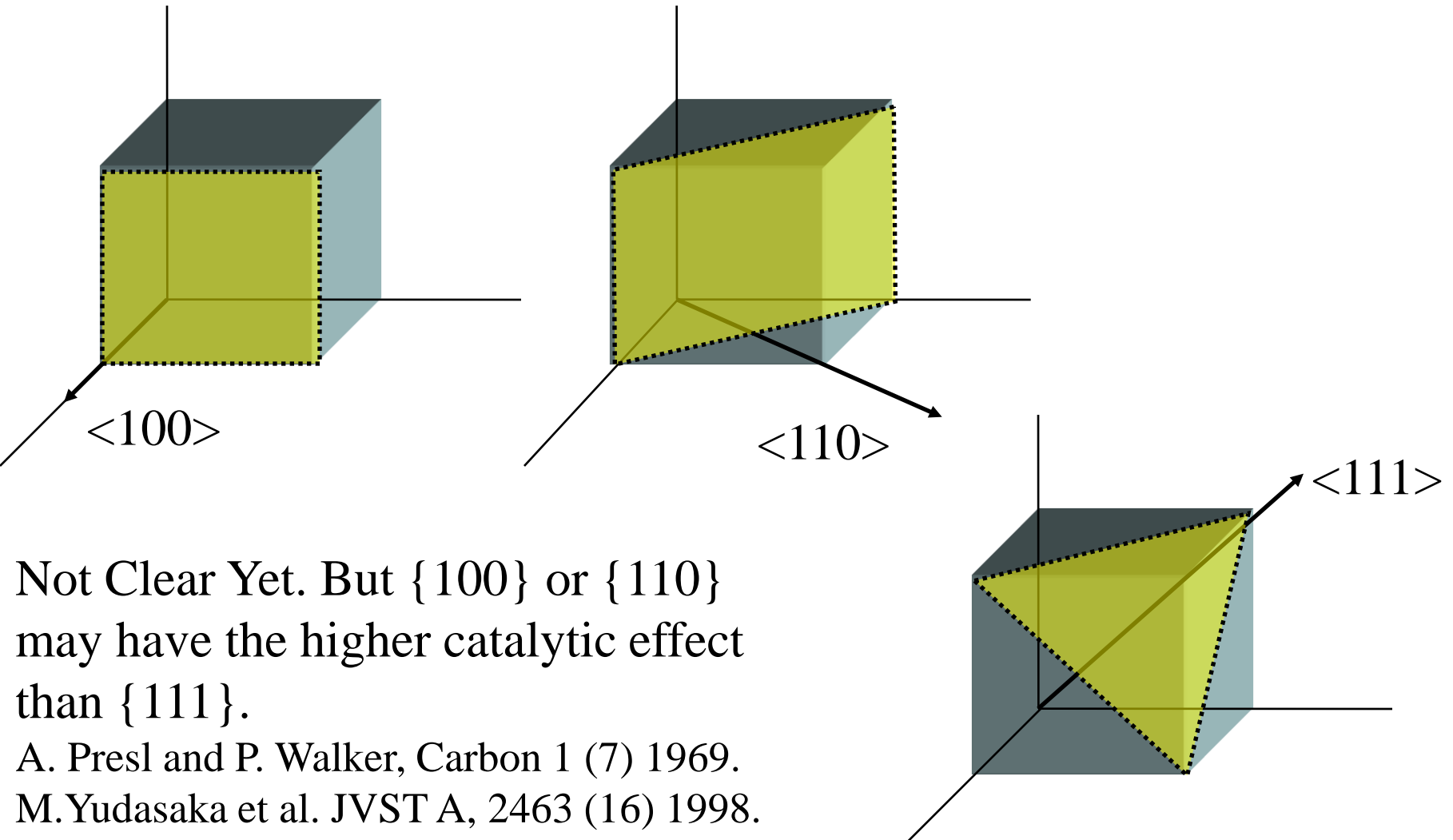


Ni/Graphite



Co/Graphite

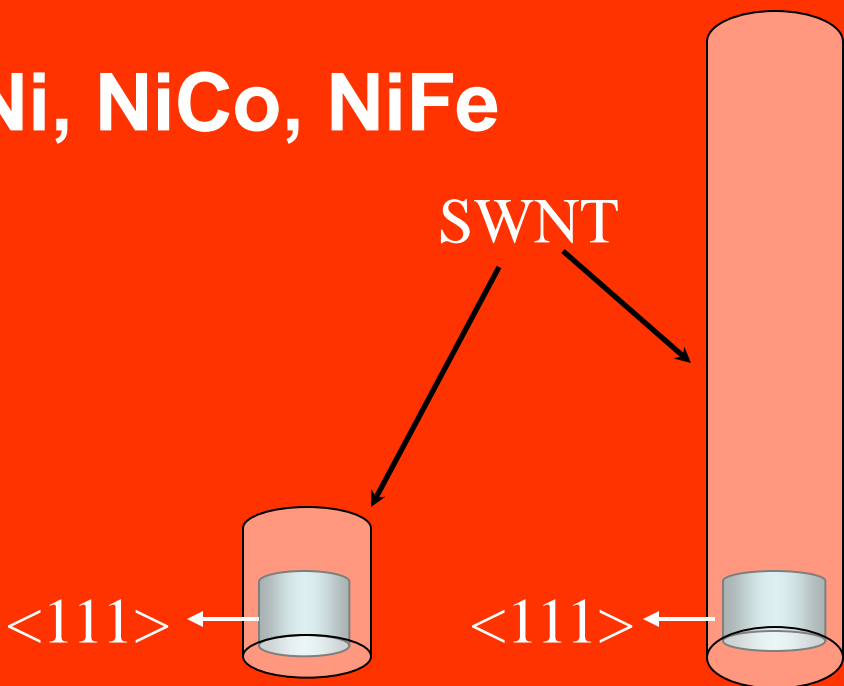
Graphitization Ability Depends on Crystallographic Faces of Metals



Not Clear Yet. But $\{100\}$ or $\{110\}$ may have the higher catalytic effect than $\{111\}$.

A. Presl and P. Walker, Carbon 1 (7) 1969.
M. Yudasaka et al. JVST A, 2463 (16) 1998.

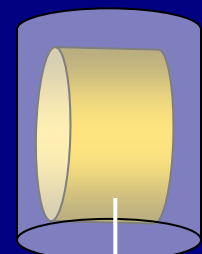
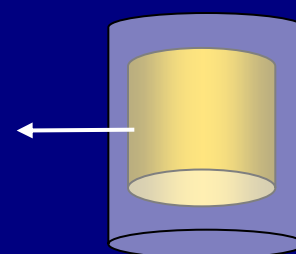
Ni, NiCo, NiFe



**Orientation Instability of Co
on Graphite Hinder
SWNT Growth**

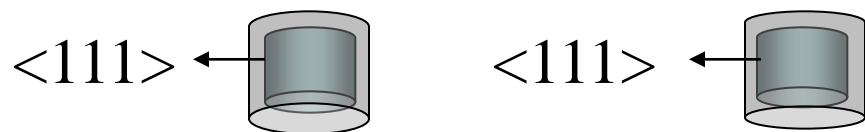
Co

$\langle 111 \rangle$



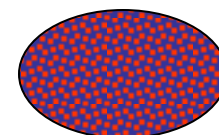
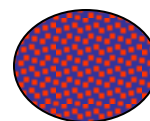
$\langle 111 \rangle$

Pt



Poor graphitization-catalyst

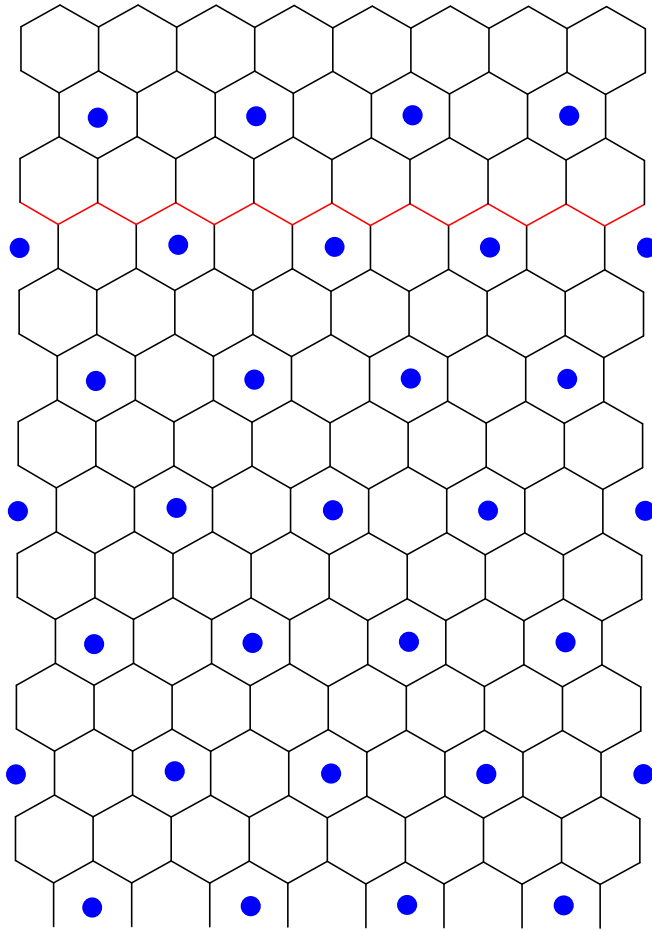
Fe



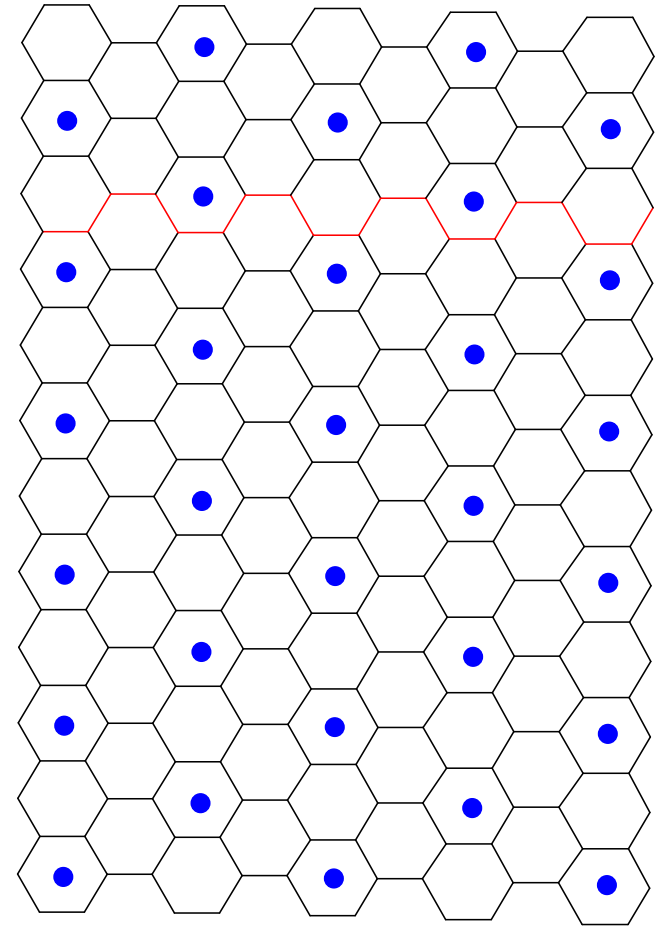
High solubility in carbon

SWNT Chirality Controlled by Lattice Matching of Graphite and Ni (111)

