

# C, B, N and P: "Low Doping vs. Heteronanotubes"

Paola Ayala

Workshop Carbon Nanotube Growth, Separation and Spectroscopy The University of Tokyo, 05.10.2012



# <u>OUTLINE</u>

#### State of the art

Morphological characteristics

Theoretical approaches

Successful experimental attempts, Experimental data

#### **Our Samples: Experimental Evidence of SW doped CNTs**

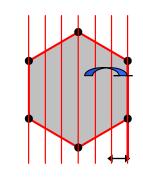
Sample morphology: SWNT/MWNT

Diameter/diameter distribution and defect concentration

Heteroatom-content, doping environment

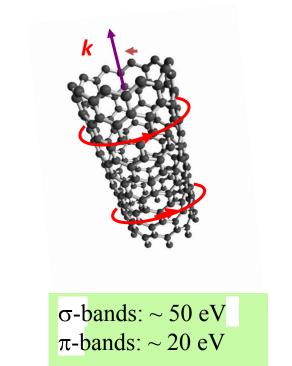
#### **Future Perspectives regarding doped CNTs**

### **Electronic Properties and Structure of SWNTs**

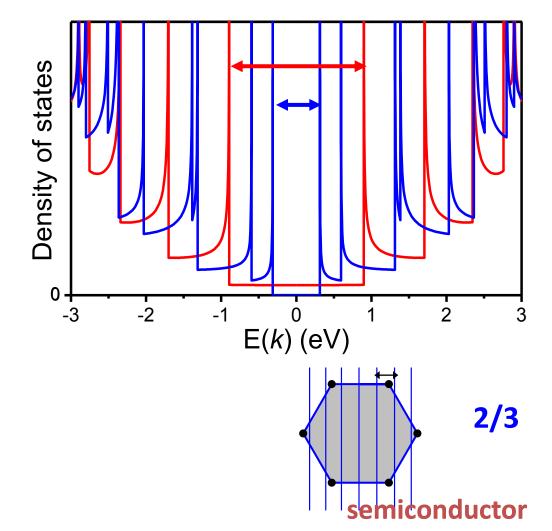


1/3

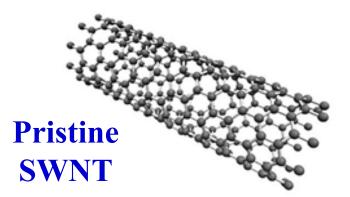
metal

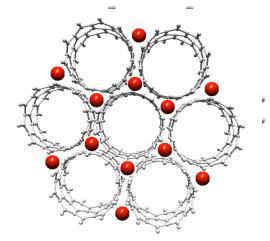


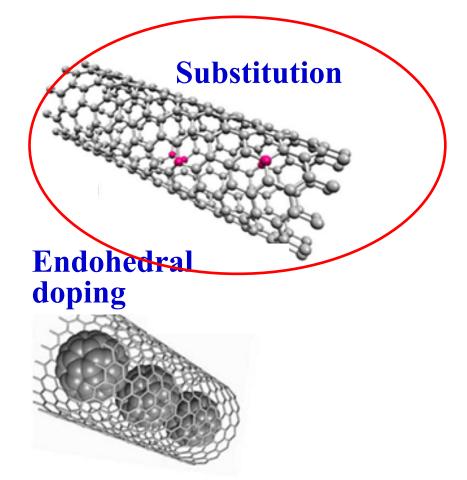
Electronic properties fully determined by geometry!!!



#### Modifications to induce changes in the electronic structure...







Intercalation

P.Ayala, et al., Reviews of Modern Physics 82(2010)1843-1885. P.Ayala, et al., Carbon, 41 (2010) 1853.

# Doping with N, B, P?

# **Different features in MW and SW doped nanotubes.**

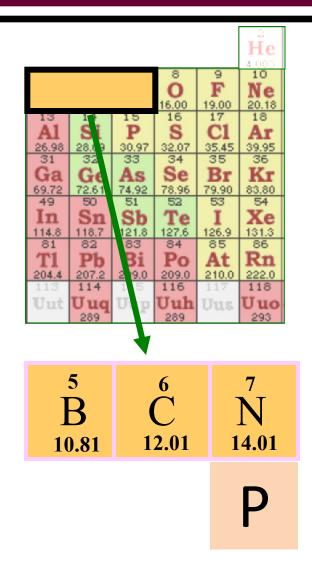
#### •SWNTs

Sharp localized states above and below the Fermi level (n- or ptype doping), by substitution or removal of C atoms.