ZigZag



Arm Chair



● **Ni (111)**

SWNT Formation by CVD

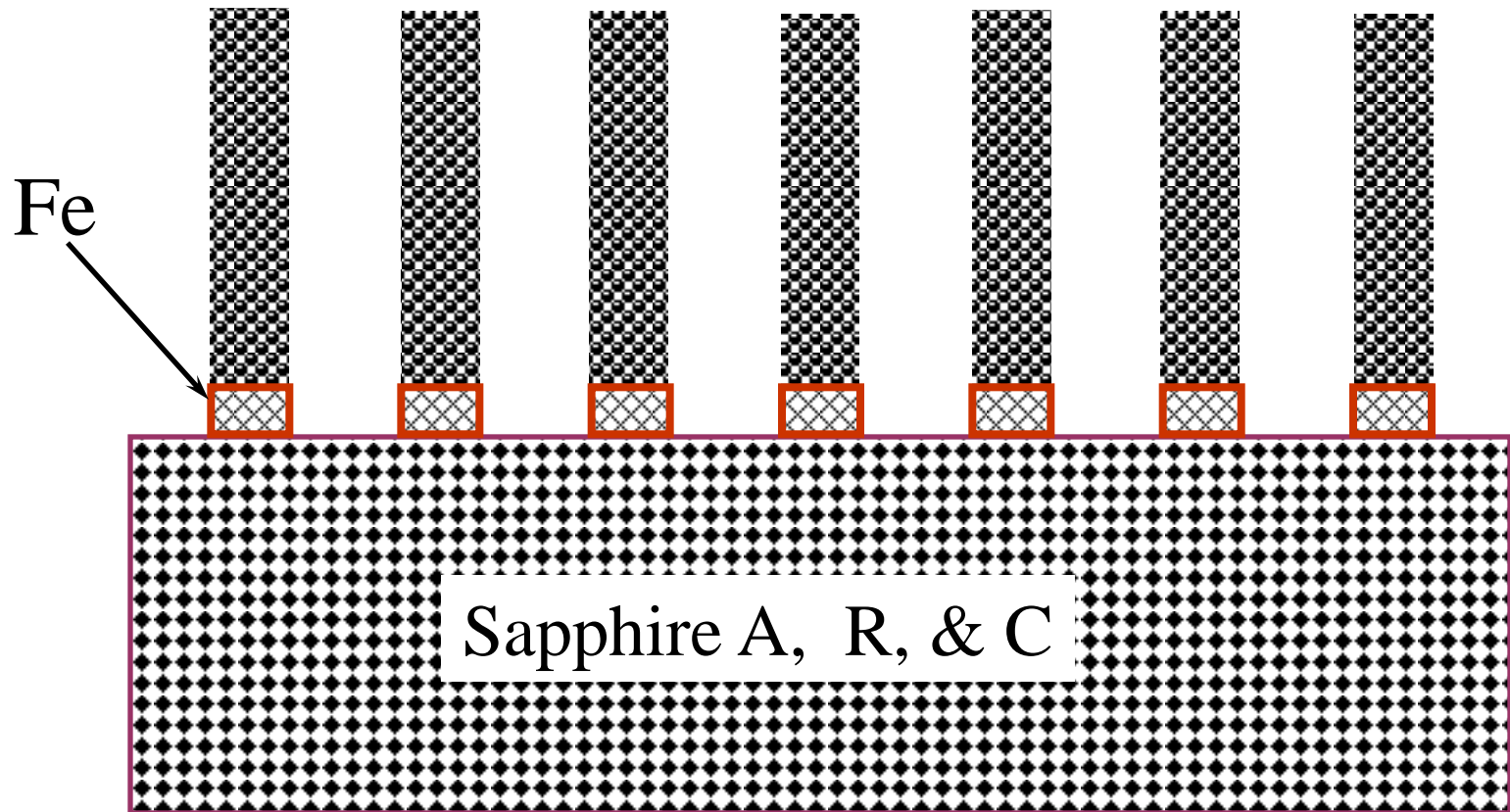
Catalysts: **Fe, FeCo, Co, Mo**

Controlling these metal particle sizes at 1- 2 nm

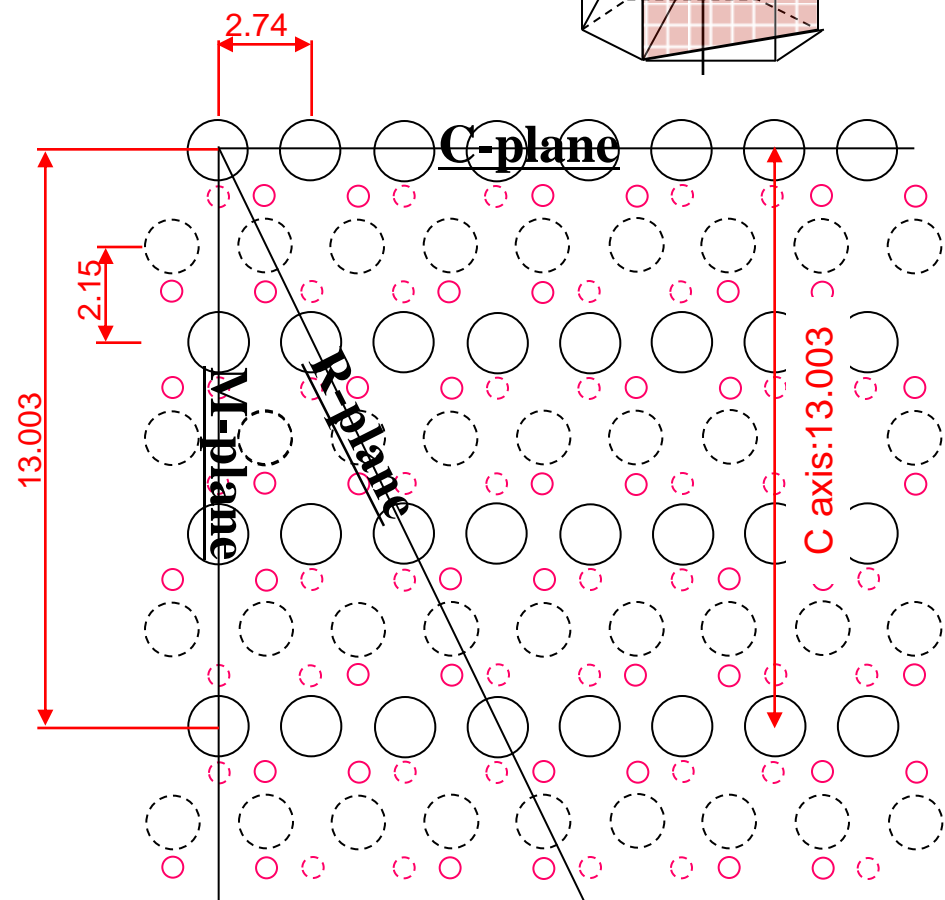
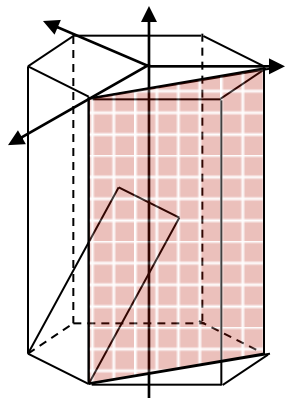
References of SWNT growth by CVD

1. H. Dai A.G. Rinzler, P. Nikolaev, A. Thess, d.T. Colbert, R.E. Smalley Chem. Phys. Lett., 260(1996)471. **NiCo, Mo**
2. Fonseca, A, K. Hernadi, P. Piedigrosso, L.P. Biro, S. Lazarescu, Ph. Lambin,P.A. Thiry, D. Bernaerts, J.B. Nagy ., *Proc. of Fullerenes : Chemistry of Physics and New Directions IX*; The Electrochemistry Society Inc.: Montreal, 1997.
3. Fonseca A., K.Hernadi, P. Piedigrosso, J.-F. Colomer, K. Mukhopadhyay, R. Doome, S. Lazarescu, L.P. Biro, Ph. Lambin,P.A. Thiry, D. Bernaerts, J.B. Nagy, *Appl.Phys. A*, 67(1998)11. (silica, zeolite) **Co > Cu, Fe**
4. J. Kong, H.T Soh, A.M. Casell, C.F. Quate, H. Dai, *Nature* 395(1998)878. **Fe, Mo (Al)**
5. J. Kong, A.M. Casell, H. Dai, *Chem. Phys. Lett.* 292(1998)567. **Fe₂O₃/Al₂O₃**
6. Z.K. Tang, H.D. Sun, J. Wang, J. Chen, G. Li, *Appl. Phys. Lett.*, 73(1998)2287. (microporous aluminophosphate AlPo4-5 diameter=1.01 nm, tripropylamine, 500-800°C)
7. H.M. Chen, F. Li, G. Su, H.Y. Pan, L. L. He, X. Sun, M.S. Dresselhaus, *Appl. Phys. Lett.*, 72(1998)3282. **Fe (S, H)**
8. H.M. Chen, F. Li, X. Sun, S.D.M. Brown, M.A. Pimenta, A. Marucci, G. Dresselhaus, M.S. Dresselhaus, *Chem. Phys. Lett.* 289(1998)602. **Fe (S, H) 1100-1200°C**
9. J.H. Hafner, M.J. Bronikowski, B.R.azamoian, P. Nikolaev, A.G.Rinzler, D.T. Colbert, K.A.Smith, R.E.Smalley *Chem. Phys. Lett.* 296(1998)195. **FeMo(9:1)**
10. E. Flahout, A. Govindaraj, A. Peigney, Ch. Laurent, A. Rousset, C. N. R. Rao, *Chem. Phys. Lett.*, 3000(1999)236. **FeCo nanoparticles**
11. A. M. Cassell, J. A. Raymakers, J, Kong, H. J. Dai, *J. Phys. Chem. B* 103(1999)483.
12. P. Nikolaev, M. J. Bronikowski, R. K. Bradley, F. Rohmund, D. T. Colbert, K. A. Smith, R.E. Smalley *Chem. Phys. Lett.*, 313(1999)91. **Fe**
13. J.-F. Colomer, C. Stephan, S. Lefrant, G. V. Tendeloo, U. Eiliams, Z. Konya, A. Fonseca, Ch. Laurent, J.B. Nagy *Chem Phys. Lett.*, 317(2000) 83. **CoFe, Fe, Co (MgO)**
14. I. Wilems, Z. Konya, J. -F. Colomer, V. Tendeloo, N. Nagaraju, A. Fonseca, J. B. Nagy, *Chem.Phys. Lett.*, 312(2000)71. **CoMo, CoV**
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One Idea of Controlling the Chiralities of SWNTs.

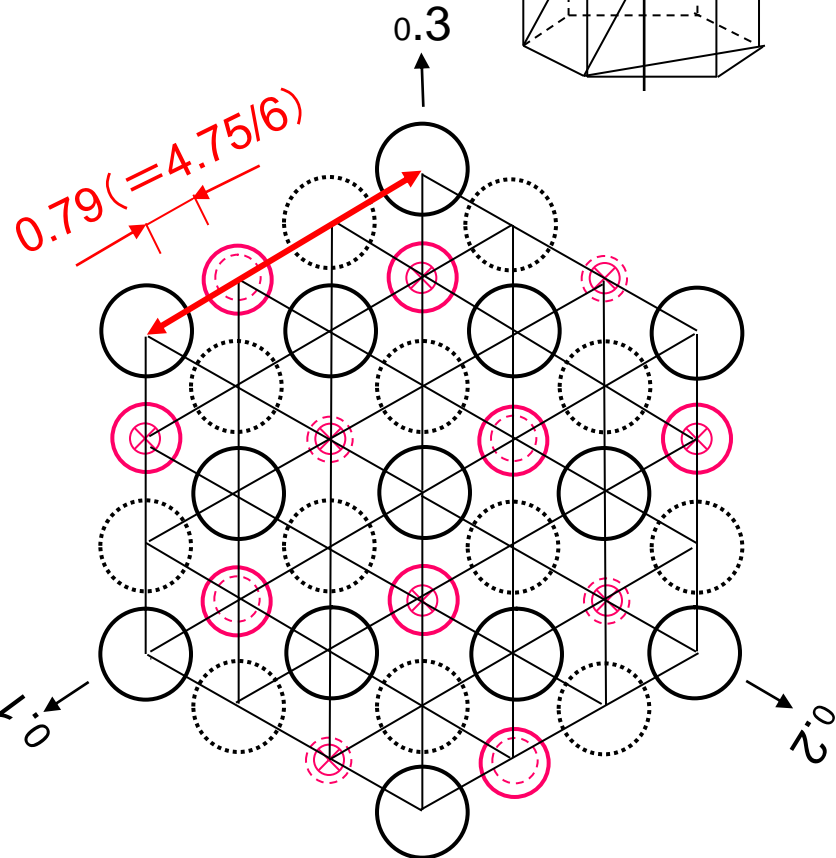
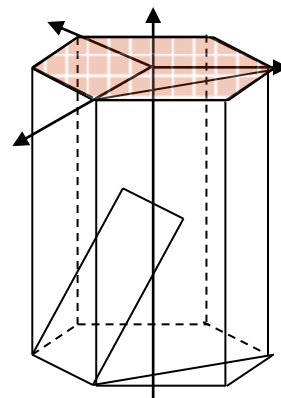


A face ($11\bar{2}0$)



○ L0(AI) ○ L1(O) ○ L2(O) ○ L3(AI)

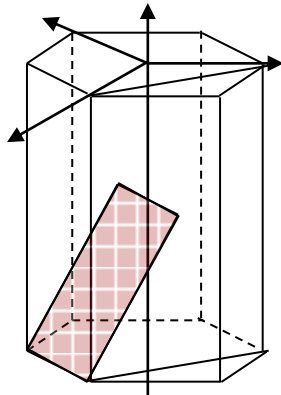
C face (0001)



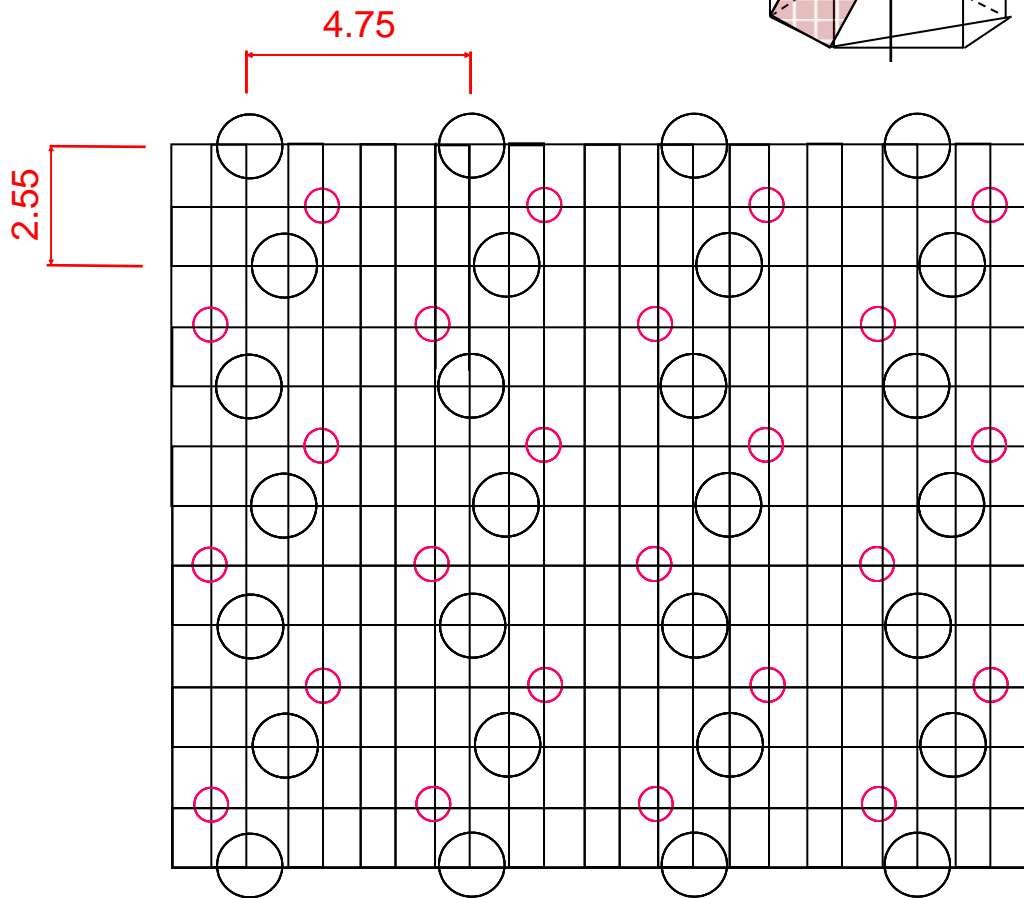
Tetragonal

Hexagonal

R face $(110\bar{2})$



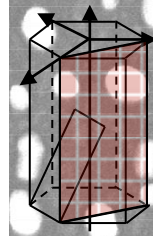
Fe:Cubic



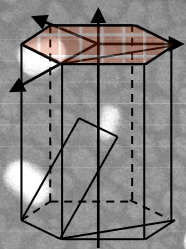
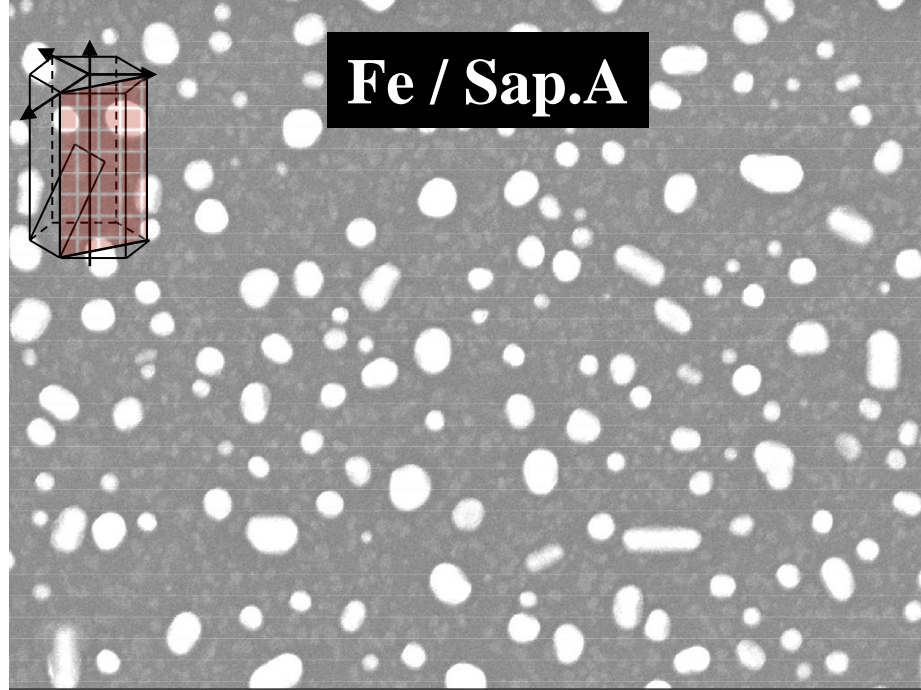
Tetragonal-like

SEM Images of
Fe(5nm)/Sapphires Heat Treated
at 600°C. Size and Shape of
Crystallites Depend on Sapphire
Faces.