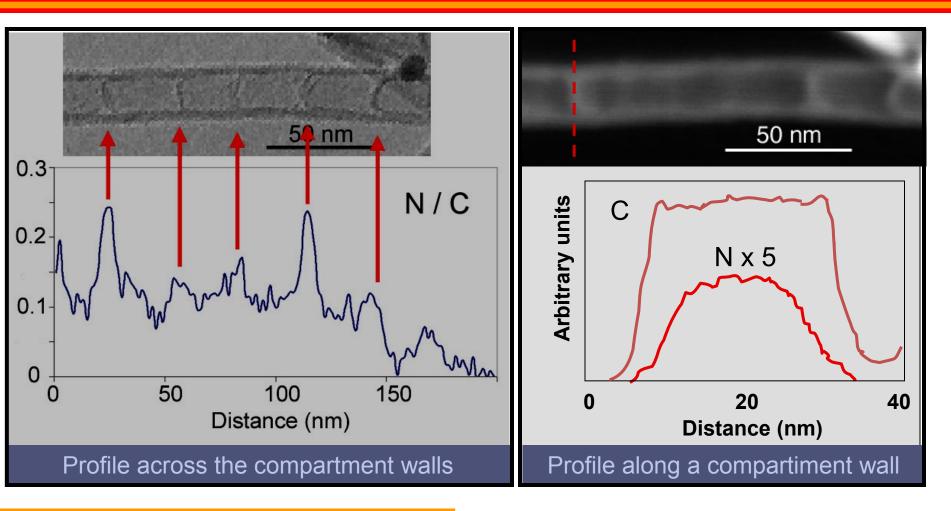
#### •MWNTs

Gain of electrochemical reactivity,

Functionalization for further applications.



## How much N can we incorporate Substitutionally?

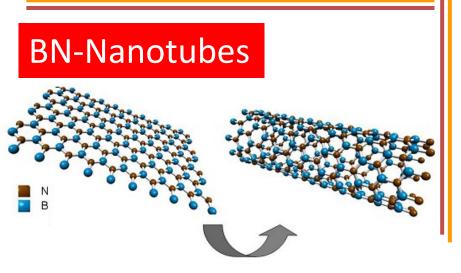


#### Observations for MW-CNx-CNTs

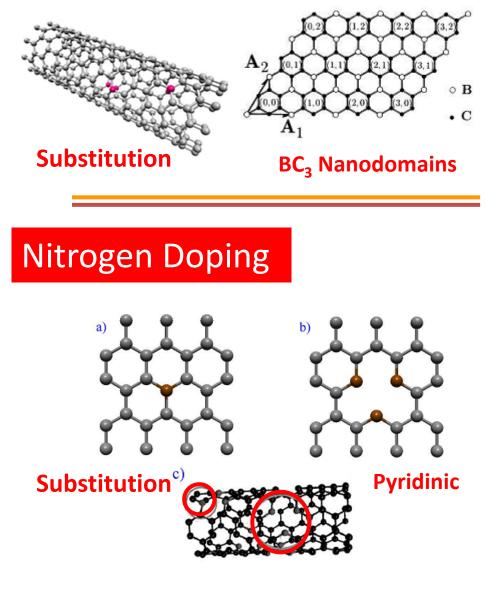
N is dominantly localized in the compartiments

Doping can reach 25 %

Courtesy: Prof.Christian Colliex. (LPS-Orsay)- CIASEM 2005. Havana We are dealing here with graphitelike planar structures that resemble the geometry of a honeycomb.

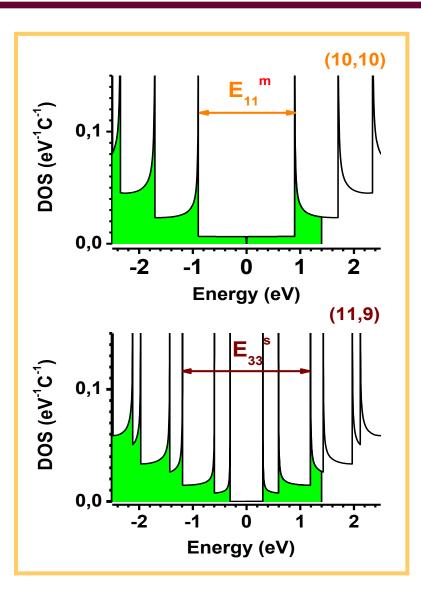


## **Boron Doping**



#### NANOTUBES:

## Changing Electronic Properties by substitutional doping



#### Low doping levels

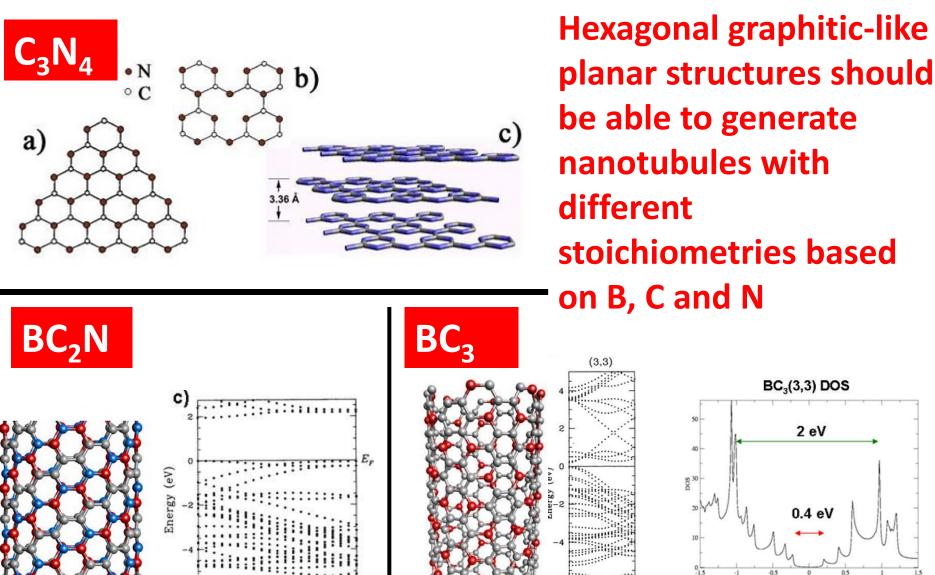
#### •Semiconductor Industry

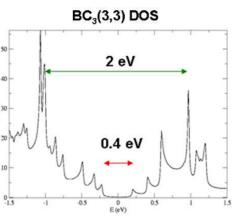
•Rigid band model

High doping levels

•Chemistry: New stable structures

•Rigid band model not applicable !!!





The incorporation of foreign atoms such as B and N can break the stability of a planar structures depending on the site they enter.

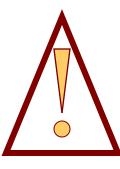
## **CNx and CBx Nanotubes Synthesis**

#### Successful Attempts

- Laser Ablation
- Arc Discharge
- CVD / AA-CVD
- Spray Pyrolysis
- Ball Milling (CBx)
- Substitution Reactions (BC<sub>3</sub>)

#### Parameters to control:

- Vapor pressure C/N Feedstock
- Feedstock Self-Pyrolysis Temperature
- Catalyst Composition and Activity
- Catalyst pre-treatment

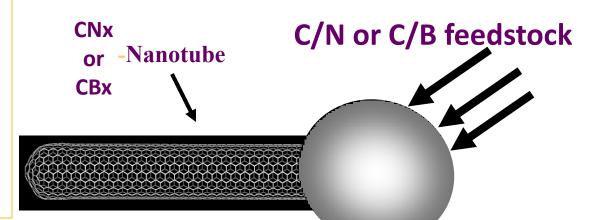


A pure C/N feedstock is needed to determine the N incorporation profile within the tubes. **Benzylamine (2%at.)** 

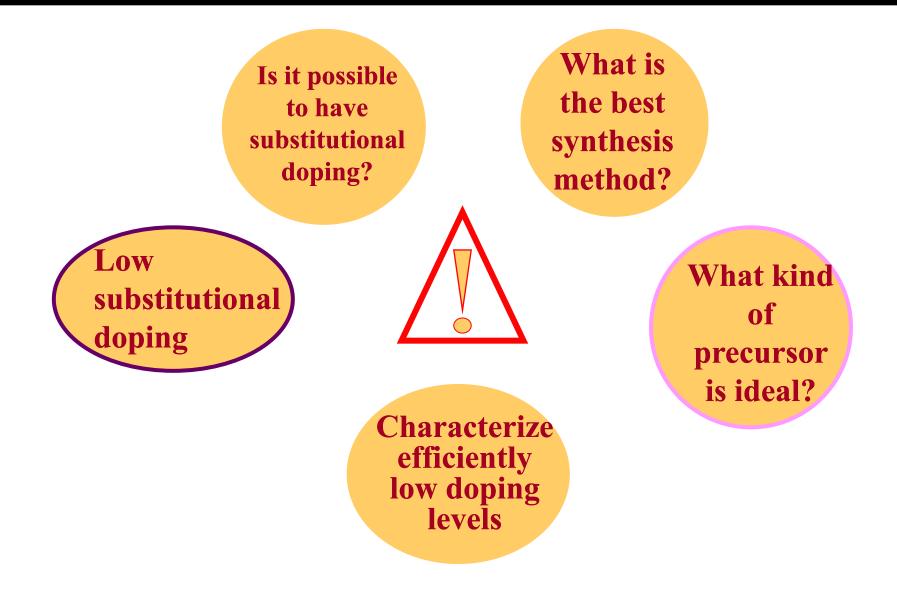
Acetonitrile (6%at.)

**Triisopropyl Borate (4% at.)** 

catalyst particle



## Substitutionally doped SWNTs: Challenges



# **Sample Characterization**

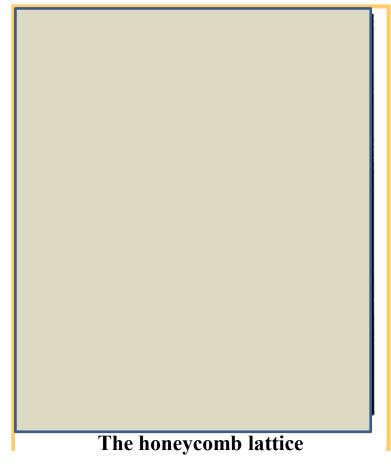
- Morphology: SEM(Bulk)/TEM (Local)
- Diameter and defect
   concentration:
   Raman Spectroscopy
   (Bulk sensitive)
- Doping level TEM-EELS (?) XPS