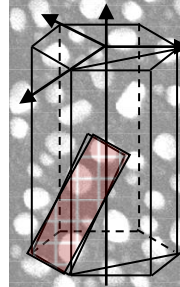
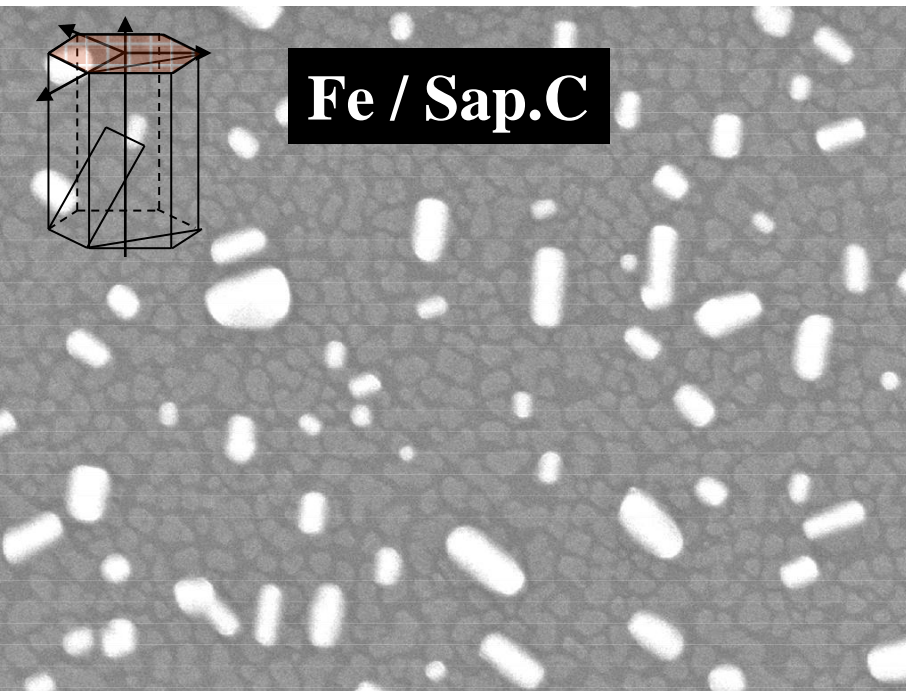
M. Yudasaka, *J. Nanoscience and Nanotechnology*,
in press.



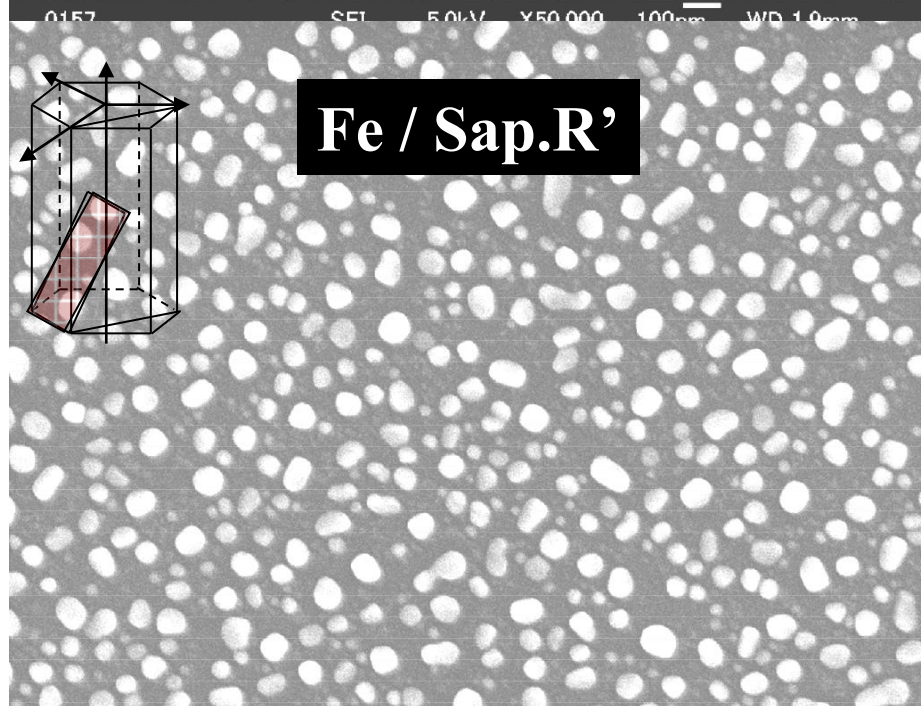
Fe / Sap.A



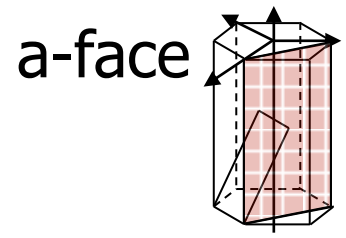
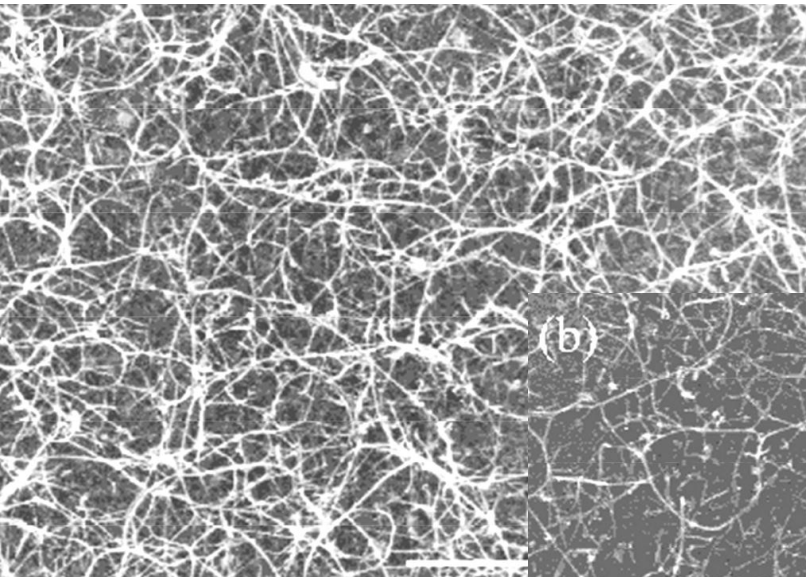
Fe / Sap.C



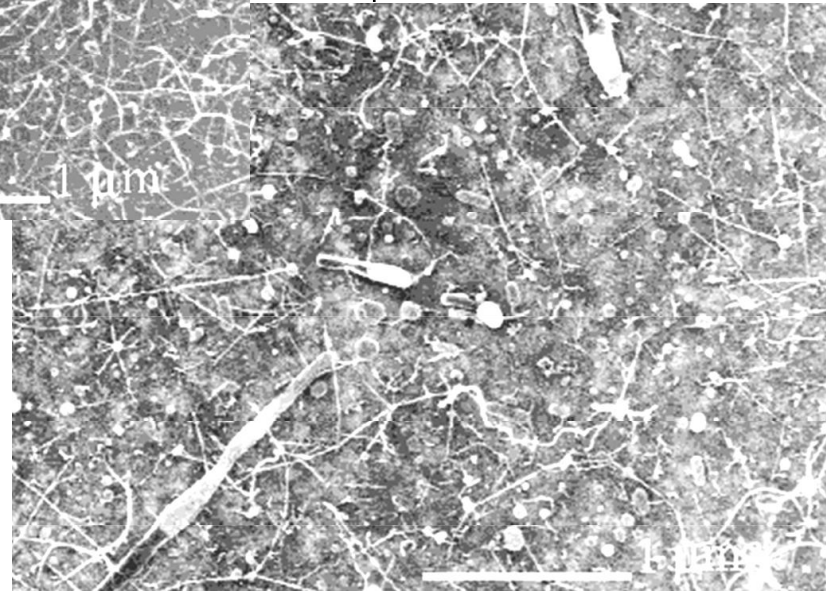
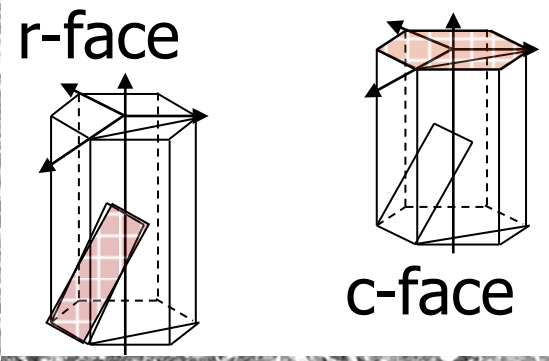
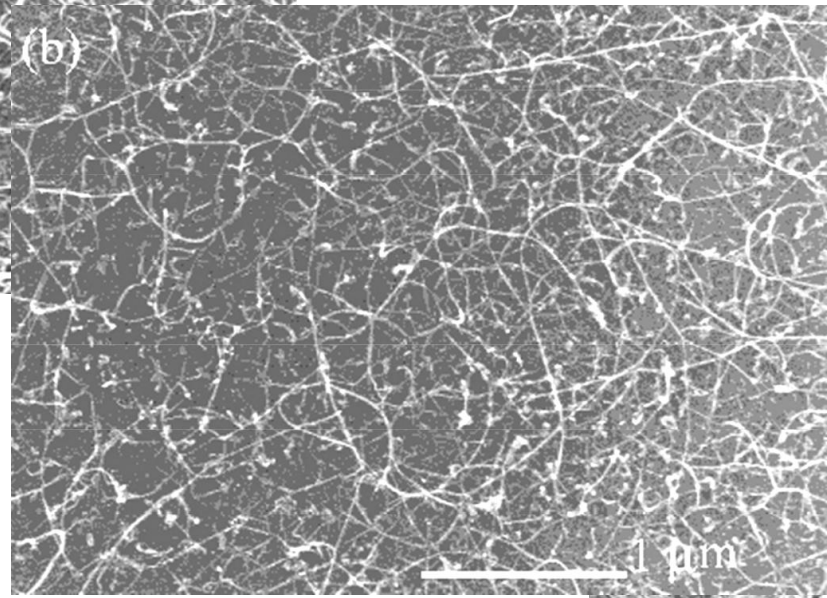
Fe / Sap.R'



SEM Images of Deposits Obtained by CVD : Fe (2 nm) /Sapphire, 800°C, CH₄ (0.6L/min)

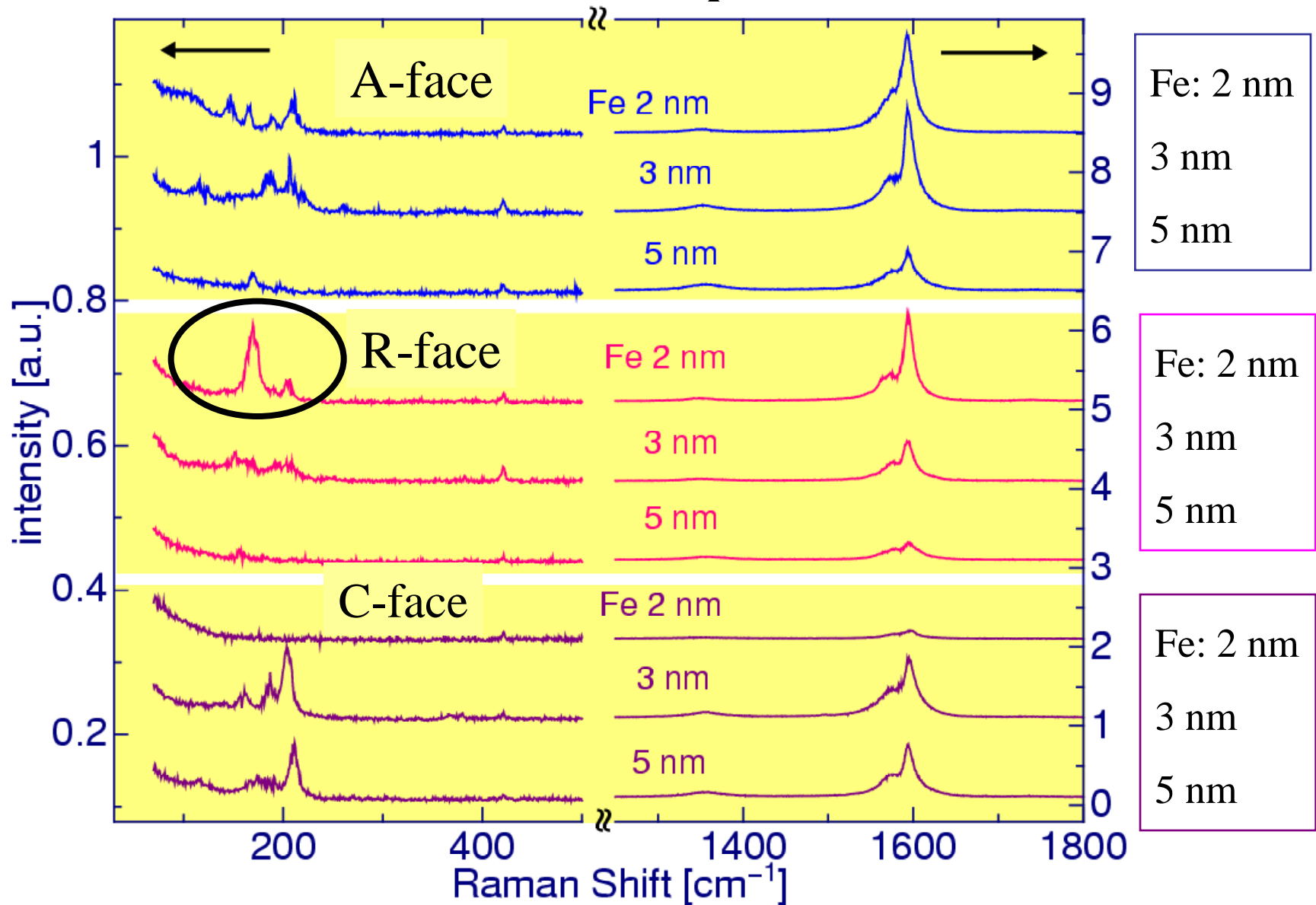


- *Many SWNTs*
- *Bundles*
- *Less MWNTs*



Hongo, Yudasaka, Ichihashi, Nihey, Iijima
Chem. Phys. Lett. 361(2002)349.

Raman spectra (488 nm)

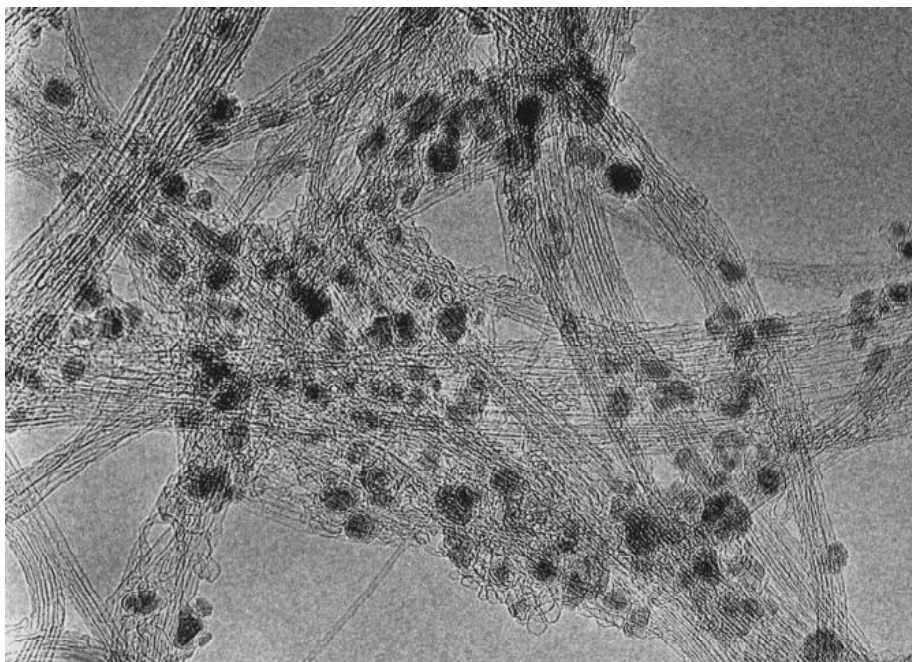
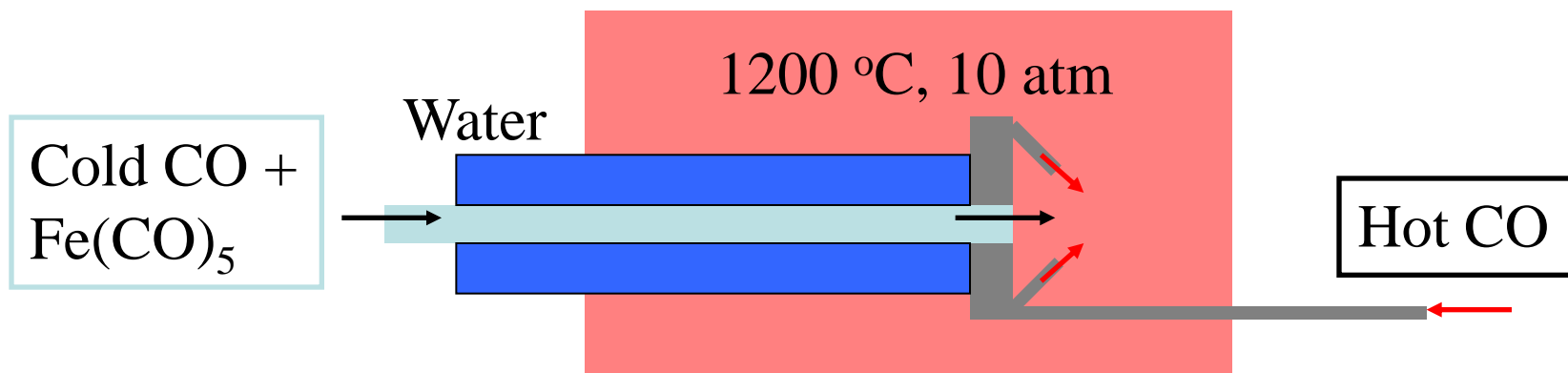


SWNT growth by CVD

Large scale
Site selective
Orientation
Long

1999: Large Scale Production of SWNTs by High Pressure Gas Phase Reaction (Smalley Group)

(P. Nikolaev et al. Chem. Phys. Lett. 313(1999)91.)