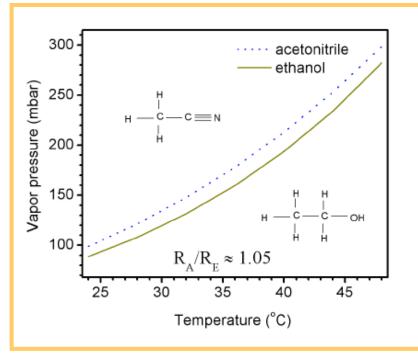
# **The Reaction Atmosphere**



P. Ayala, et al. J. Chem Physics 127 (2007) 184709
P. Ayala, et al.. Chem Mater 19(2007) 6131
P. Ayala, et al. J. Phys Chem C 101(2007)2879

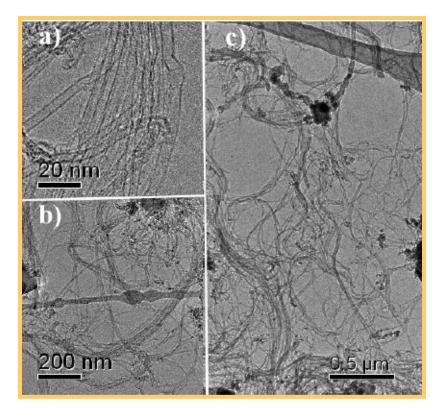
•Is N incorporation really energetically disfavored?

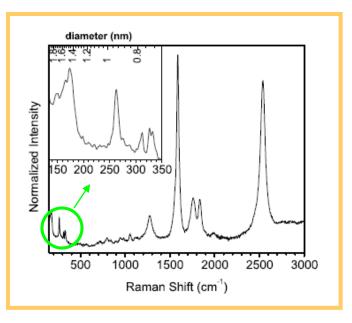
•Is it possible to synthesize N and B-doped SWNTs?



## Effects of the Reaction Atmosphere Composition

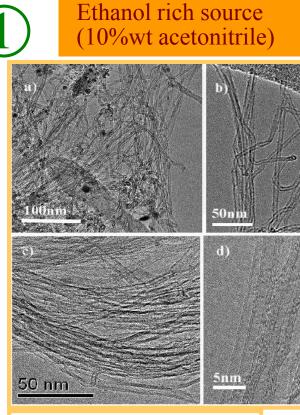
#### Pure Ethanol Nanotubes...



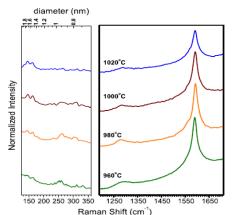


Growth: Between 800 and 1050°C
High yield of SWNTs Raman: 0.8 to 2nm TEM: Bundles

## **Composed Feedstocks**

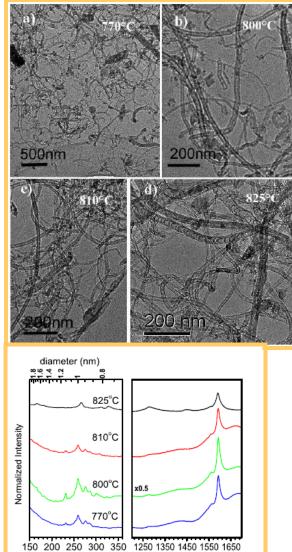


•Onset for NT growth: 880°C •TEM: SWNTs and low amount of MWNTs



Constant D/G with temperature
Onset for MWNTs growth 1020°C

## Equally contributing composition (1:1 acetonitrile vs ethanol)



Raman Shift (cm<sup>-1</sup>)

•Multiple morphology product independent of synthesis T

•Lower T, ~5% of short (500nm) and thin nanotubes (~1-15nm)

•Highest SWNTs Raman response at 800°C

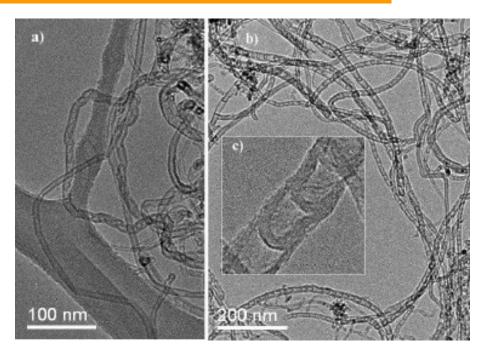
•Low formation of SWNTs at higher temperatures

•SWNT grow Twindow narrows down.

# 3

# **Pure C/N feedstock**

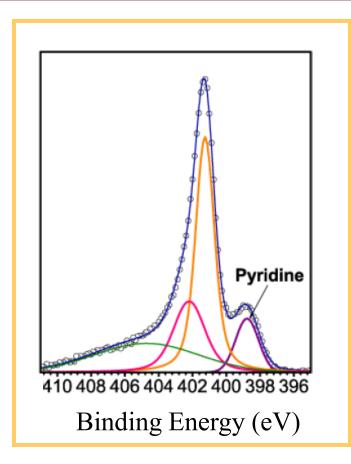
#### Pure acetonitrile feedstock....



Practically no evidence of SWNTs
MWNTs grow between 750 and 900°C

#### •Morphology:

Defective MWNTs and bamboo-like tubes with diameters between 15 to 20nm.



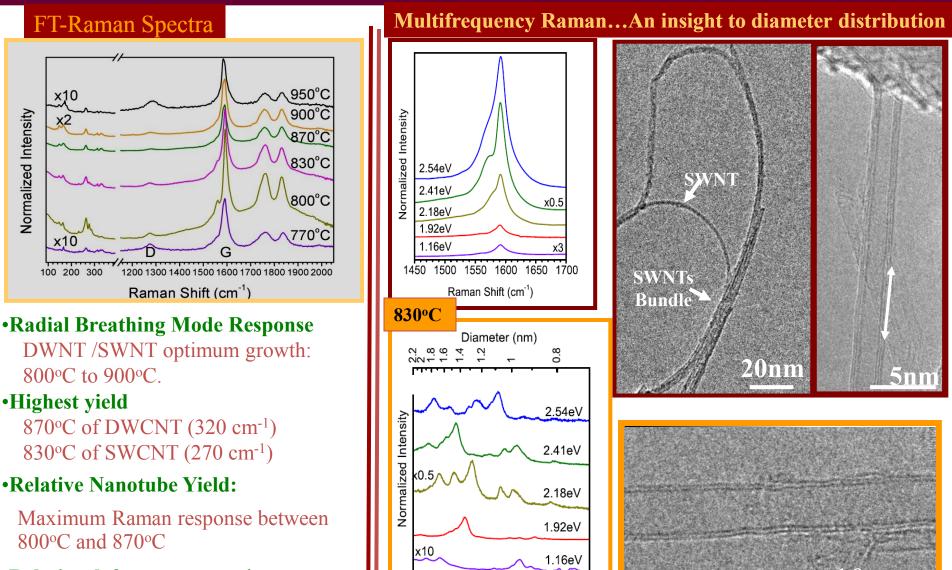
#### Max N content:

Composed Feedstocks ~1.5 0.2%at. Pure Acetonitrile Feedstock ~1.7%at C1s: 284.7eV at all T

## **Benzylamine CNx-nanotubes**

300

350



150

200

250

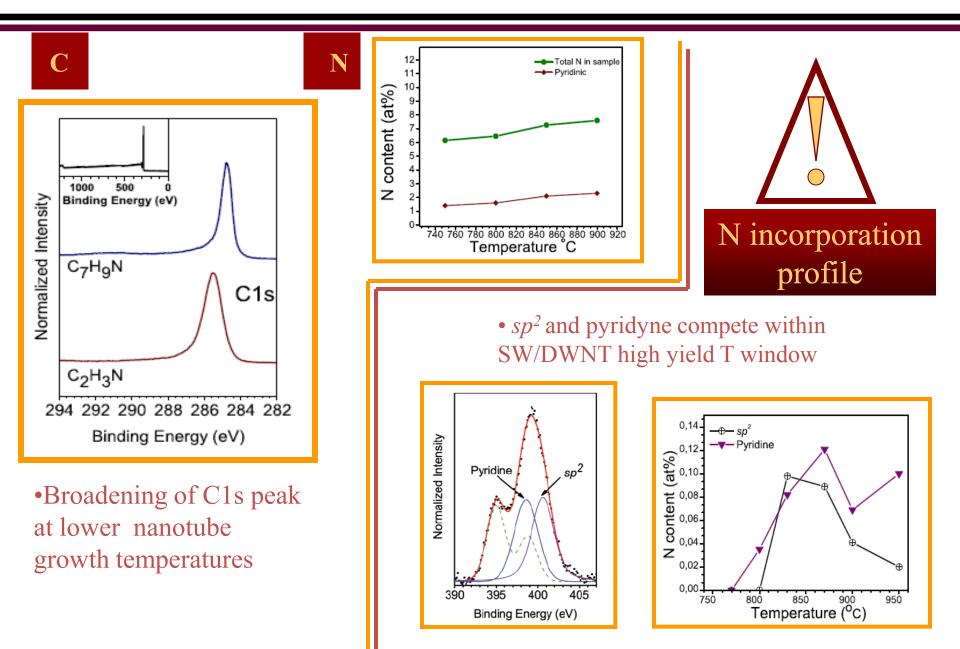
Raman Shift (cm<sup>-1</sup>)

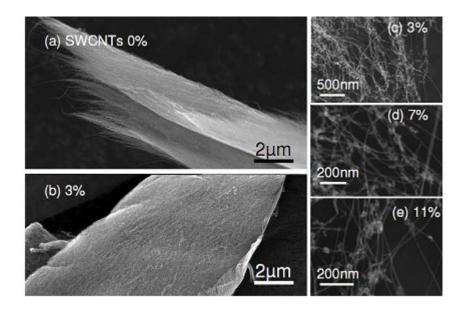
•Relative defect concentration:

Low D/G ratio down to 1/20

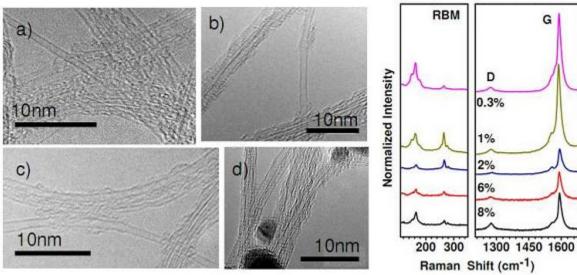
## **CNx NTs**

## **Atomic composition: XPS**





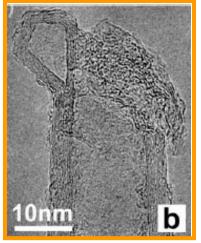
## 0,3% maximum Substitutional!!!