Production rate: ~1 g/h

Purity : 70%

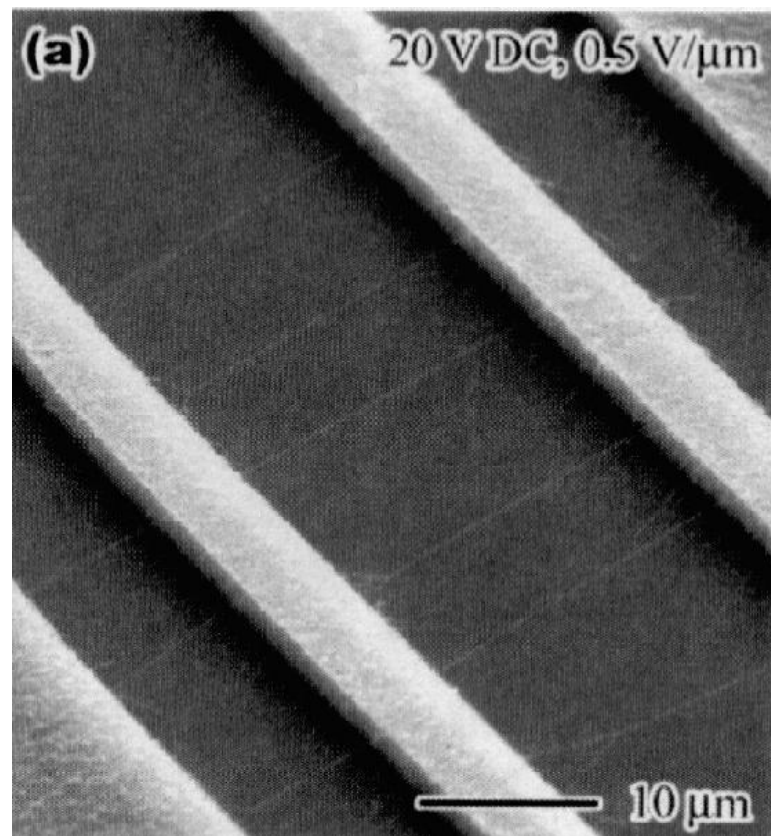
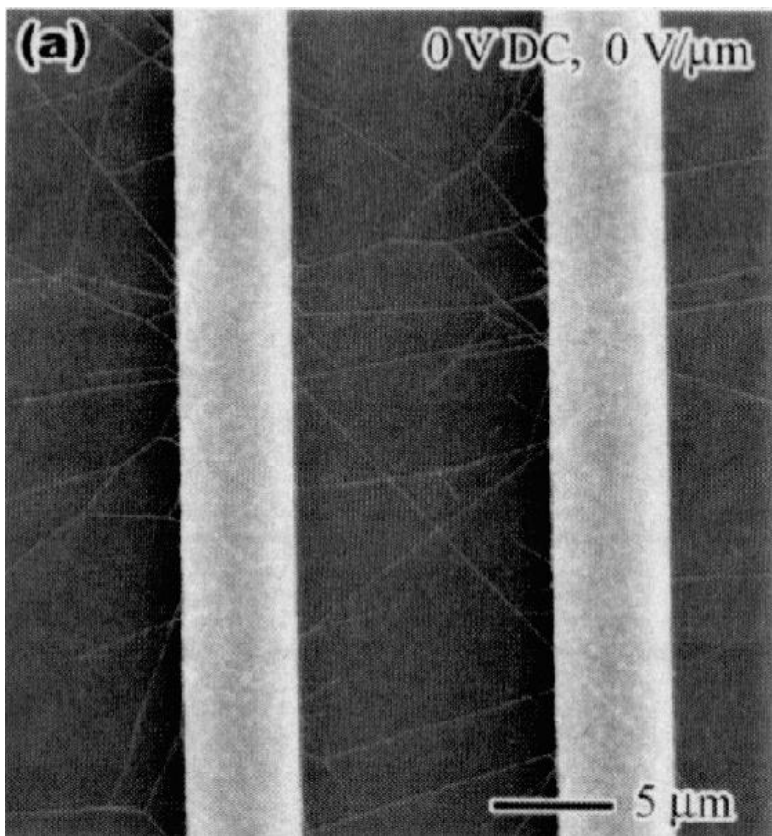
Tube diameter : 0.7~1.0 nm

Price: 500 \$/g (CNI)

Complete removal of Fe is difficult.

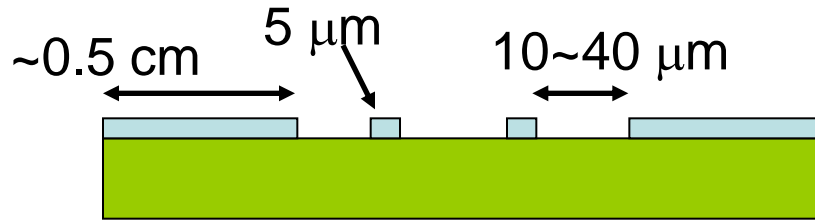
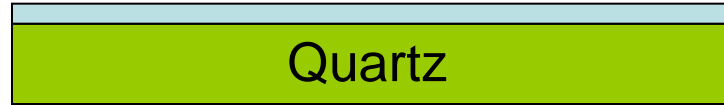
図8 電場印加CVDにより配向をそろえて成長したSWCNT

Y. Zhang et al, Appl. Phys. Lett. , 79(3083)2001



印加電圧ゼロの時にはSWCNTがいろいろな方向に成長している(左)が、0.5 V/μmかけるとSWCNTが向きをそろえて成長する(右)。

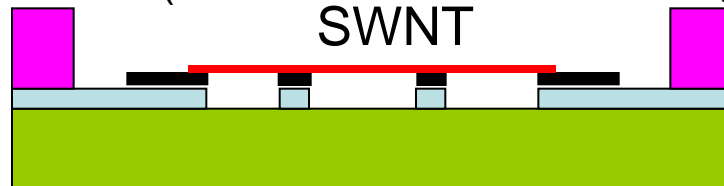
Poly-Si (3 μm)



Catalyst (Fe oxide and Mo oxide nanoparticles supported on mesoporous alumina/silica frames on the elevated poly-Si structures.)



Electrode (dc 0~200 V or ac 30 MHz, 10V)



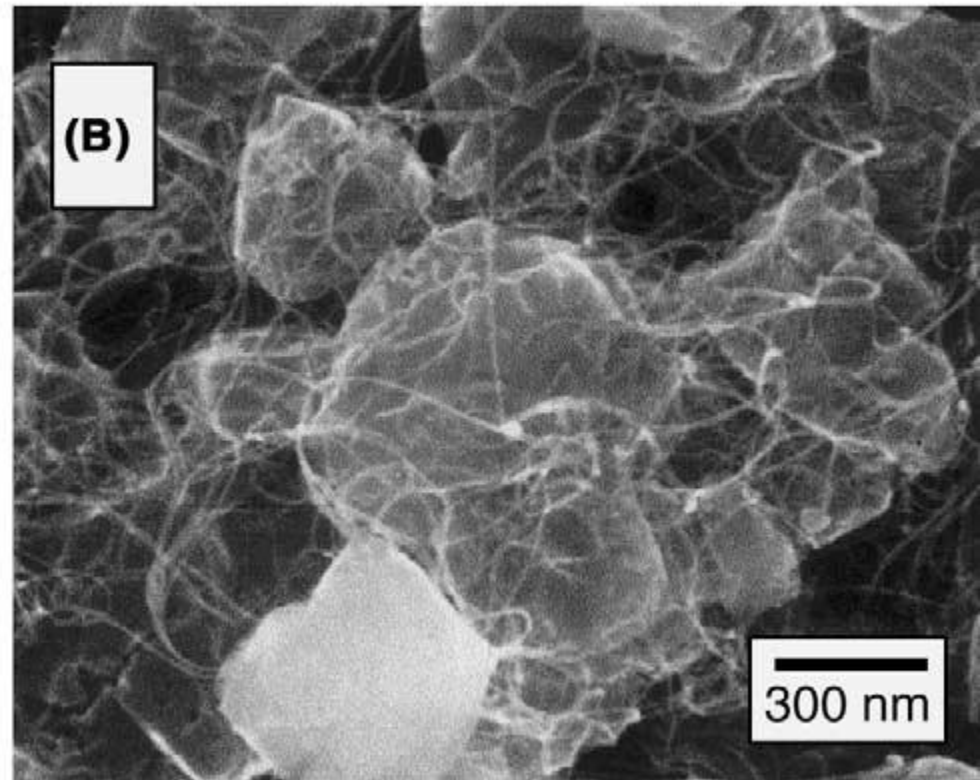
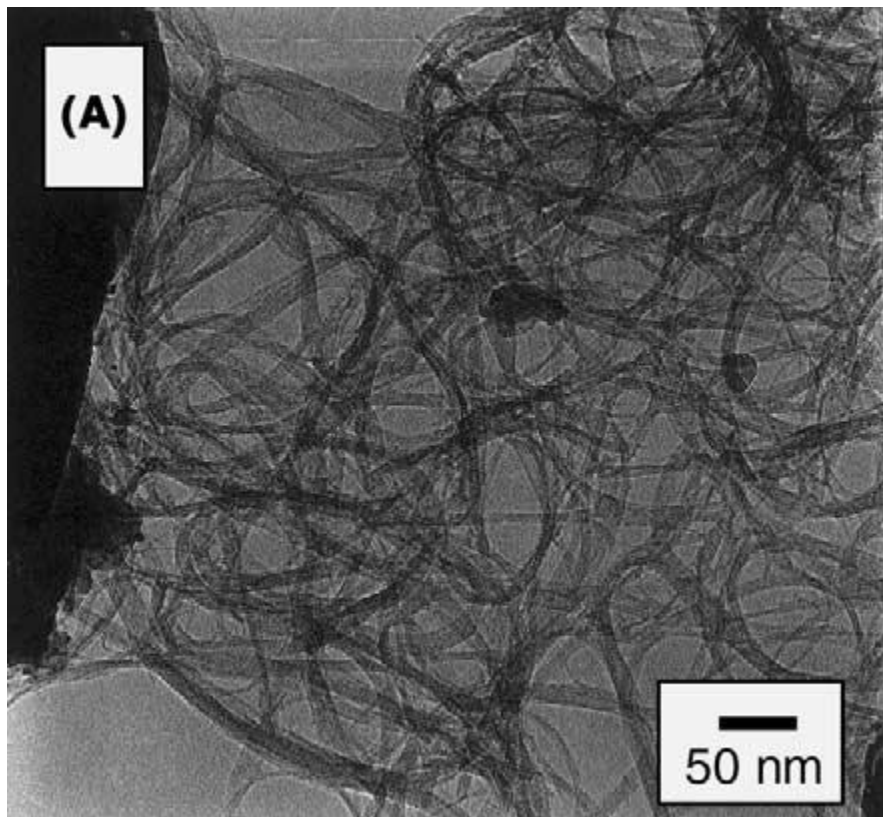
Photolithography and plasma etching.

A liquid phase catalyst precursor film (AlCl_3 , SiCl_4 , FeCl_3 , MoO_2Cl_2 , EtOH, MeOH, P103 block copolymer) was transferred onto the top of the poly-Si pattern by contact printing using a poly(dimethylsiloxane) elastomer stamp. Calcination at 300°C for 12h in air to Remove organic materials.

CVD: 900°C , CH_4 (500 sccm), H_2 (200sccm), 4 min, 1 in. tube furnace.

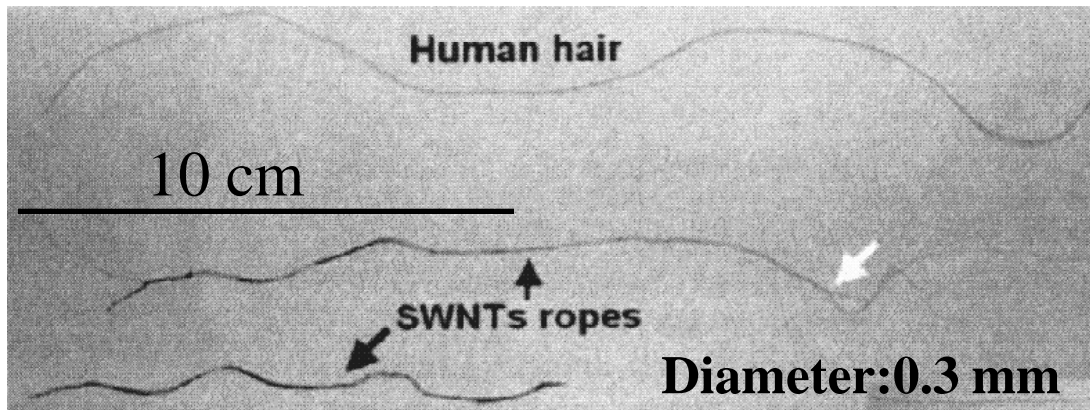
Alcohol CVD

Maruyama et al 2002



Long SWNT Strands.