A.L.Elias, P.Ayala, M.Terrones et al. Journal of Nanoscience and Nanotechnology Vol.10, 1–6, 2010

# **Boron Doped Nanotubes**

#### MWCBx-nanotubes (Illed Caps)



•MW-CBx nanotubes

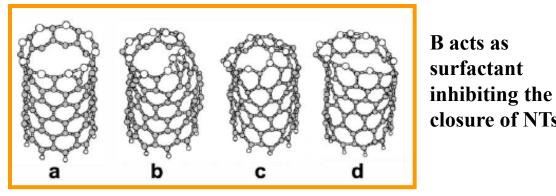
**B** acts as

surfactant

closure of NTs

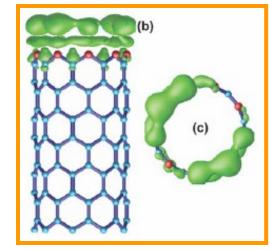
- •Straight
- •Illed caps

#### SWCBx-nanotubes



Blase et al. PRB 1998

#### Atomic Welders



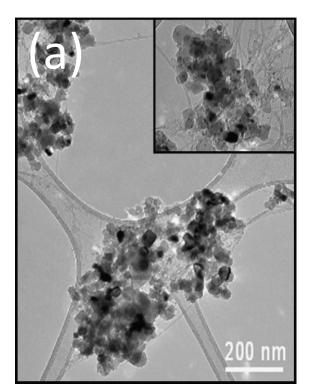
Terrones, M., et al, Materialstoday 7 (2006) 10, 30

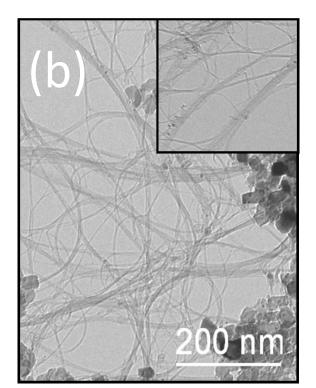
 SW synthesized throug substitution reactions

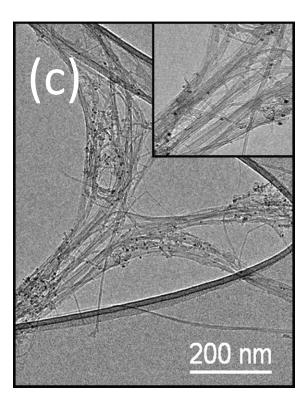
- Doping up to 25%
- Arc Discharge-From mixture in target

# Is it possible to use a liquid feedstock for CBx-SWNTs ?

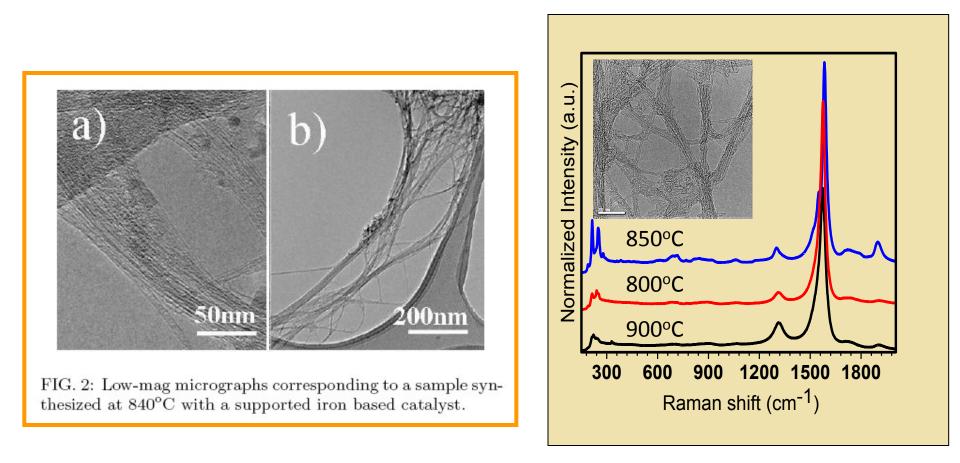
- Problem 1: Experimental Setup
- Problem 2: Precursor and Catayst
- Problem 3: Co-products