H. W. Zhu et al. Science **296**(2002)884.



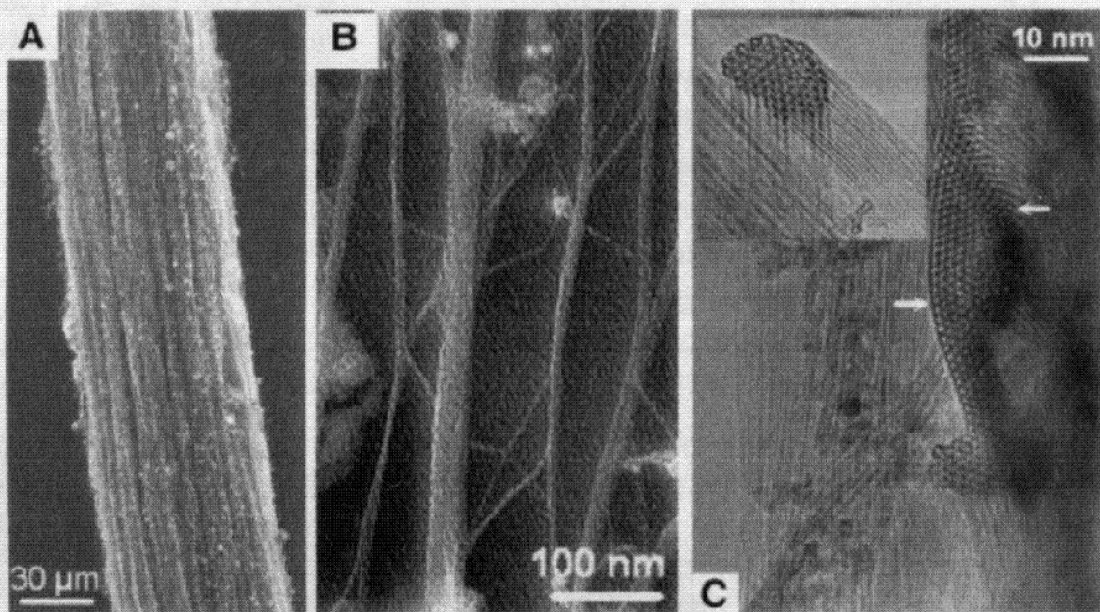
Catalytic CVD (Floating catalyst method) in a vertical furnace.

n-Hexane solution of ferrocene (0.018 g/ml) + Thiophene (Sulfur additive, 0.4 weight %): 0.5 ml/min.

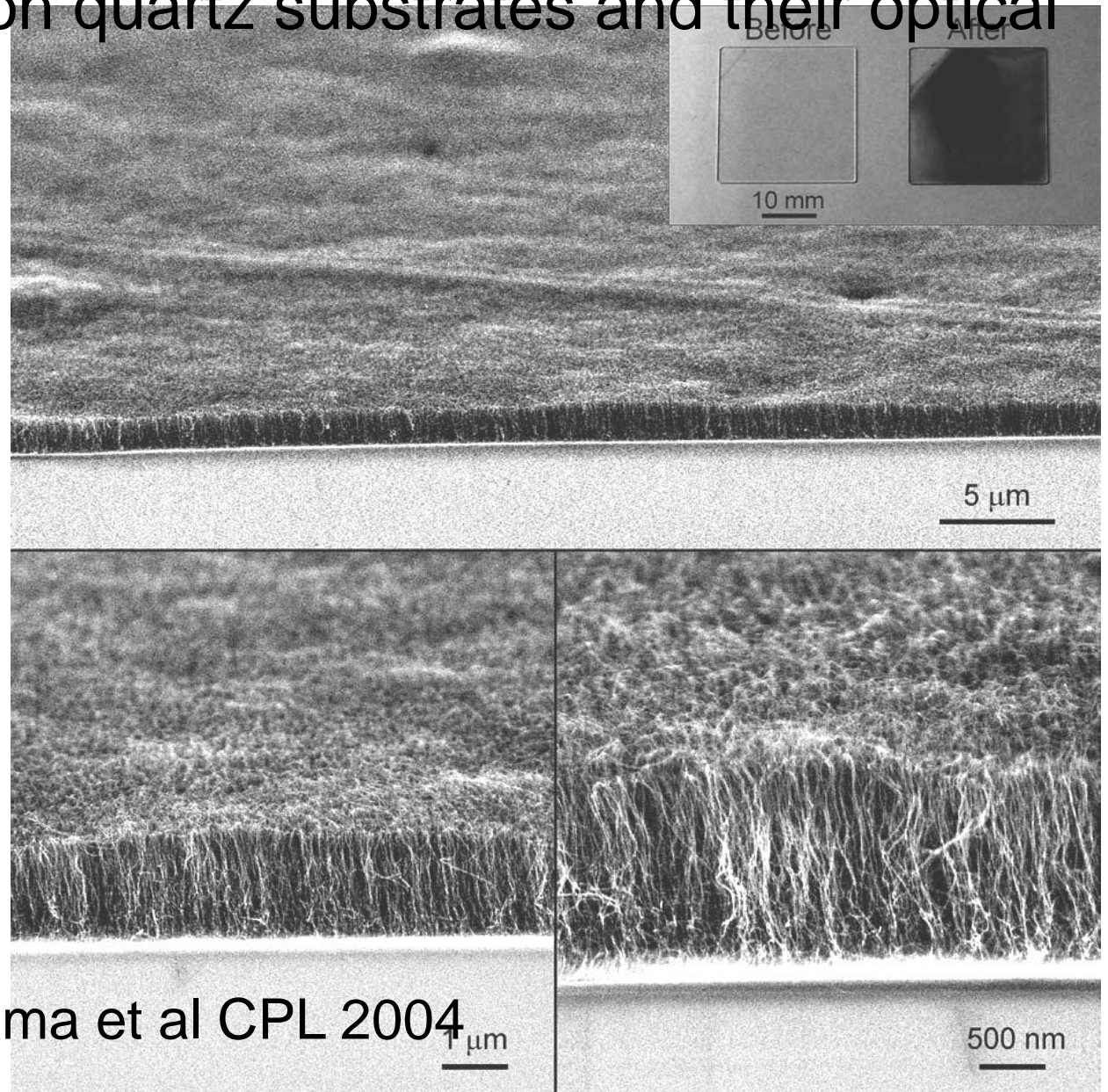
Carrier gas(H_2): 250 ml/min.
1200°C

SWNTs: ~0.5 g/h
Impurity(Fe, a-C): ~5 wt%)

Young's modulus: 77 GPa

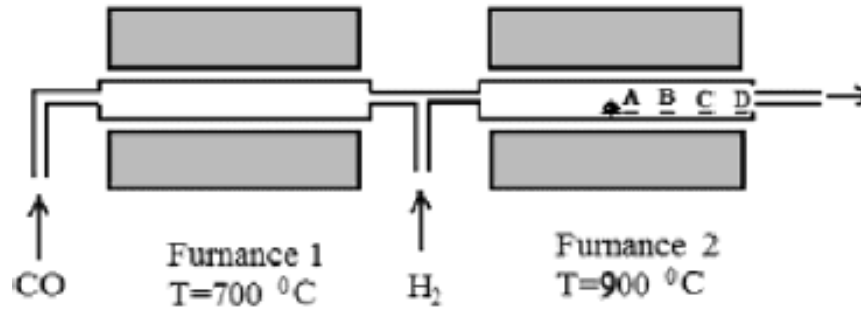


Growth of vertically aligned single-walled carbon nanotube films on quartz substrates and their optical anisotropy

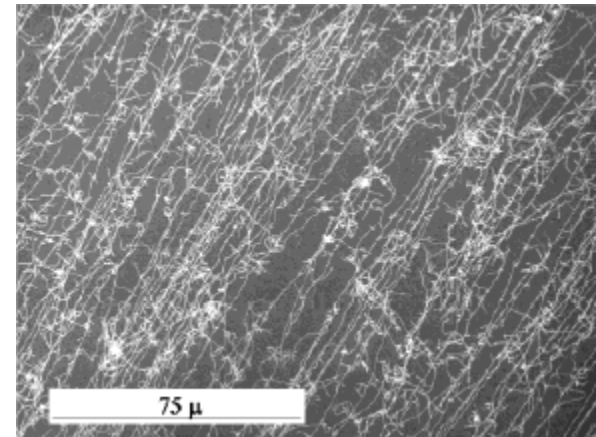
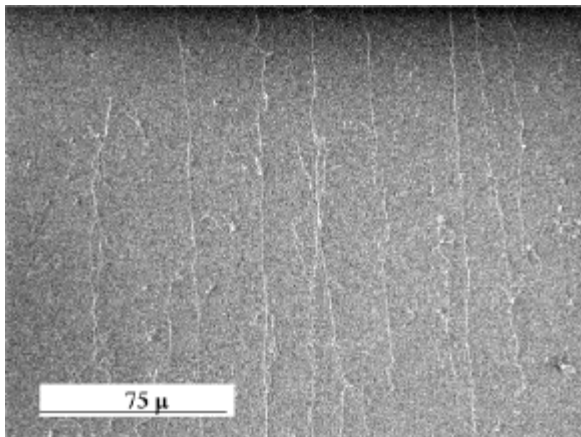
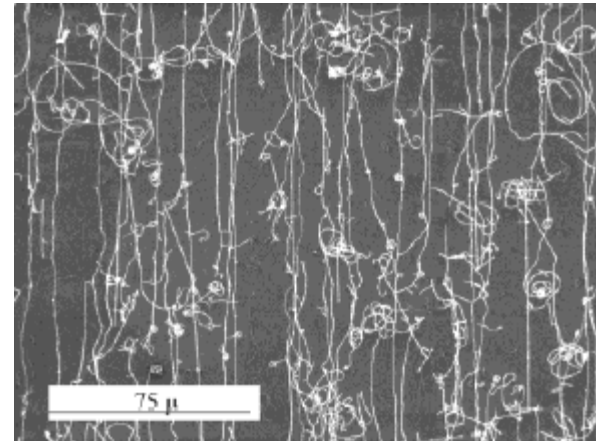
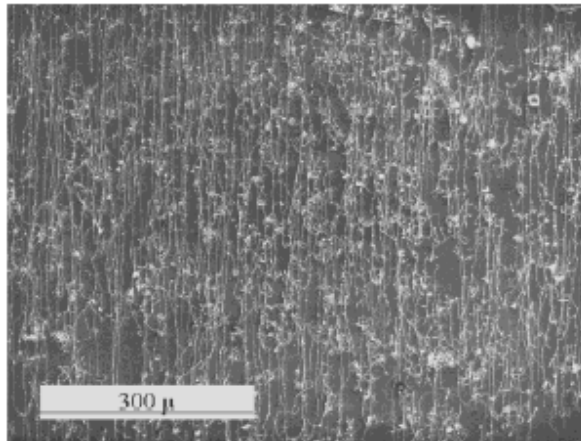


Murakami, Maruyama et al CPL 2004

Long Ropes of SWNTs for Elevators to Moon.

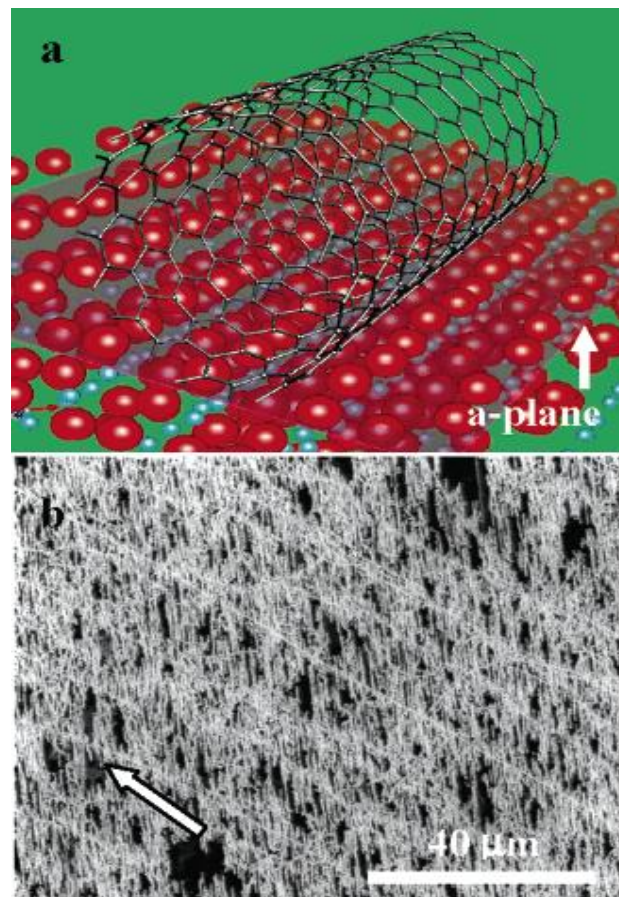
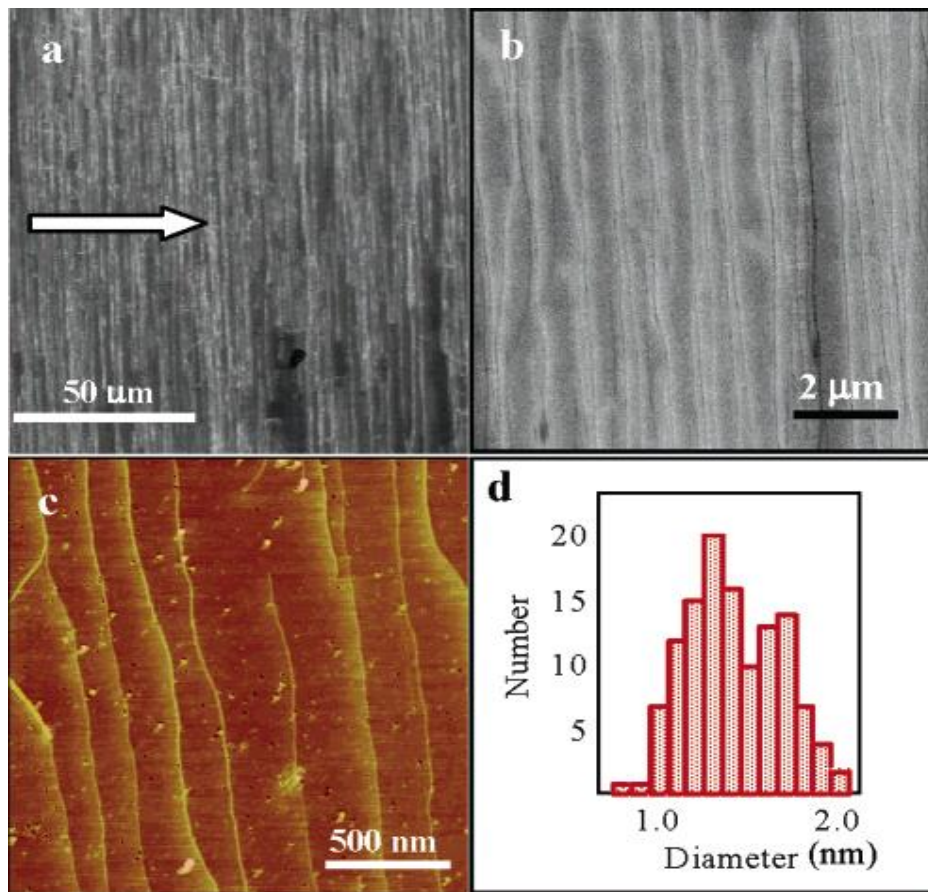


S. Huang et al.
J. Chem. Phys. B
jp0364708



Template-Free Directional Growth of Single-Walled Carbon Nanotubes on a- and r-Plane Sapphire

Song Han, Xiaolei Liu, and Chongwu Zhou



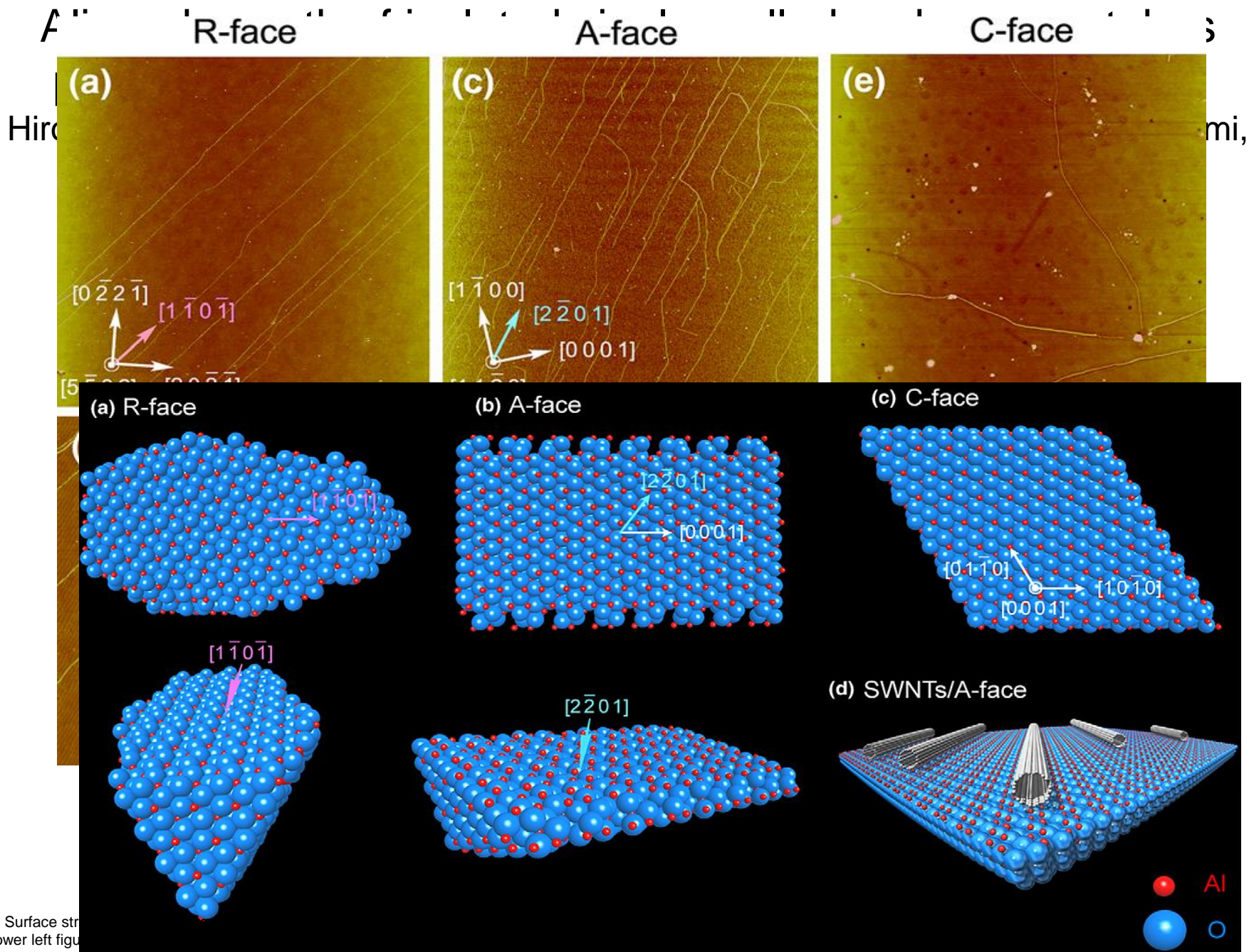


Fig. 5. Surface structure of AlN. (a) R-face, (b) A-face, and (c) C-face. The pink and light blue arrows for the R and A faces, respectively. A schematic diagram of SWNTs grown on the A-face substrate is shown in (d). (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Water-Assisted Highly Efficient Synthesis of Impurity-Free Single-Walled Carbon Nanotubes

Kenji Hata,* Don N. Futaba,* Kohei Mizuno, Tatsunori Namai, Motoo Yumura, Sumio Iijima