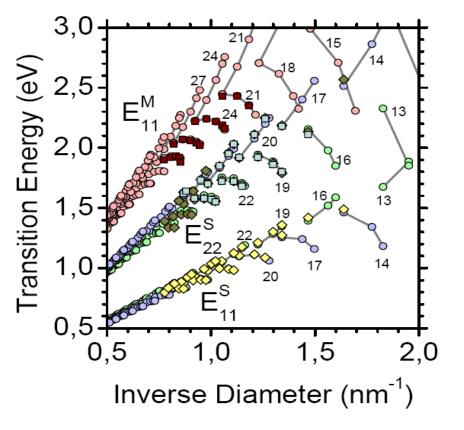


# Nanotubes from Triisopropil borate

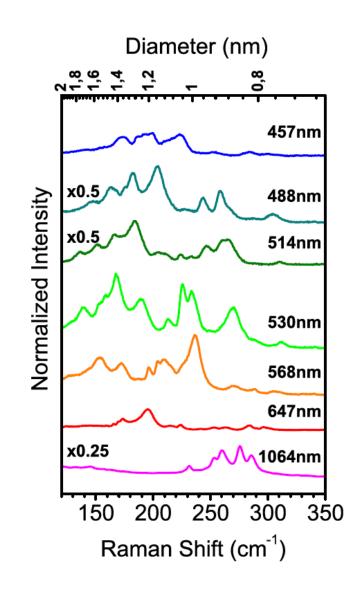


S. Daothong, J. Parjanne, M. Valkeapää, T. Pichler, P. Ayala, et al, Phys Stat Sol b 246 (2009) 2518

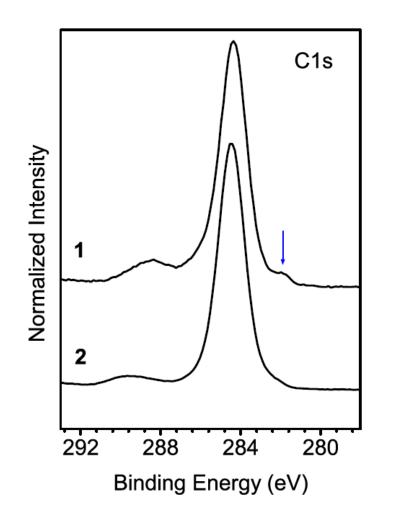
## **Multifrequency Raman**



Multifrequency Raman response in the RBM region with the different laser excitation energies of a samples synthesized at 840°C



## **XPS:C and B bonding environments**



•C1s signal for pure SWNCNTs 284-286 eV

•C1s peak centered in 284.5 eV for B doped SWNTs samples

•Shoulder at 281.5eV (more prominent according to the B neighboring atoms)

Shirasaki et. al.

BC4  $\rightarrow$  3eV separation

Low doping occurs at lower

temperatures!!!

XPS spectra of the carbon C1s binding energy for samples synthesized at 890C and 840C.

P.Ayala, et al. J Mat Chem18(2008)5676

## **XPS:C and B bonding environments**

#### 4 major peaks

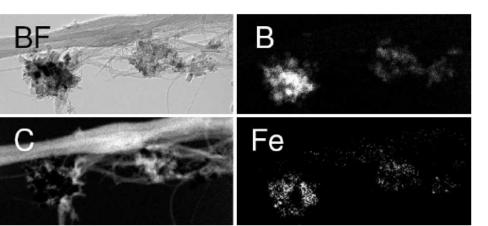
• ~191.5-192.1 Substitutional doping (Pachankarla et. al

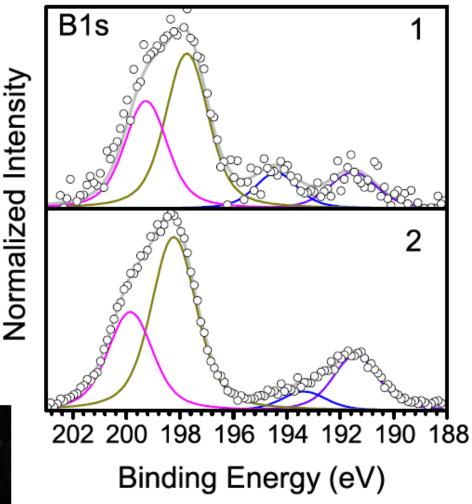
ACSNano, Nov 2007)

- ~193 Oxide formation
- Major signals

198eV and 199,5ev

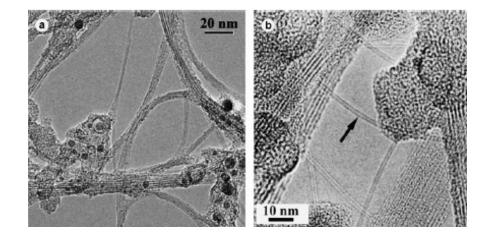
B/Fe containing compounds???





P.Ayala, et al. J Mat Chem18(2008)5676

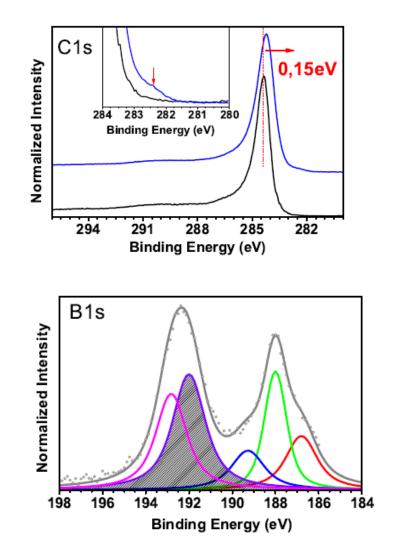
#### B-doped SWCNTs with narrow diameter distribution



Laser ablation B-doped nanotubes. Micrographs correspond to material produced from targets with nominal boron concentrations of (a) 0at.%, (b) 2.5at.%.

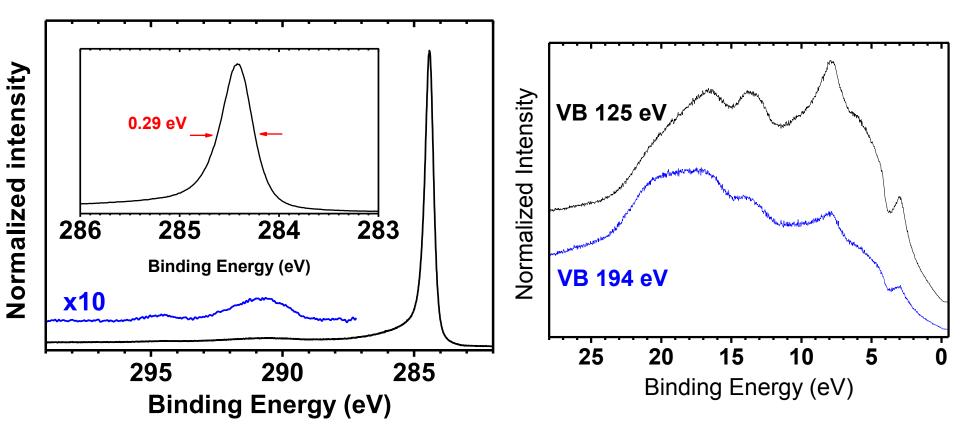
From McGuire et al. Carbon 43(2005) 219.

A. Rao (Clemson University)



P. Ayala et al. Applied Physics Letters 96 (2010) 18311

## Ultra low doping in B-doped SWCNTs



## Ultra low doping in B-doped SWCNTs

#### **C1s:**

**Overall shape** 

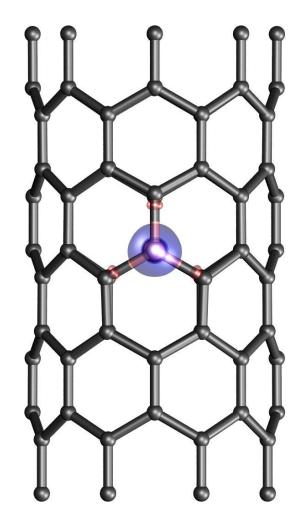
π\* resonance at 285.4 eV

 $\sigma^*$  threshold at 291.7 eV

#### **B1s :**

•First identification of heteroatoms in doped purified SWCNTs.

•Boron atomic concentration 0.0005% at.

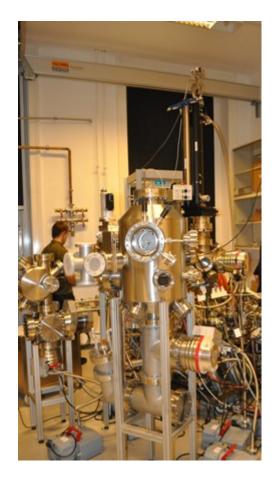


D.Mowbray, P.Ayala, T. Pichler, A. Rubio, Materials Express, 1(3) (2011) 225



A one-of-its-kind multifunctional spectrometer which includes high resolution photoemission (with XPS and ARPES) has been developed in the group. It is coupled to a multifrequency Raman and a broad range optical (1meV-6eV) setup, including preparation chambers with low temperature (He cryostats) and doping cells for insitu doping.





## Summarizing B and N doped CNTs

- Fesibility to use a pure feedstocks kept at room temperature to produce high quality doped SW nanotubes.
- Pure C/N feedstock, however C to N ratio of source is not reachable.
- Cleanliness of the system is crucial.
- Base pressure influences sample quality
- In absence of N gaseous forms, sp<sup>2</sup> and pyridine compete
- Doping at very low levels (<0.5%)</p>
- Characterize efficiently low doping levels

Techniques Implementation!!!!!

## Acknowledgements



- •Prof. Apparao Rao Clemson University USA
- •Prof. Angel Rubio University of the Basque Country-Spain
- •Prof. Esko Kaupinen , Dr. Hua Jiang Aalto University –Finland
- •Dr. Raul Arenal University of Zaragoza Spain
- •Dr. Andreas Berger nanoGUNE Spain
- •Prof. Mauricio Terrones Penn State University USA
- •Prof. Martha McCartney Arizona State University USA
- •Prof. Fernando Lázaro Freire Jr. PUC-Rio Brazil
- •Prof. Hiromichi Kataura AIST Japan
- •Prof. Kazuhiro Yanagi Tokyo Metropolitan University Japan
- •Dr. Giorgio Lanzani- University of Oulu Finland
- •Dr. Suphaporn Daothong Chiang Mai University Thailand
- •Dr.Virginia Ruiz-CIDETEC- San Sebastian Spain
- •Dr. Annick Loiseau ONERA France



# Thank you!!